

Medical Applications of Particle Physics

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Introduction: an historical review





Applications in medical diagnostics



Particle accelerators for medicine



Applications in conventional radiation therapy



- Hadrontherapy, the frontier of cancer radiation therapy
 - Proton-therapy
 - Carbon ion therapy
 - Neutrons in cancer therapy



Fundamental research in particle physics and modern medicine

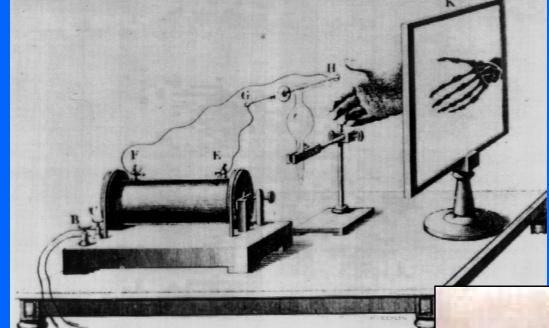
Two parallel stories with many interlinks

The starting point

November 1895: discovery of X rays



Wilhelm Conrad Röntgen



December 1895 : first radiography

...the real first medical radiography...



Radiography of the hand of Roentgent's wife Bertha

Important points:

- X-rays penetrate matter
- X-rays are differently absorbed by bone and muscle tissues
- Exposure time 15 minutes
- Exposure nowadays for a hand radiography:
 - 1/25 to 1/50 second
 - Dose: about <0.1 mSv (natural background 1-2 mSv/year)

Curiosities

- Roentgen convinced his wife to participate in an experiment
- Bertha was horrified and saw in the image a premonition of death

Basic units

- The physical dose absorbed by matter is measured in Gray
 - 1 Gy = 1 J / 1 Kg
 - Used in radiation therapy
- In radioprotection the equivalent dose is measured in Sievert
 - Equivalent dose = (Physical dose) x Wr
 - Wr takes into account the biological effects of specific radiation

Fattori di ponderazione della radiazione incidente per ottenere dalla dose assorbita (in Gy) il valore di dose equivalente (in Sv)

| Tipo di radiazione | Fattore di ponderazione della radiazione (Wr) | |
|---|--|--|
| Fotoni di tutte le energie ed elettroni | 1 | |
| Neutroni di energia inferiore a 10 KeV | 5 | |
| Neutroni tra 10 KeV e 100 KeV | 10 | |
| Neutroni tra 100 KeV e 2 MeV | 20 | |
| Neutroni tra 2 MeV e 20 MeV | 10 | |
| Neutroni di energia maggiore di 10 MeV | 5 | |
| Protoni | 5 | |
| Particella alfa, nuclei pesanti | 20 | |

Basic units

- In radioprotection the <u>effective dose</u> is defined as
 - Effective dose = (Physical dose) x Wr x Wt
 - Wt takes into account the sensitivity to radiation of different organs (stochastic effects)

Fattori di ponderazione tessutale per ottenere il passaggio dalla dose equivalente alla dose efficace (secondo ICRP 60/1990 o L.241/00)

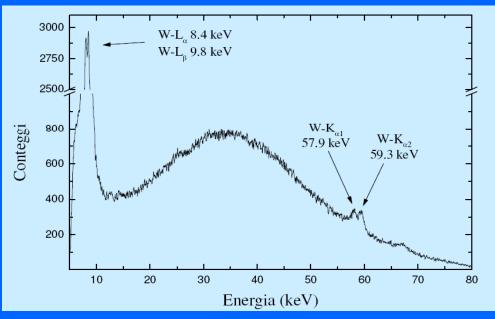
| Tessuto/Organo | Fattore di ponderazione tessutale Wt Icrp60 |
|--------------------|---|
| Gonadi | 0,20 |
| Colon | 0,12 |
| Esofago | 0,05 |
| Fegato | 0,05 |
| Mammella femminile | 0,05 |
| Midollo rosso | 0,12 |
| Pelle | 0,01 |
| Polmone | 0,12 |
| Stomaco | 0,12 |
| Superficie ossea | 0,01 |
| Tiroide | 0,05 |
| Vescica | 0,05 |
| Organi restanti | 0,05 |

....what we know today....

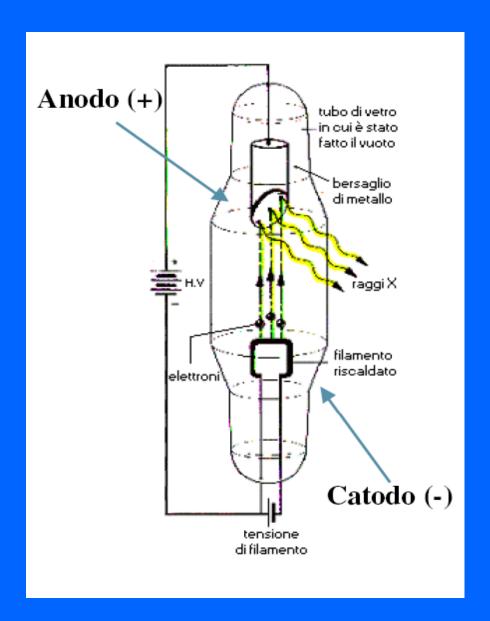
X-ray tube: accelerated electrons interact and radiate photons

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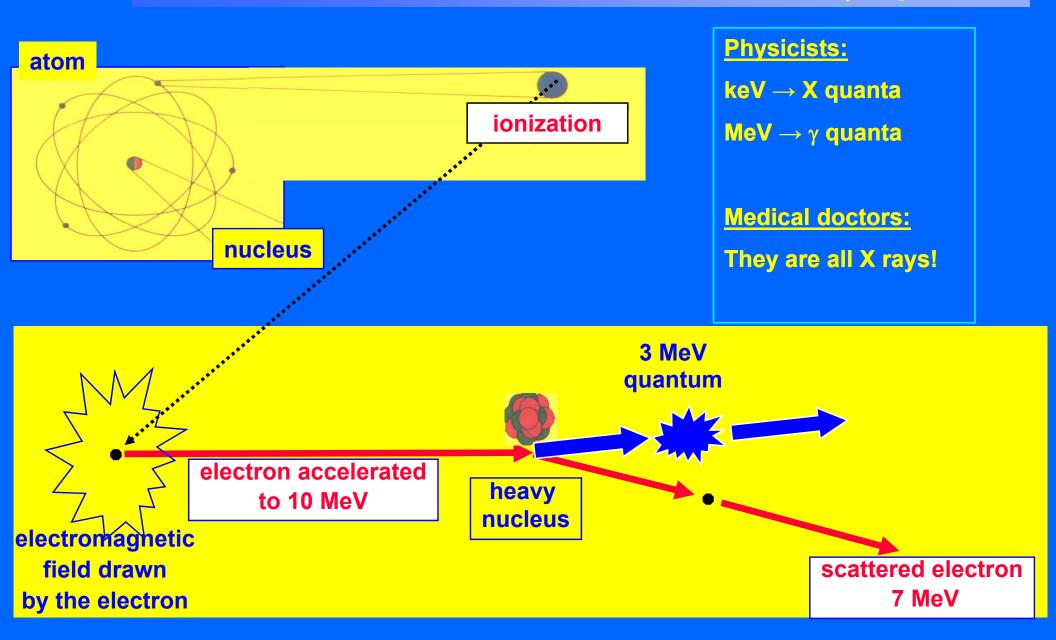
X-ray production



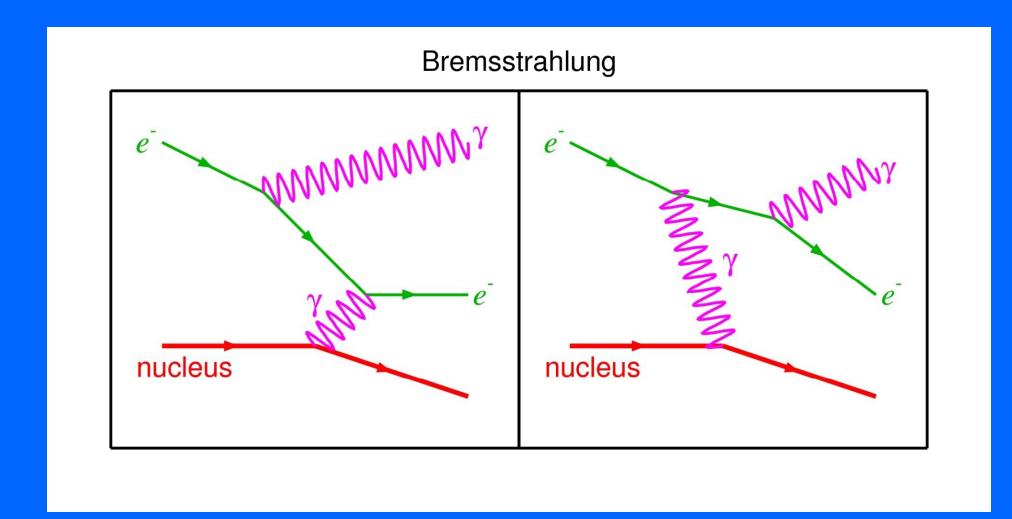
X-ray energy spectrum



Production of X and y "quanta"



The physical process



Photons interact with matter

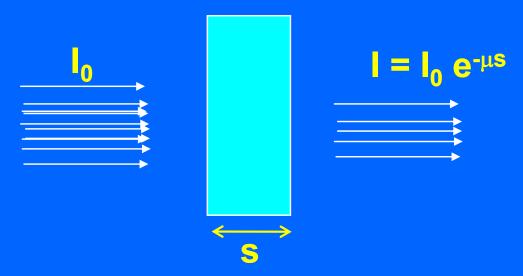
X-rays interact with matter (material Z) through different mechanisms:

- Photo-electric effect
- Compton effect
- Electron Positron pair production

(Threshold: 2 x 511 keV)

Probability $\sim \langle Z^4 \rangle$ Z^2

What happens to a photon beam?

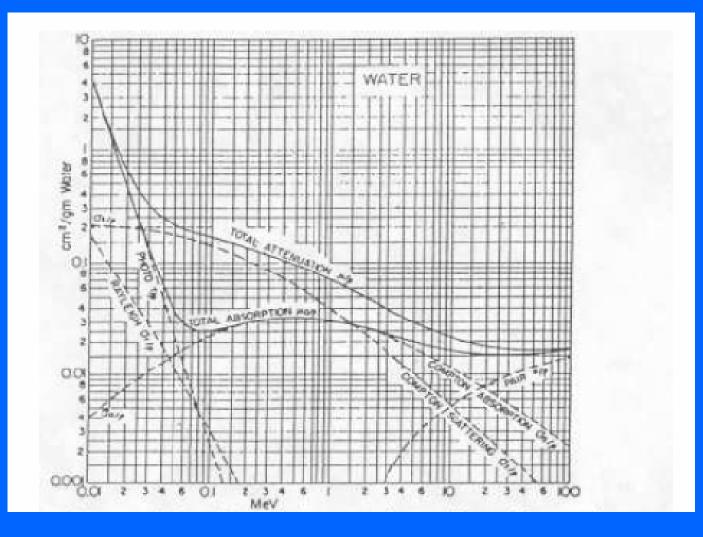


μ attenuation coefficient

Depends on:

- The material (z)
- The density
- The energy of the photons

The attenuation coefficient



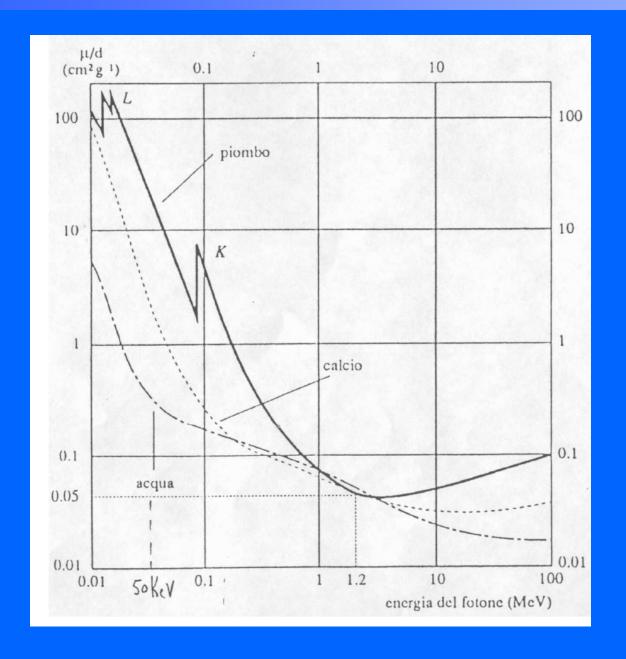
Why water?

Water is the main component of tissues

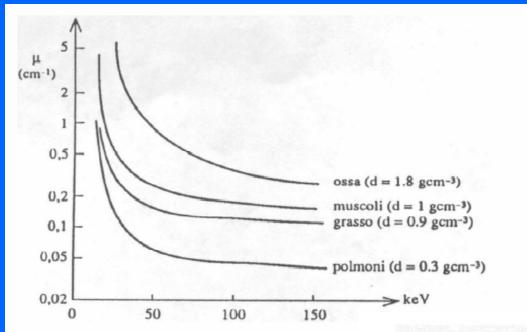


"Water equivalent" a very used concept in medical applications

Water and other materials

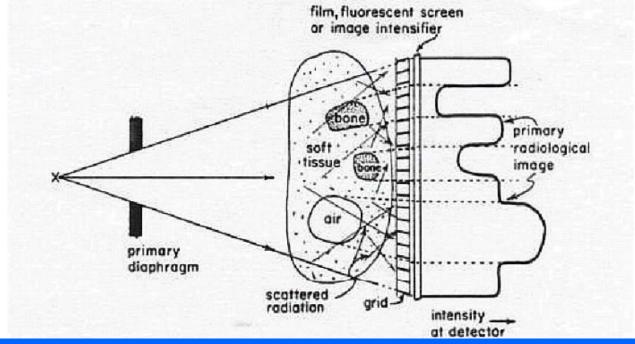


Let's get back to Roentgen



Many X-rays stop in the tissues

Dose to the patient!



Wilhelm Conrad Röntgen

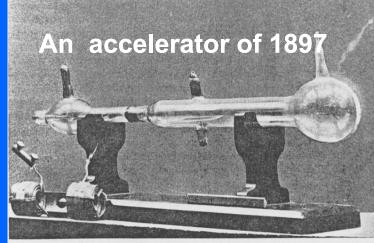


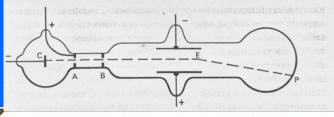
The beginning of modern physics and medical physics

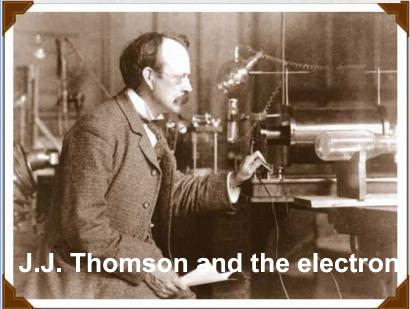
1895 – starting date of four magnificent years in experimental physics

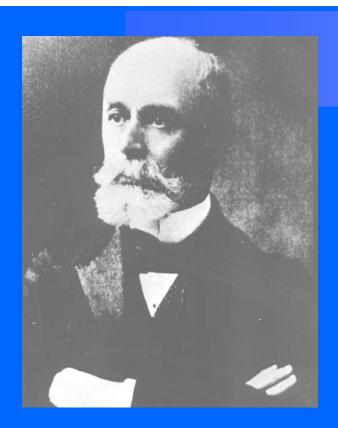
1895 discovery of X rays

Ucher eine here Art von Strahlens von W. P. Routgen. 1. Laid man dure eine Kittorf beh Vacuum. robre, oder einen geningen Evacuarten Levard schen, Crooker Elen oder, abulichen apparat der Inthadengen sines grossen Ruhmkorff: geben und dedekt the Bealand algennat urt linean Lienlich einz auliegenden Mantel aus dienen whosever Certon, so sight man in dem valla Standie berdenkellen Temmer lenen in die Waln der Wiggarate getracile, mit Barein platencyaning Angestrichenen Taparschiras her selve Entladung heir anticuerten , Hurrereiten gleichguithing ob die duzestichene oder die dubine Veits des Schimes seen to the Sung apparat durence det it the Fluoresseur it noch in 2 m Entferning oven Apparet Benerabar. Man cherryt dire lavet, dass die Urnden der Herere seems wom Francon des Entladeurs apparates un orn Klines anderen Stelle der Leitung ausgeht.







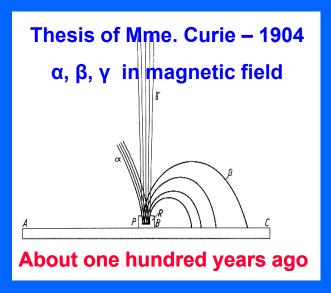


The beginning of modern physics and medical physics

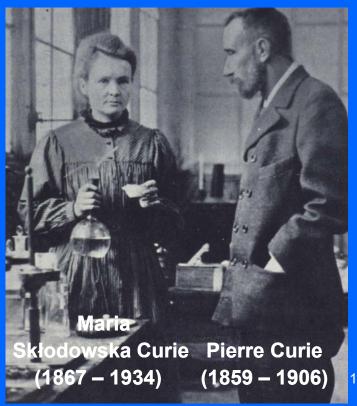
Henri Becquerel (1852-1908)

> 1896: **Discovery of natural** radioactivity

96. . Salfoh Pull D'urry & d D. Pol.



1898 **Discovery of radium**



First applications in cancer therapy



Basic concept
Local control
of the tumour



1908 : first attempts of skin cancer radiation therapy in France ("Curiethérapie")

A big step forward...

...in physics and in

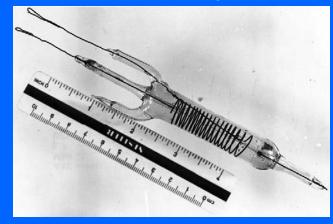
- Medical diagnostics
- Cancer radiation therapy

due to the development of three fundamental tools

- Particle accelerators
- Particle detectors
- Computers



M. S. Livingston and E. Lawrence with the 25 inches cyclotron

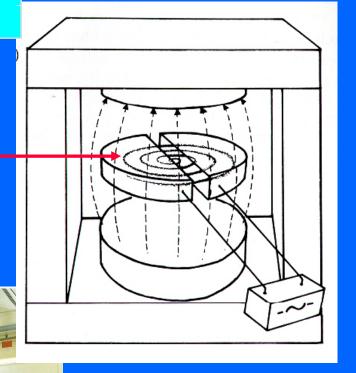


Geiger-Müller counter built by E. Fermi and his group in Rome

1930: the beginning of four other magnificent years

1930: invention of the cyclotron

Spiral trajectory of an accelerated nucleus



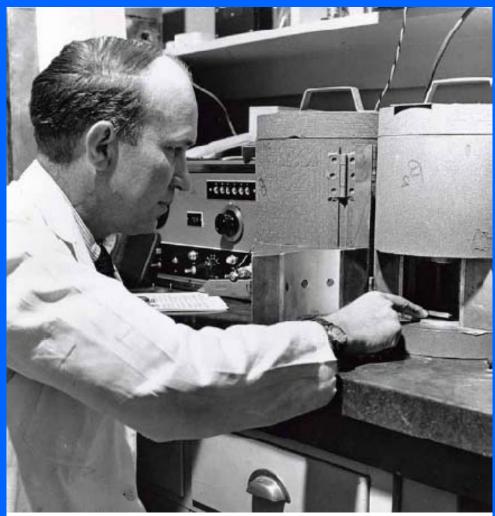




Ernest Lawrence (1901 – 1958)

A copy is on display at CERN Microcosm

The Lawrence brothers



John H. Lawrence made the first clinical therapeutic application of an artificial radionuclide when he used phosphorus-32 to treat leukemia. (1936)

- John Lawrence, brother of Ernest, was a medical doctor
- They were both working in Berkley
- First use of artificially produced isotopes for medical diagnostics and therapy
- Beginning of nuclear medicine

An interdisciplinary environment helps innovation!

Discovery of the neutron

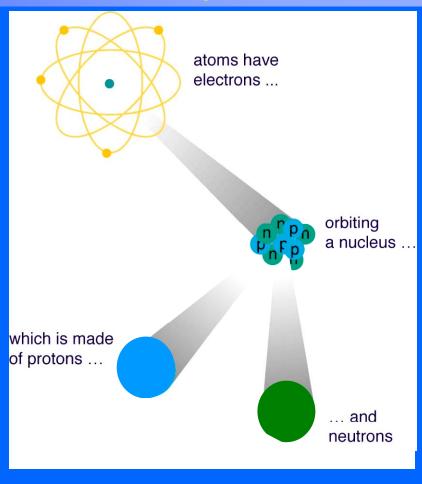


1932

James Chadwick (1891 – 1974)



Student of Ernest Rutherford



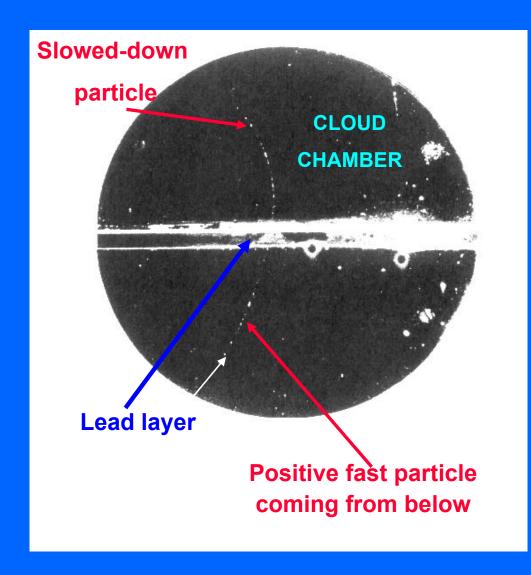
Neutrons are used today to

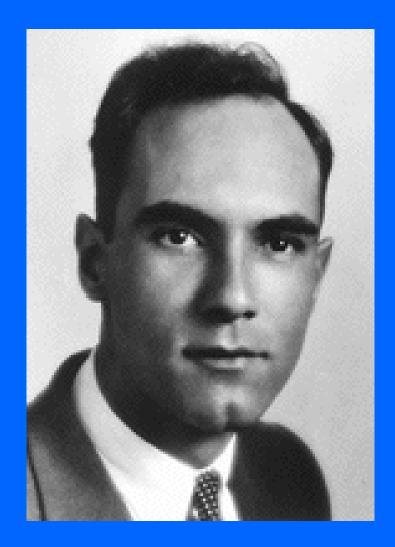
- Produce isotopes for medical diagnostics and therapy
- Cure some kind of cancer

Matter and antimatter...



1932 – discovery of antimatter: the positron

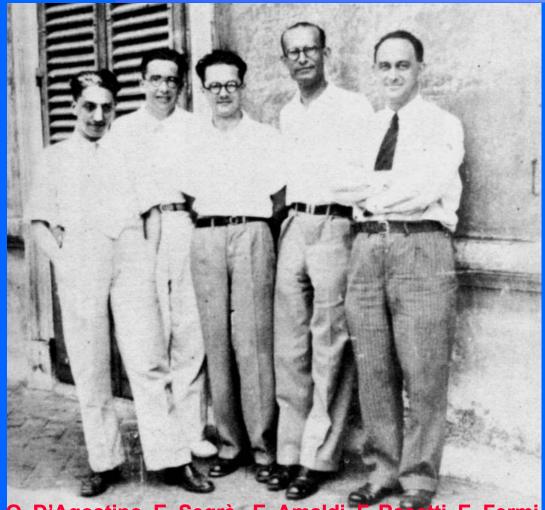




Carl D. Anderson - Caltech

The positron is at the basis of Positron Emission Tomography (PET)

Discovery of the effectiveness of slow neutrons



O. D'Agostino E. Segrè E. Amaldi F. Rasetti E. Fermi

1934

First radioisotope of lodine

among fifty new artificial species

RADIOATTIVITÀ «BETA» PROVOCATA DA BOMBARDAMENTO DI NEUTRONI. — III.

E. Amaldi, O. D'Agostino, E. Fermi, F. Rasetti, E. Segrè Ric. Scientifica *, 5 (1), 452-453 (1934).

Sono state proseguite ed estese le esperienze di cui alle Note precedenti (*) coi risultati che ricordiamo appresso.

Idrogeno – Carbonio – Asoto – Ossigeno. – Non dànno effetto apprezzabile. Sono stati esaminati paraffina irradiata al solito modo per 15 ore con una sorgente di 220 mC, acqua irradiata per 14 ore con 670 mC e carbonato di guanidina irradiato per 14 ore con 500 mC.

Fluoro. – Il periodo del Fluoro è sensibilmente minore di quanto indicato precedentemente e cioè di pochi secondi.

Magnesio. - Il Magnesio ha due periodi, uno di circa 40 secondi e uno più lungo.

Alluminio. – Oltre al periodo di 12 minuti segnalato precedentemente ve ne è anche un altro dell'ordine di grandezza di un giorno. L'attività corrispondente a questo secondo periodo segue le reazioni chimiche caratteristiche del Sodio. Si tratta probabilmente di un Na²⁴.

Zolfo. – Il periodo dello S è assai lungo, certamente di molti giorni. L'attività si separa con le reazioni caratteristiche del Fosforo.

Cloro. - Si comporta analogamente allo S. Anche qui si può separare

corrispondente a questo secondo periodo segue le reazioni chimiche caratteristiche del Sodio. Si tratta probabilmente di un Na²⁴.

Zolfo. – Il periodo dello S è assai lungo, certamente di molti giorni. L'attività si separa con le reazioni caratteristiche del Fosforo.

Cloro. – Si comporta analogamente allo S. Anche qui si può separare un principio attivo; probabilmente si tratta di un P³² identico a quello che si ricava dallo S.

Manganese. – Ha un effetto debole con un periodo di circa 15 minuti. Cobalto. – Ha un effetto di 2 ore. Il principio attivo si comporta come Mn. Data l'identità di periodo e di comportamento chimico si tratta quasi certo di un Mn⁵⁶ identico a quello che si forma irradiando il Fe.

Zinco. - Ha due periodi, uno di 6 minuti e uno assai più lungo.

Gallio. - Periodo 30 minuti.

Bromo. – Ha due periodi, uno di 30 minuti e l'altro di 6 ore. L'attività corrispondente al periodo lungo e probabilmente anche l'altra, seguono chimicamente il Br.

Palladio. - Periodo di alcune ore.

Jodio. - Periodo 30 minuti. L'attività segue chimicamente lo Jodio.

Praseodimio. - Ha due periodi. Uno di 5 minuti e l'altro più lungo. Neodimio. - Periodo 55 minuti.

Samario. - Ha due periodi uno di 40 minuti e uno più lungo.

Oro. - Periodo dell'ordine di grandezza di 1 o 2 giorni.

Also for physicists patents are important!

Discovery: Saturday 20 October (*)

Patent: Friday 26 October because of Orso Mario Corbino



"To obtain radioactive substances in quantities of practical importance"

(*) A. De Gregorio: not on October 22!



Istituto Superiore di Sanità - 1934

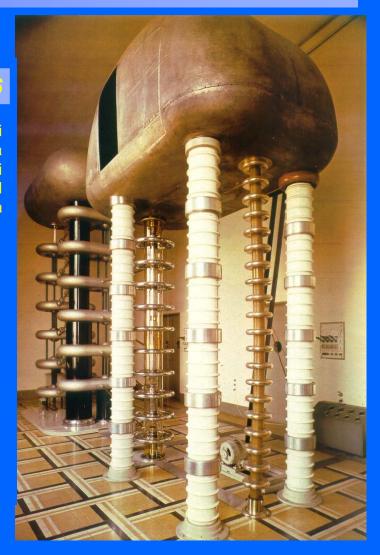


"Il tubo" – 1 MeV Cockcroft-Walton ion accelerator - 1938

Letter by D. Marotta to E. Fermi - 16.11.36

I nuovi orizzonti che ha aperto per la terapia dei tumori maligni la possibilità di fabbricare sostanze radioattive artificiali in quantità considerevoli mi fa pensare alla convenienza che l'Istituto di Sanità faccia il possibile per organizzare i mezzi tecnici per tali preparazioni. Prima però di prendere in considerazione il progetto proposto dal Capo del Labratorio di Fisica di questo Istituto desidererei avere il parere di Vostra Eccellenza.

The new vistas opened for tumour therapy by the possibility of producing large quantities of radioactive substances convince me that it is convenient for 'Istituto di Sanità' to procure the technical means for such productions. However, before considering the project proposed by the Chief of the Physics Laboratory (G. C. Trabacchi), I would like to have the opinion of Your Excellency.



Fermi and the use of radio isotopes in medicine

Lecture by Enrico Fermi at Istituto di Sanità Pubblica 29.5.1938

PROSPETTIVE DI APPLICAZIONE DELLA RADIOATTIVITÀ ARTIFICIALE

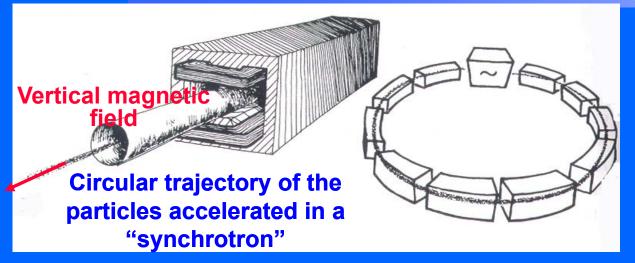
"It can be forseen WITHOUT DOUBTS that the (new) radioactive substances will find THERAPEUTICAL APPLICATIONS similar to the one of natural occurring radioactive substances.

Moreover and independently, the use of large quantities of radioactive substances will open, I HOPE, the way to many interesting studies in biology and chemistry through the use of radioelements as 'INDICATORS' "

È da prevedere senz'altro che le sostanze radioattive artificiali troveranno un impego terapeutico analogo a quello delle sostanze radioattive naturale.

Ma anche indipendentemente da queste possibilità, l'uso delle sostanze radioattive artificiali in quantità rilevanti renderà possibili, io spero, anche molte interessanti ricerche nel campo della biologia e della chimica, usando i radioelementi come "indicatori".

Four other crucial years: the synchrotron



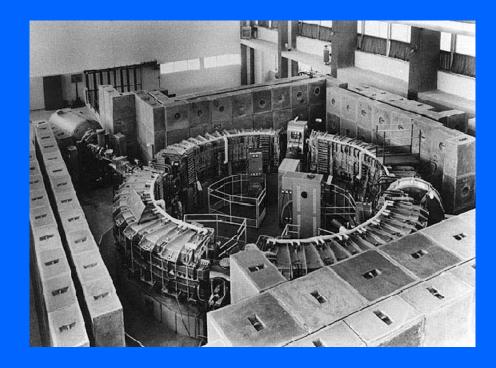
1944 principle of phase stability

1 GeV electron synchrotron Frascati - INFN - 1959



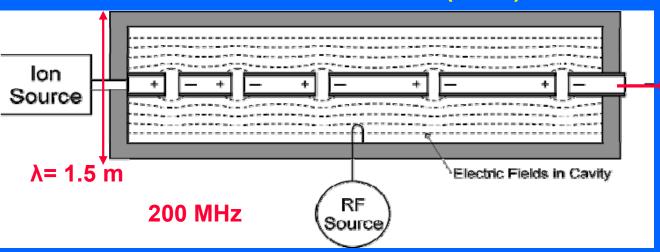
Veksler visits McMillan

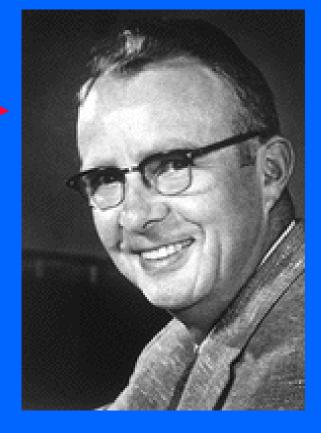
1959 - Berkeley



Radio-frequency linacs for protons and ions

Linear accelerator (linac)







100 MeV linac on display at CERN Microcosm

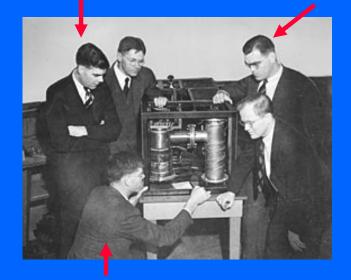
L. Alvarez

1946 – Drift Tube Linac

The electron linac

Sigurd Varian

William W. Hansen

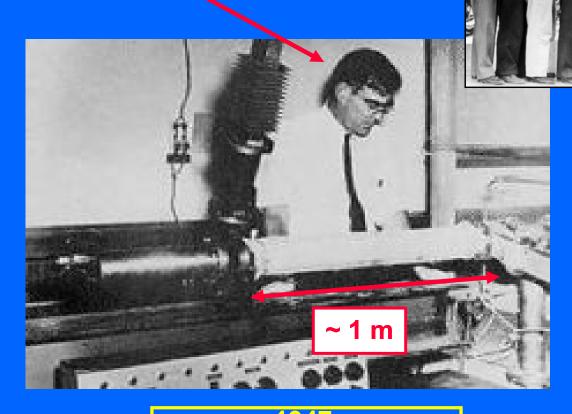


Russell Varian

1939

Invention of the klystron

The electron linac is used today in hospital based conventional radiation therapy facilities



1947
first linac for electrons
4.5 MeV and 3 GHz

The beginning of CERN 50 years ago



Isidor Rabi
UNESCO talk in 1950

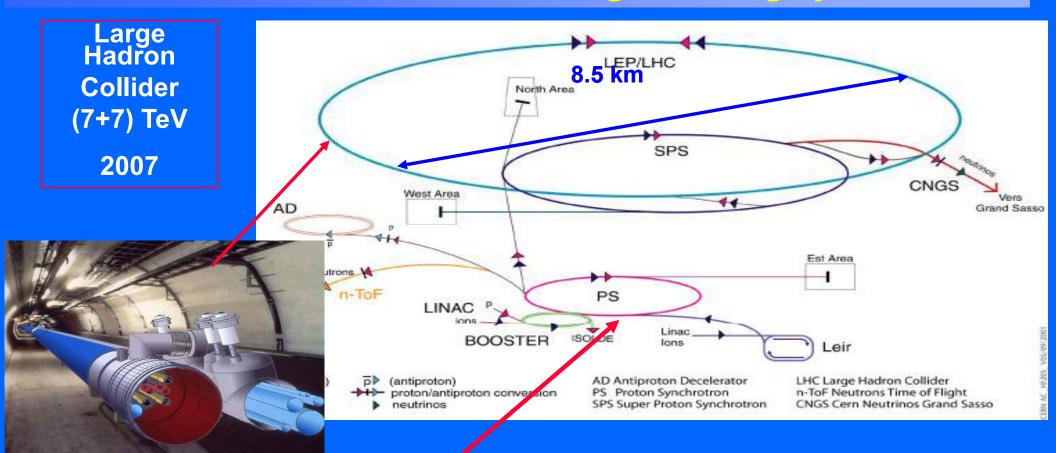


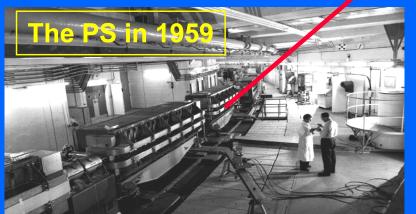
1952: Pierre Auger Edoardo Amaldi Secretary General

1952-54

at the meeting that created the provisional CERN

At CERN we have linacs and strong-focusing synchrotrons





In 1952 the "strong-focusing" method invented at BNL (USA) was chosen for the CERN PS

Accelerators running in the world

| CATEGORY OF ACCELERATORS | NUMBER IN USE (*) | |
|---|-------------------|--|
| High Energy acc. (E >1GeV) | ~120 | |
| Synchrotron radiation sources | <u>>100</u> | |
| Medical radioisotope production | <u>~200</u> | |
| Radiotherapy accelerators | > 7500 > 9000 | |
| Research acc. included biomedical research | ~1000 | |
| Acc. for industrial processing and research | ~1500 | |
| Ion implanters, surface modification | >7000 | |
| TOTAL | <u>> 17500</u> | |
| (*) W. Masia-awaki and W. Cabarfi Int. L. of Dadiation Oncology, 2004 | | |

(*) W. Maciszewski and W. Scharf: Int. J. of Radiation Oncology, 2004

About half are used for bio-medical applications

Particle detectors

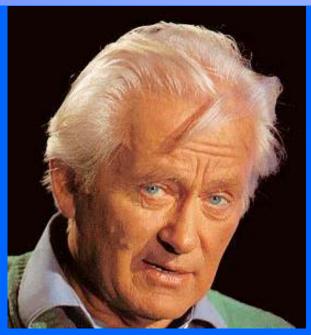
They are the "eyes" of particle physicists

- A very impressive development in the last 100 years
 - From the Geiger counter to ATLAS and CMS at CERN!

Crucial in many medical applications

One example: the multiwire proportional chamber

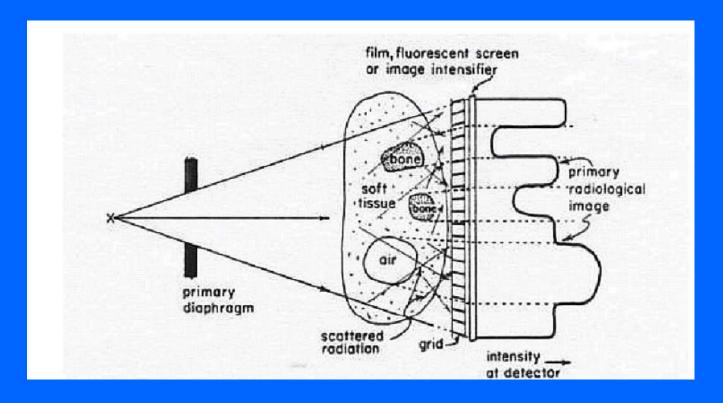




Georges Charpak, CERN physicist since 1959,
Nober prize 1992

- Invented in 1968, launched the era of fully electronic particle detection
- Used for biological research and could eventually replace photographic recording in applied radio-biology
- The increased recording speeds translate into faster scanning and lower body doses in medical diagnostic tools based on radiation or particle beams

Radiography and imaging with radiations



General features:

Sensitivity of the detector = less dose to the patient Granularity of the detector = better image definition Speed of the detector = detection of movements

End of part I