

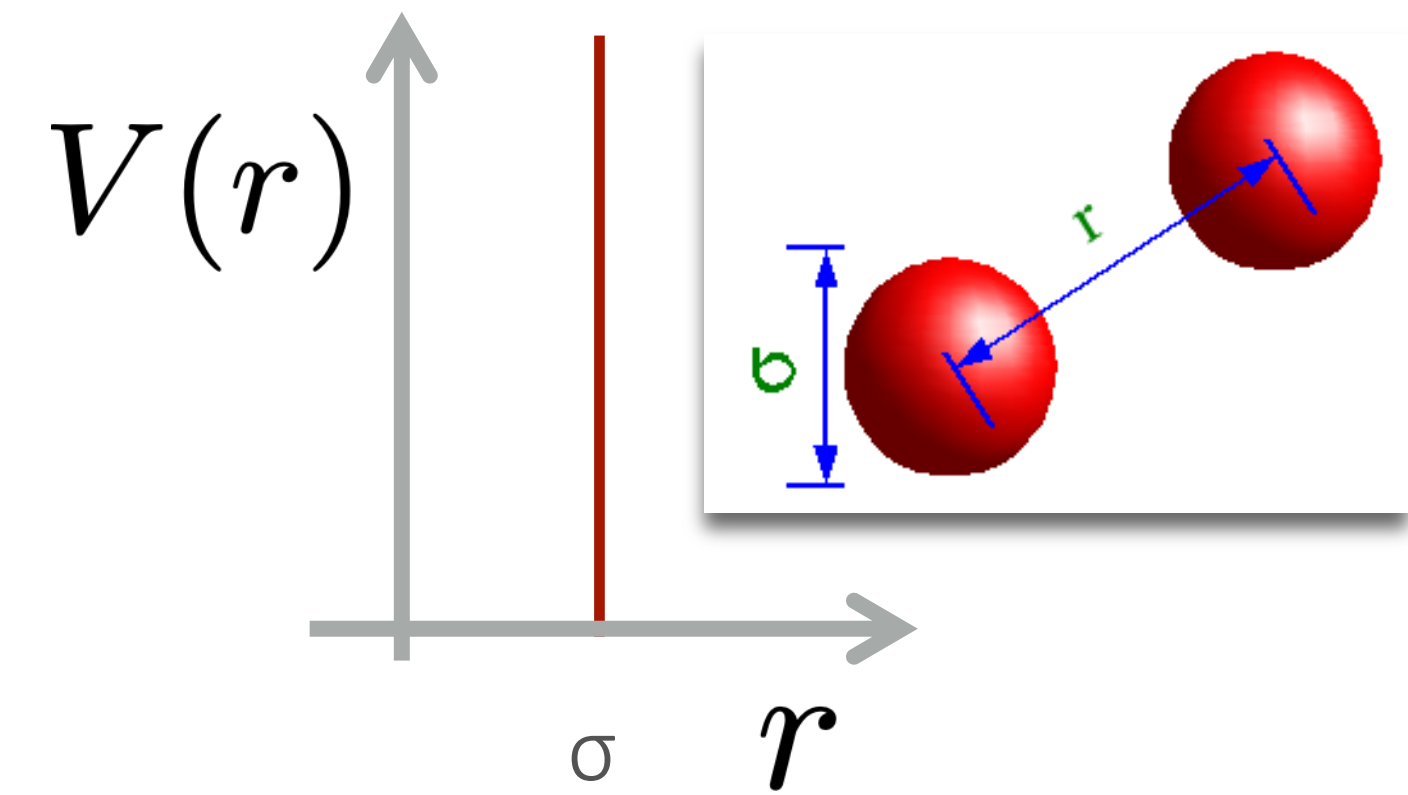
## PROTEIN PHASE DIAGRAMS

Emanuela Zaccarelli

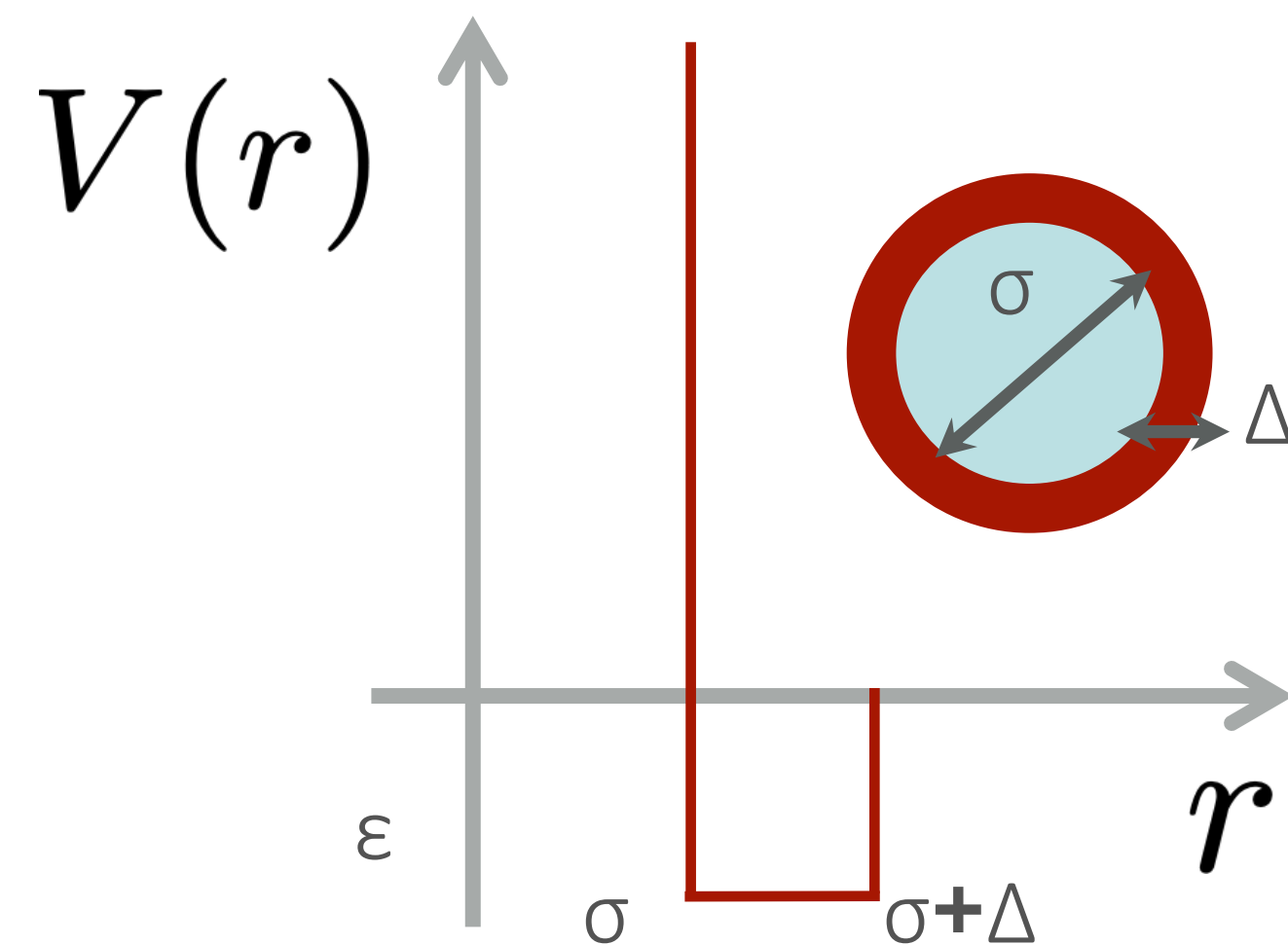
CNR Institute for Complex Systems and  
Department of Physics, Sapienza University, Rome



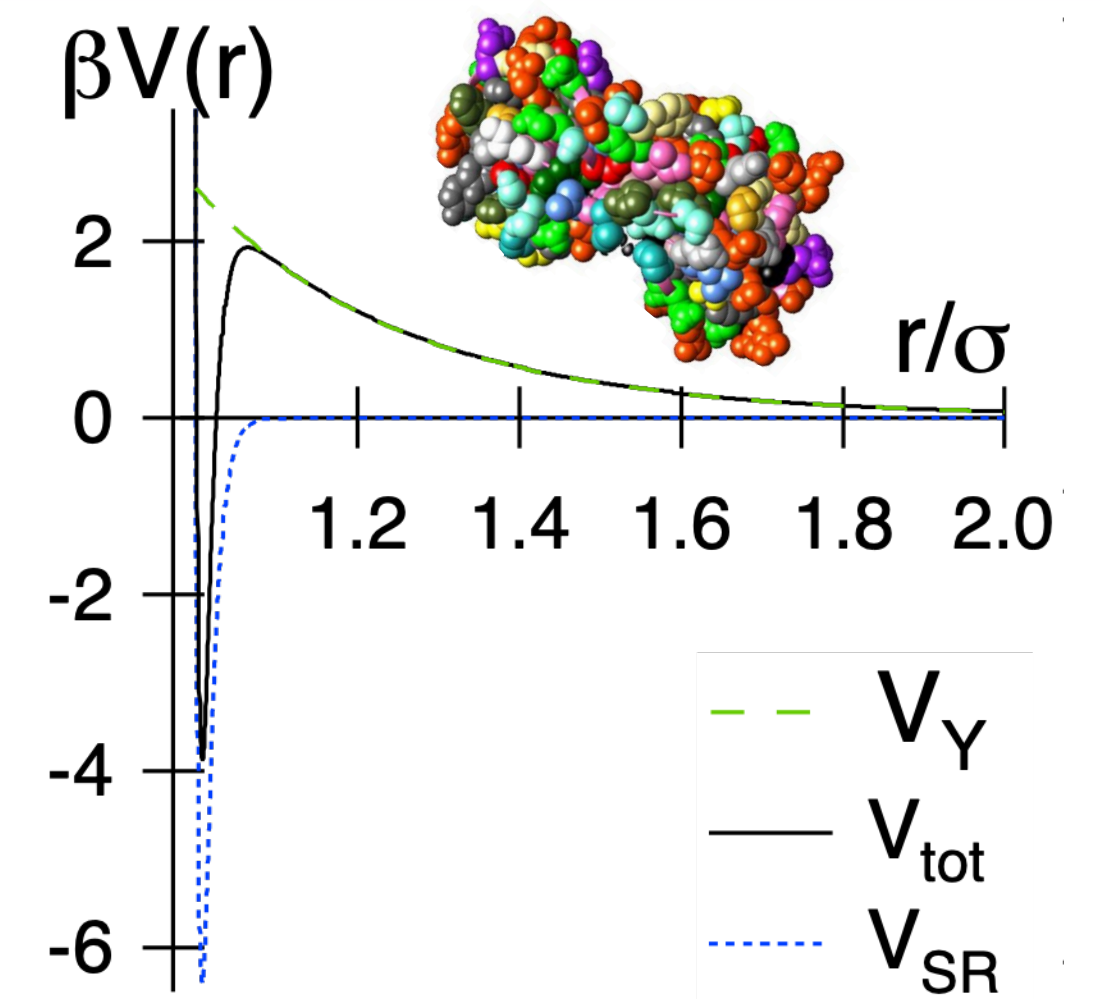
# Soft Matter models I have enjoyed working with...



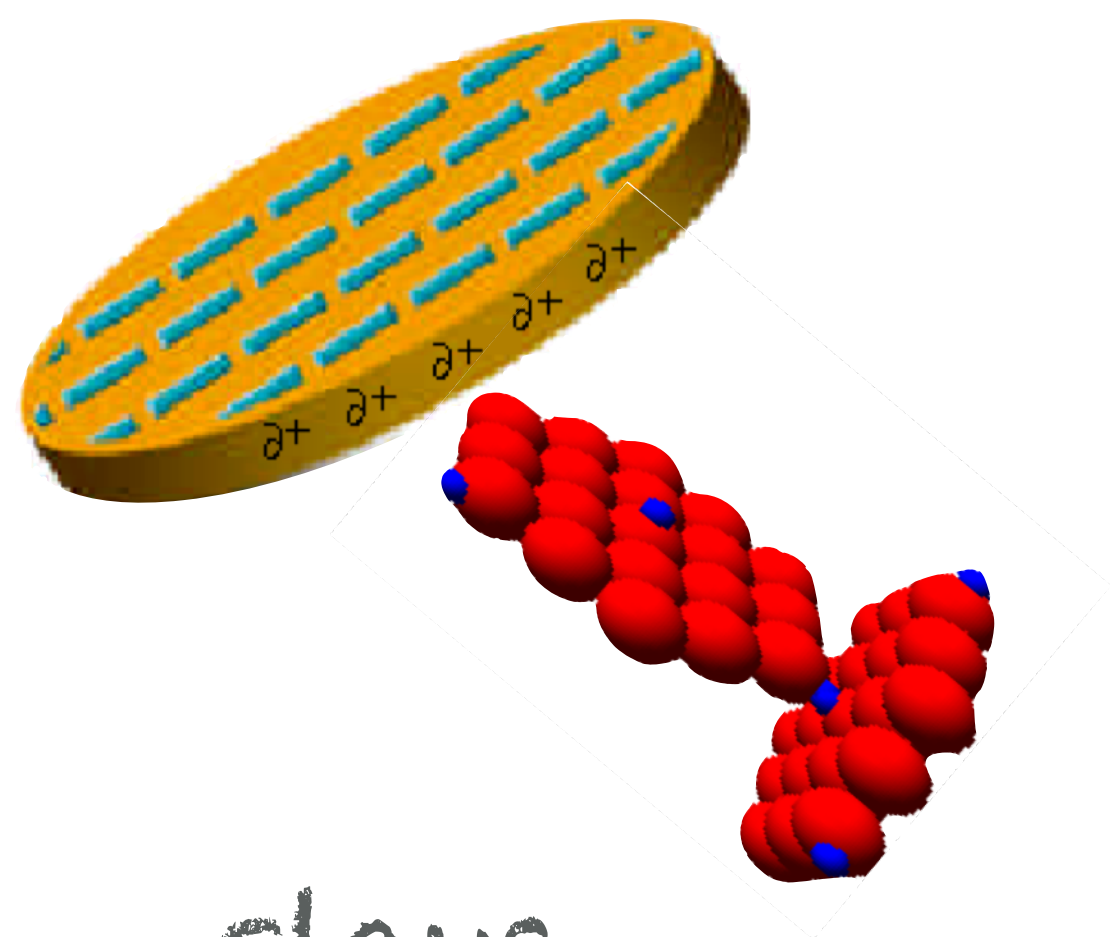
hard spheres



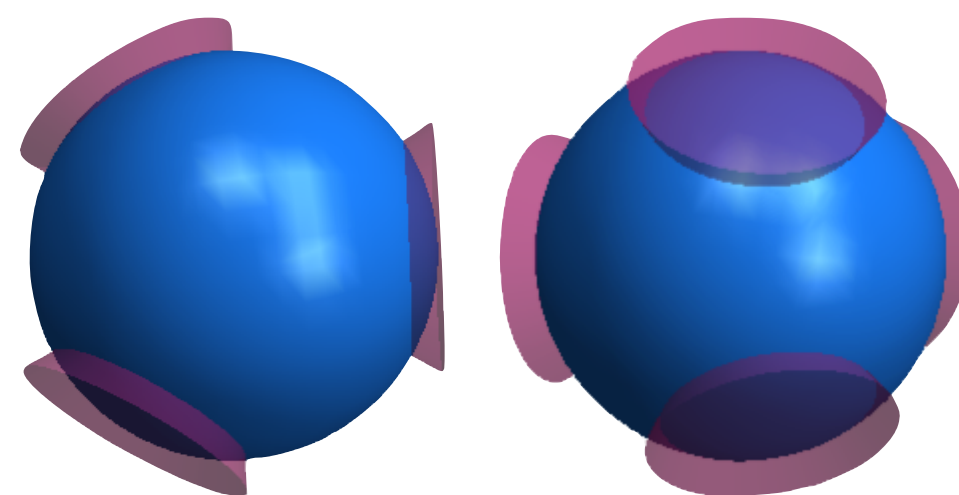
attractive colloids



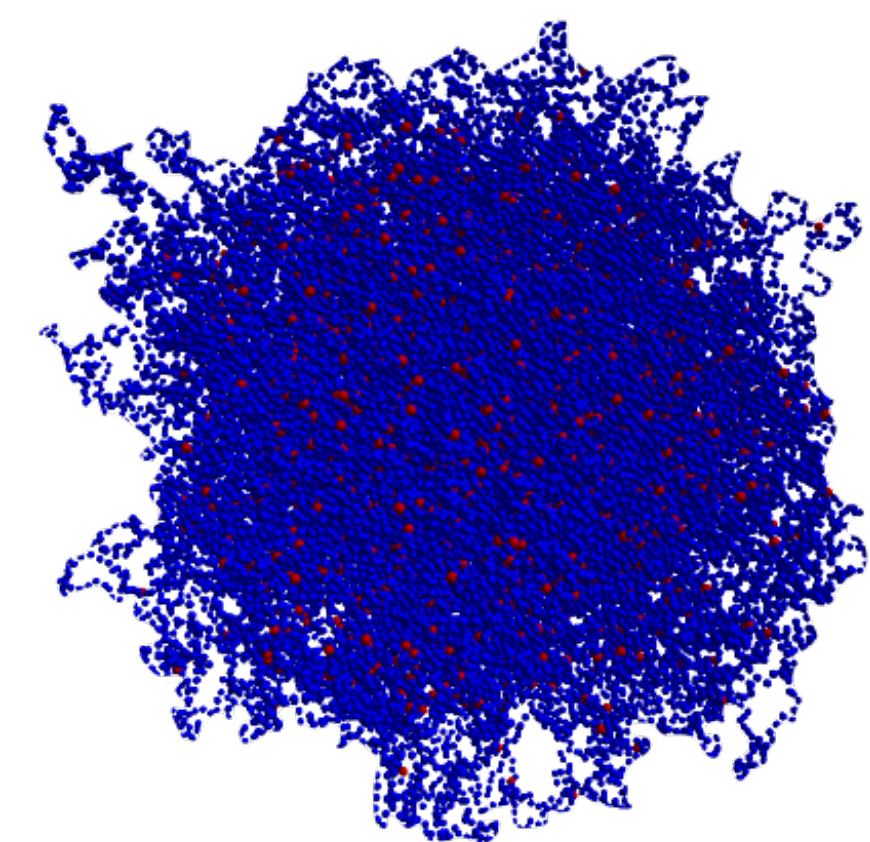
globular proteins



clays



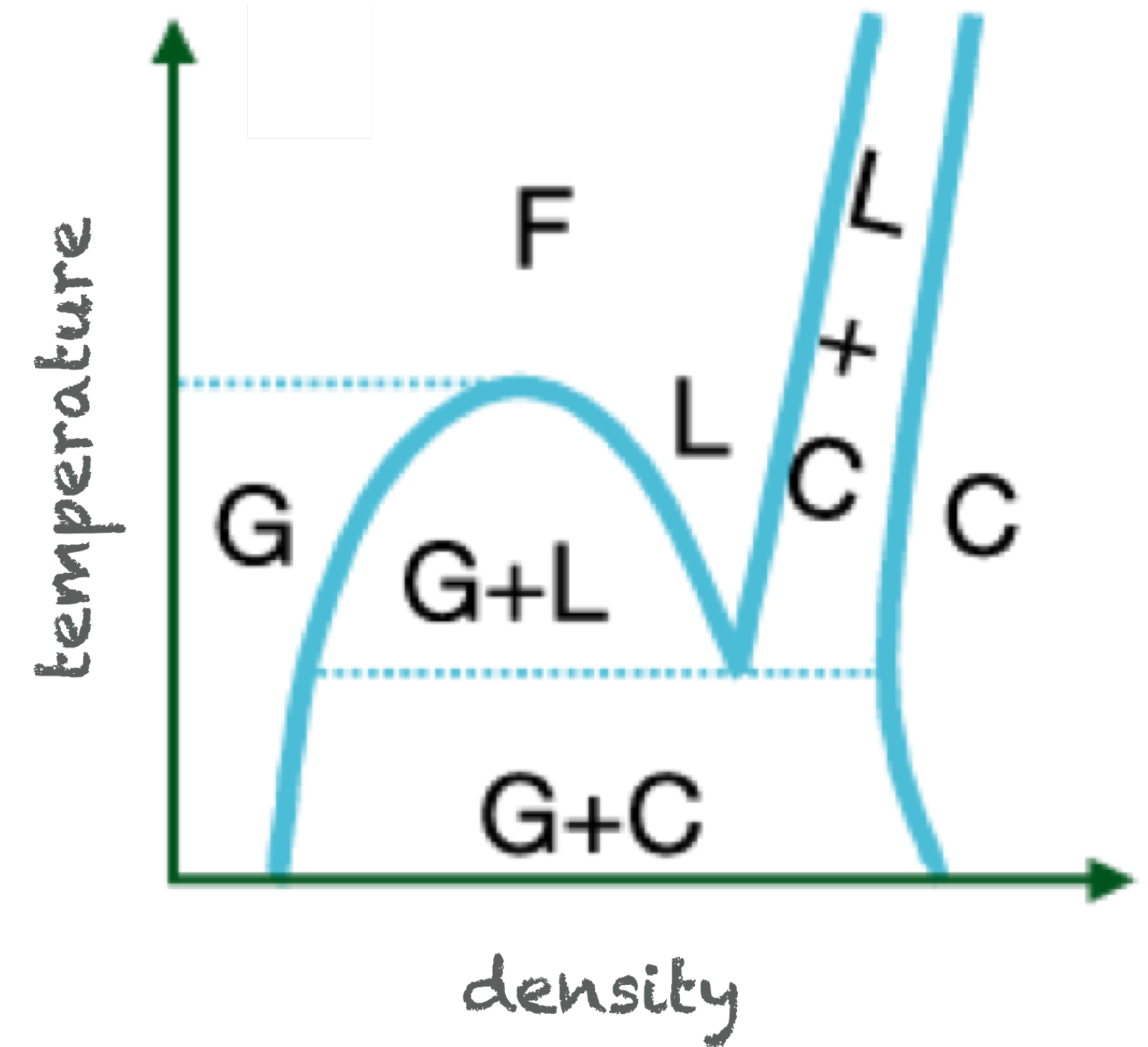
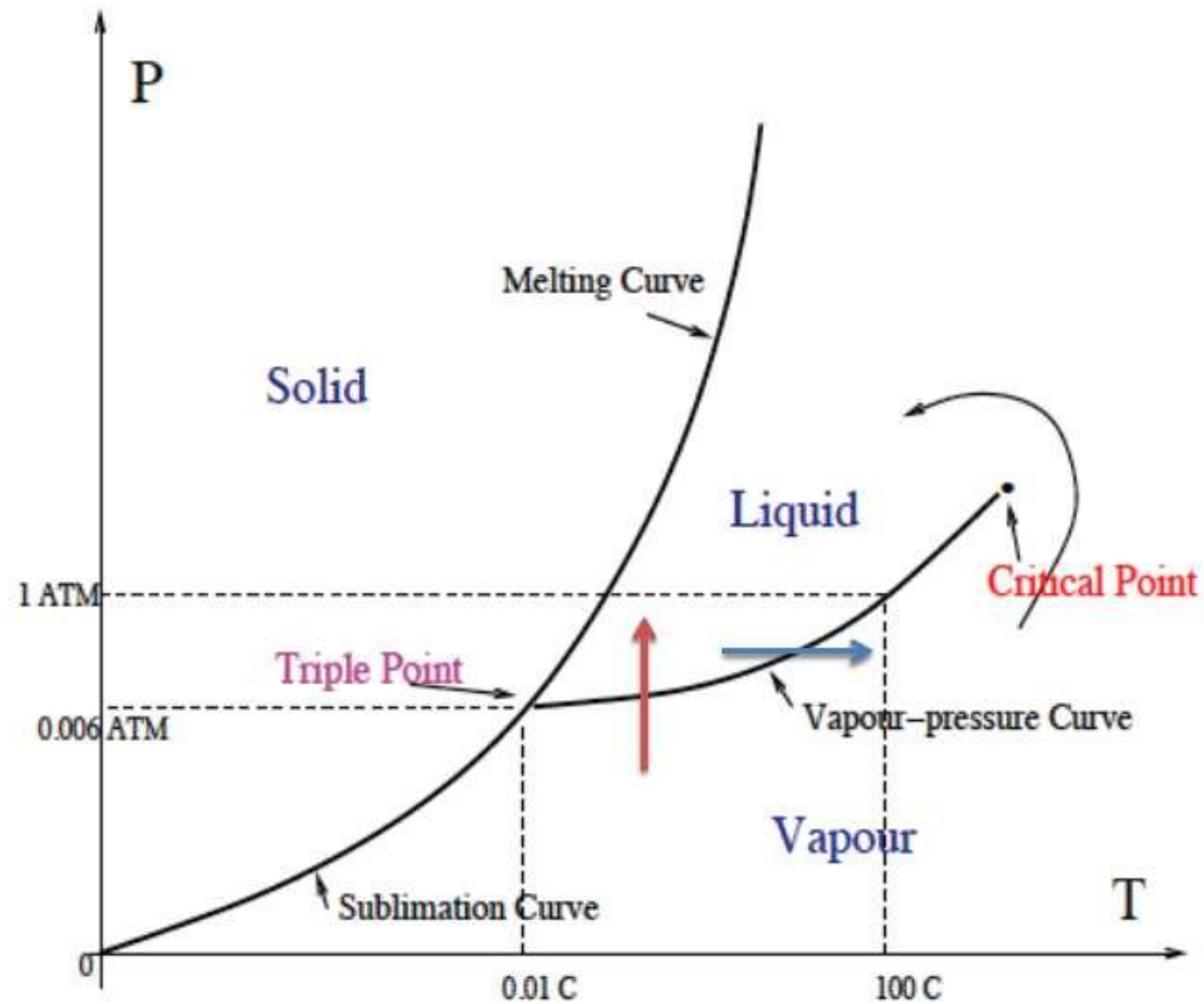
patchy particles



microgels, star polymers

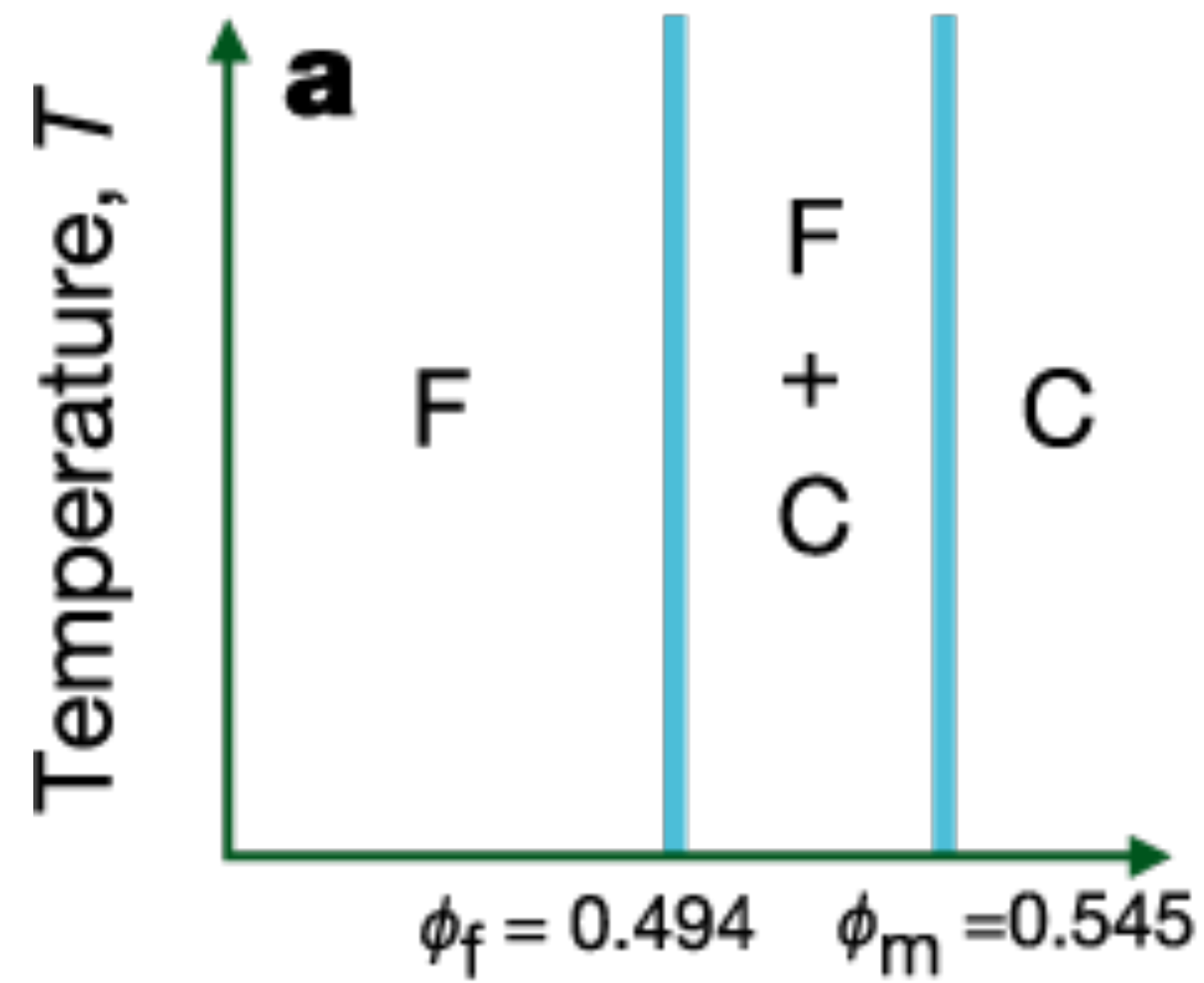
# Phase diagrams

maps of (thermodynamic) stability  
atomic/molecular liquids

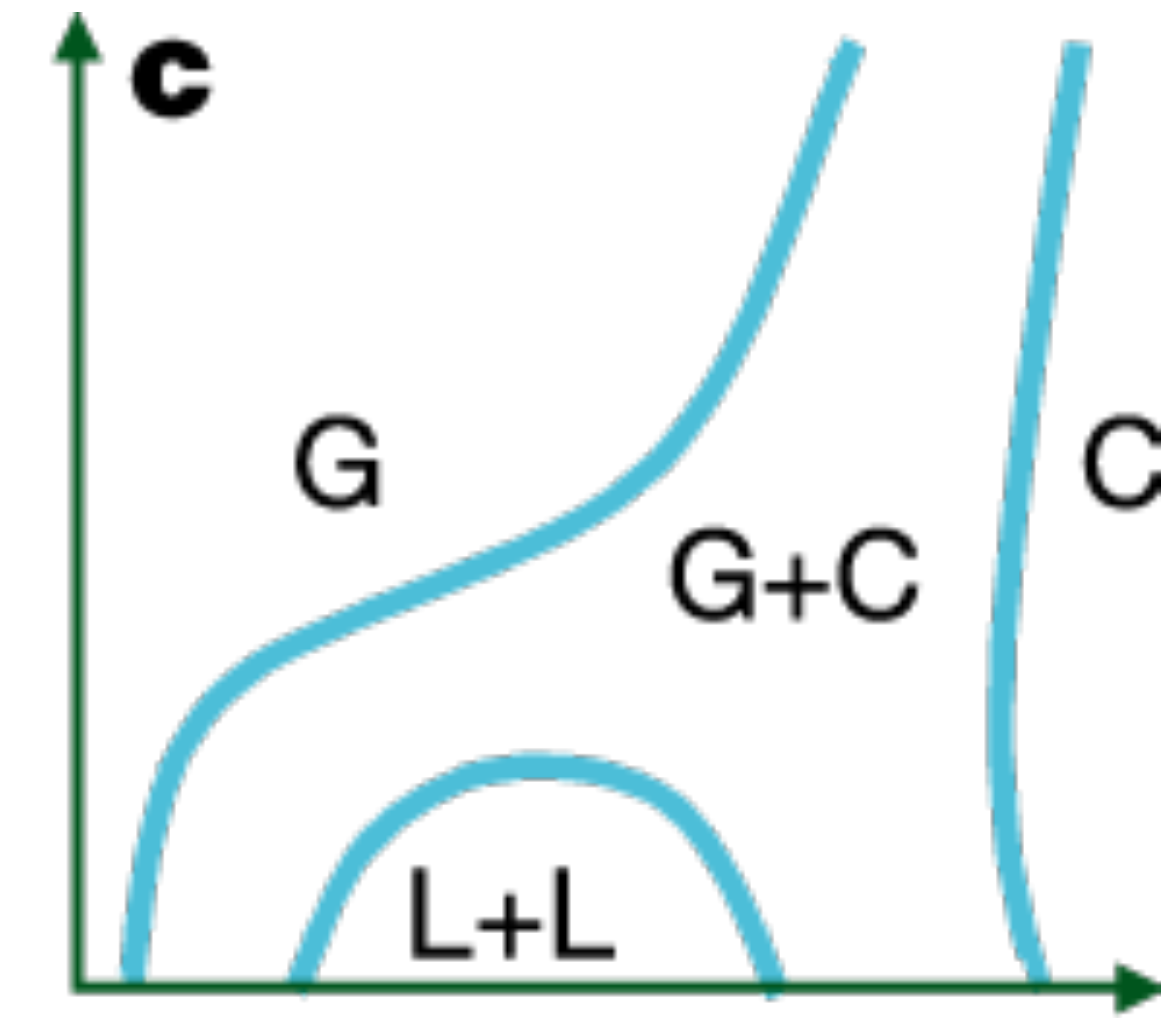
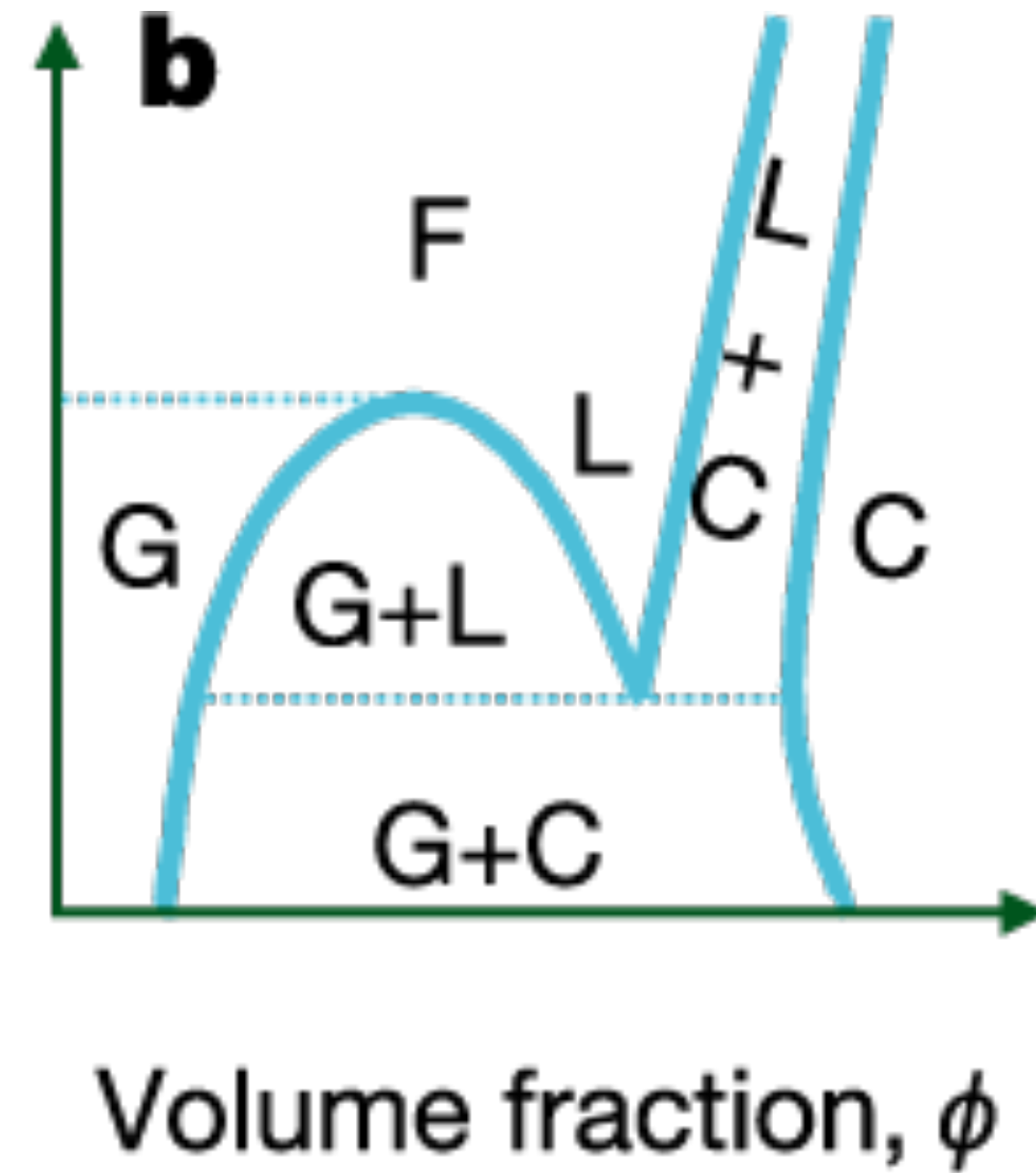


# Colloidal phase diagrams

hard spheres



short-range attractions

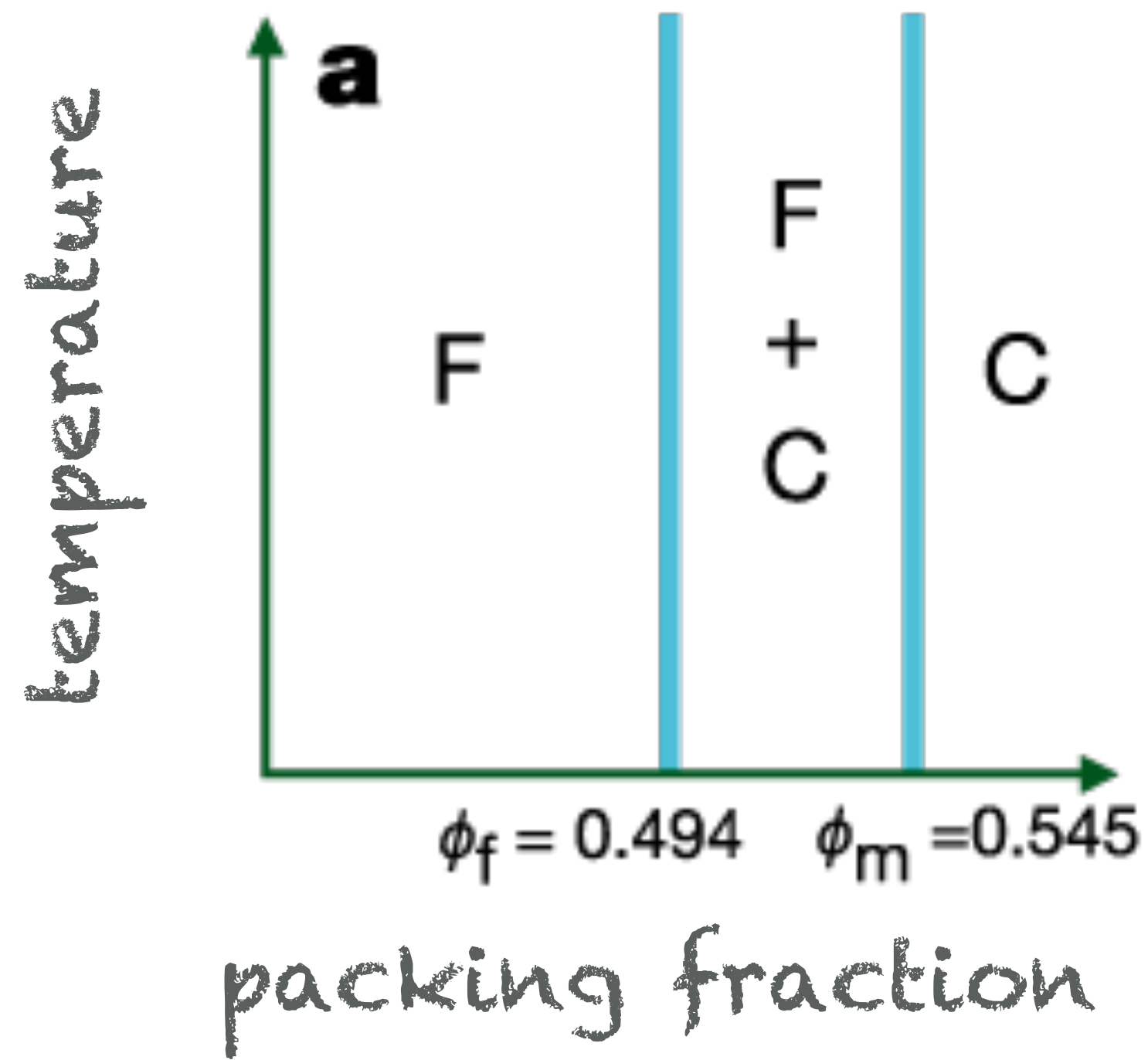


Anderson and Lekkerkerker  
Nature 2002

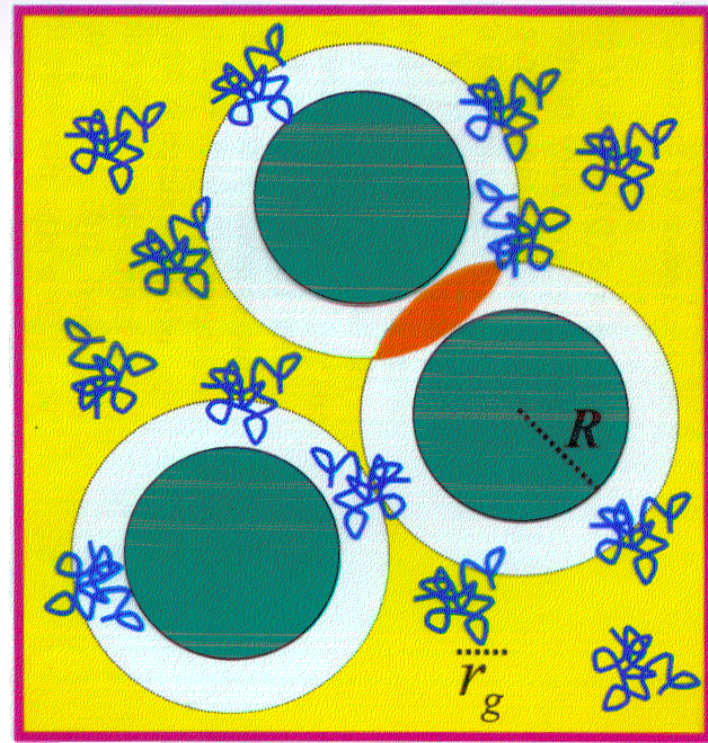
(atomic/molecular liquids)

colloids with long-range attraction

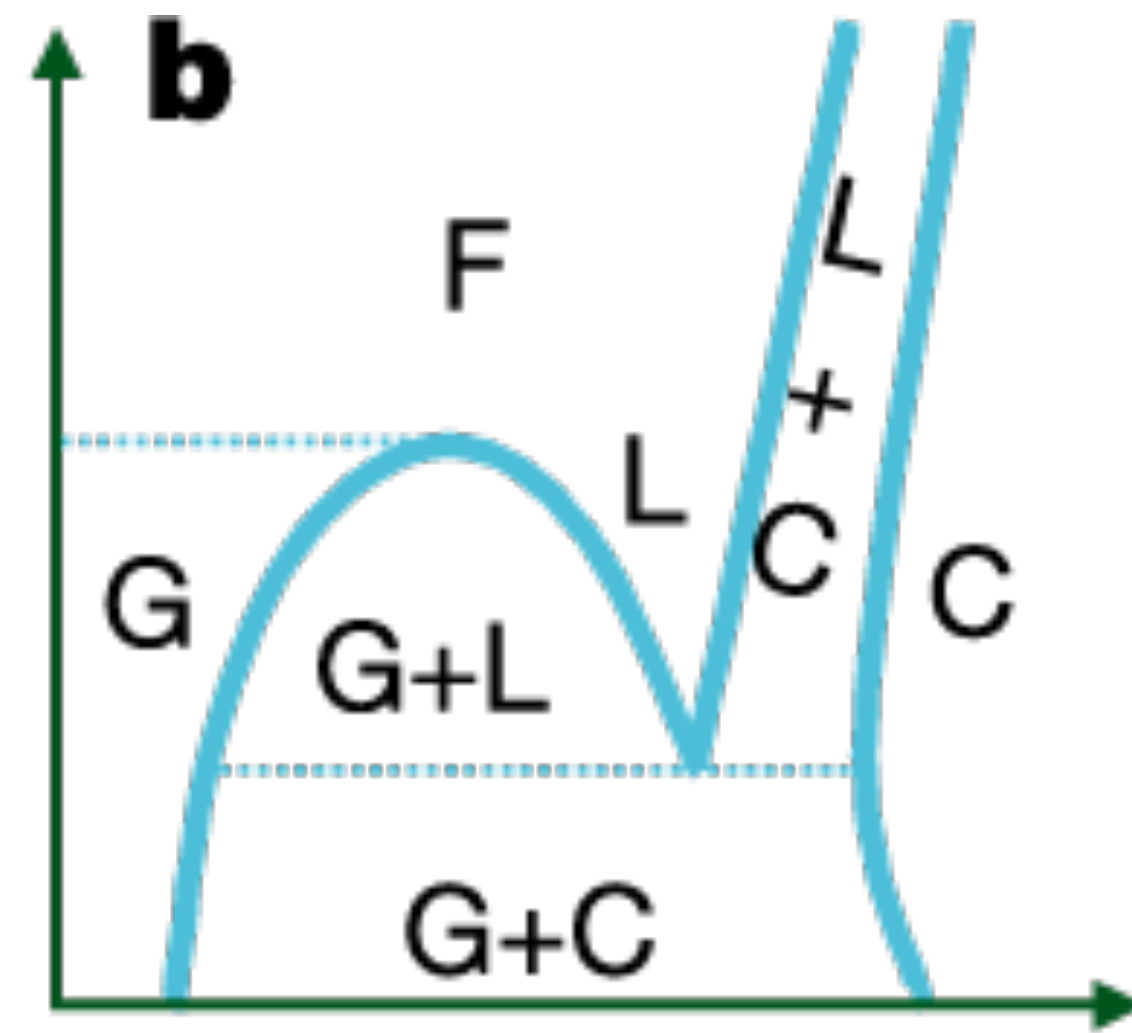
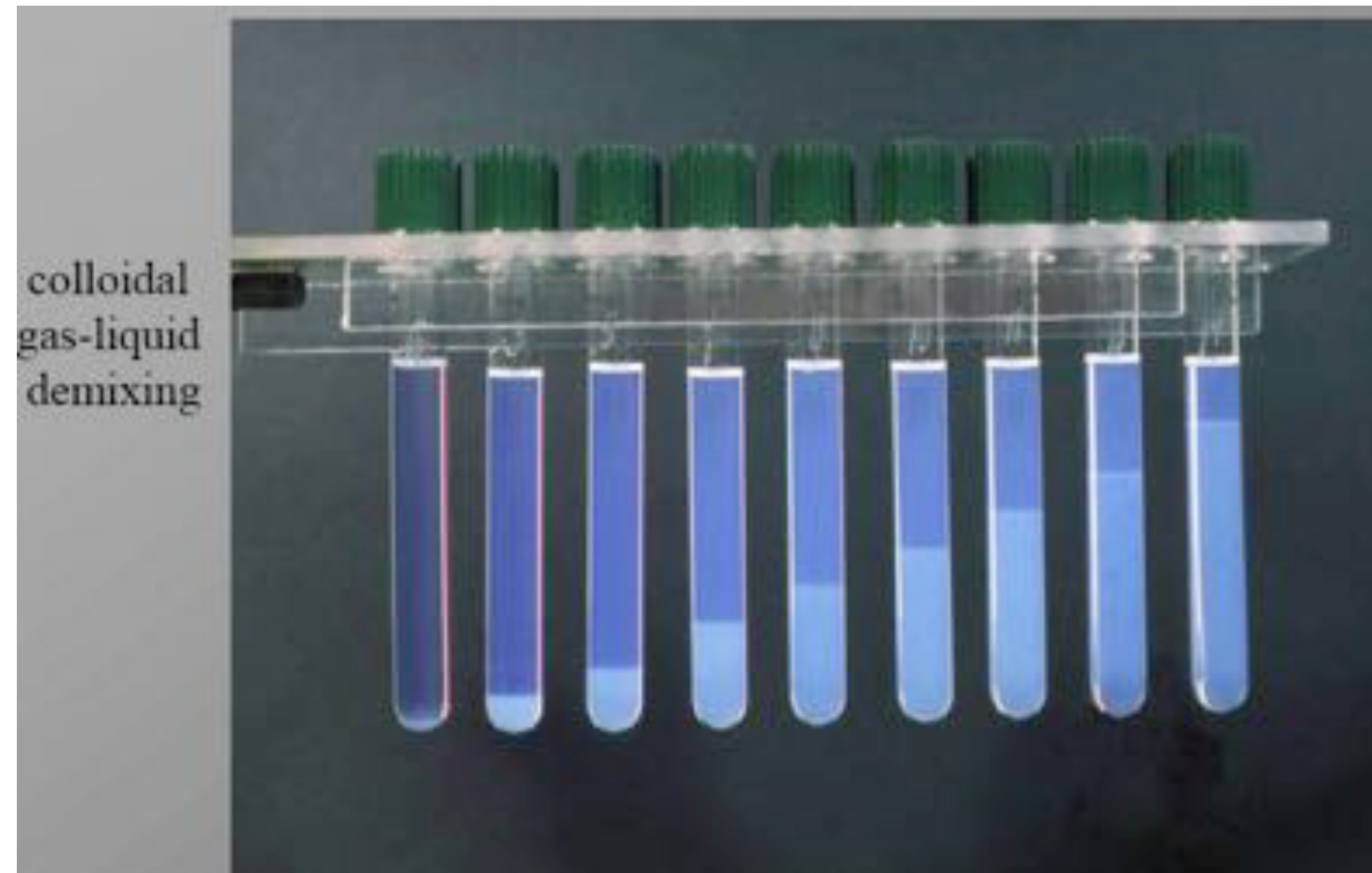
# Hard spheres



# Attractive colloids

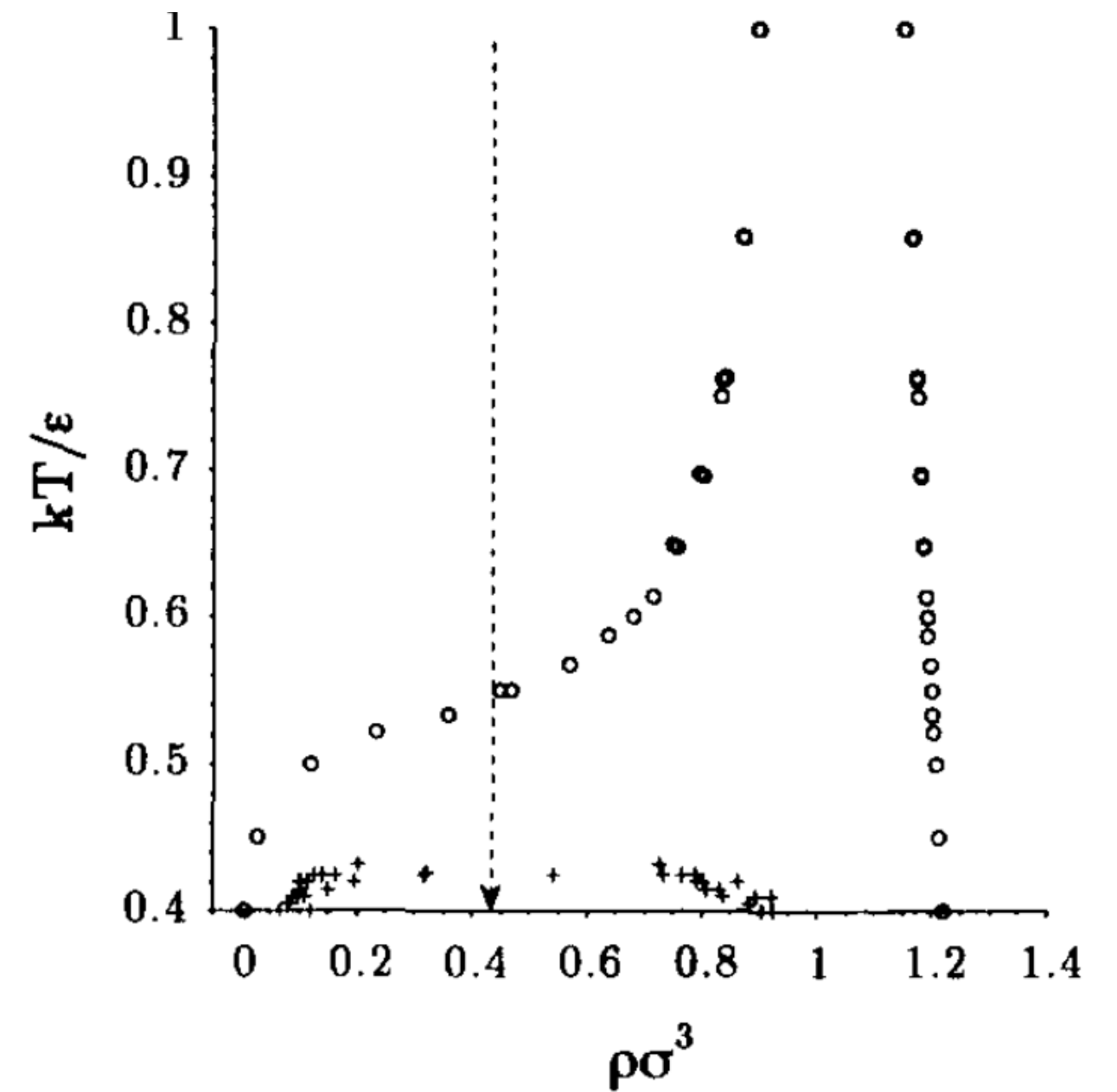
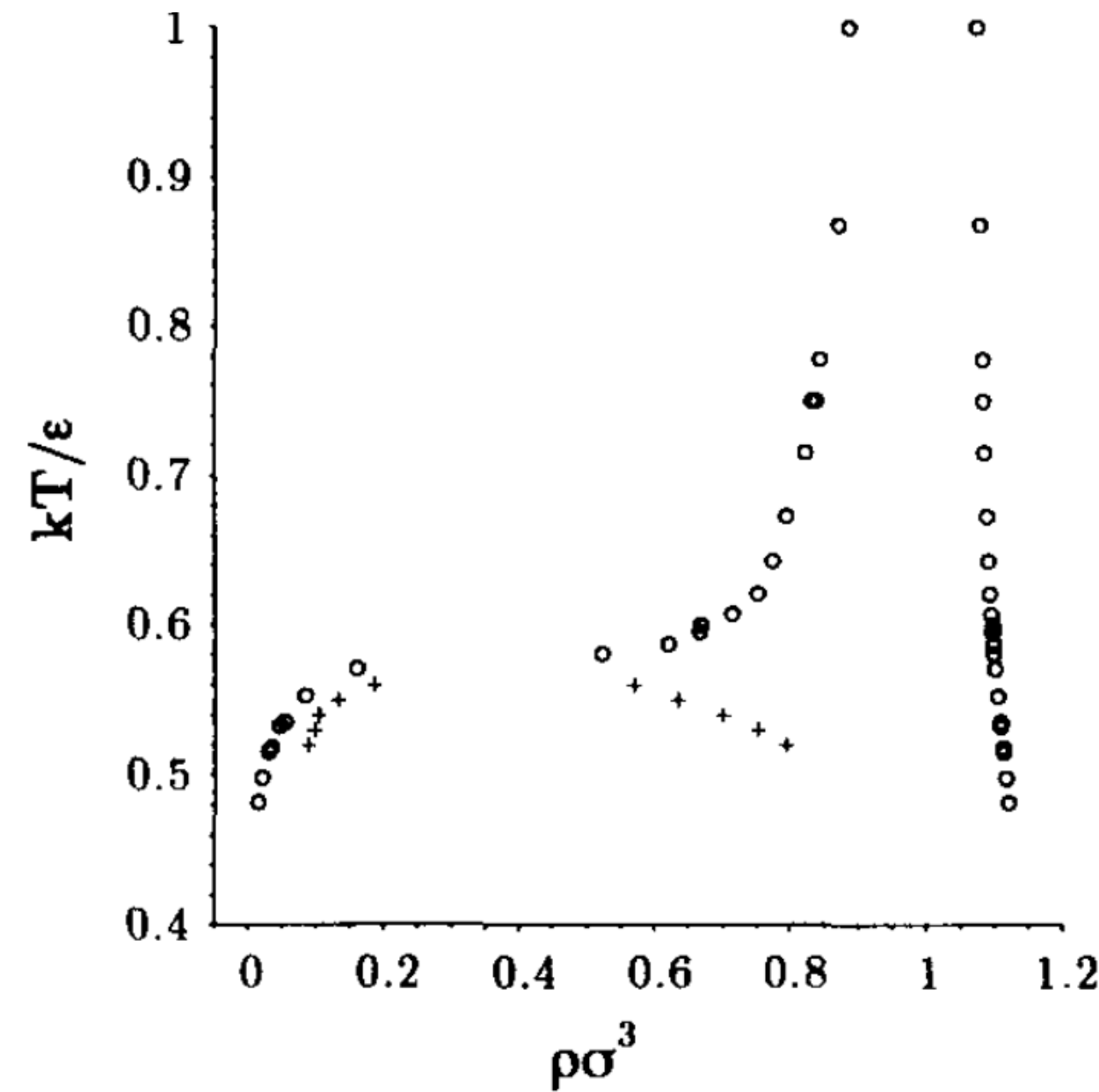
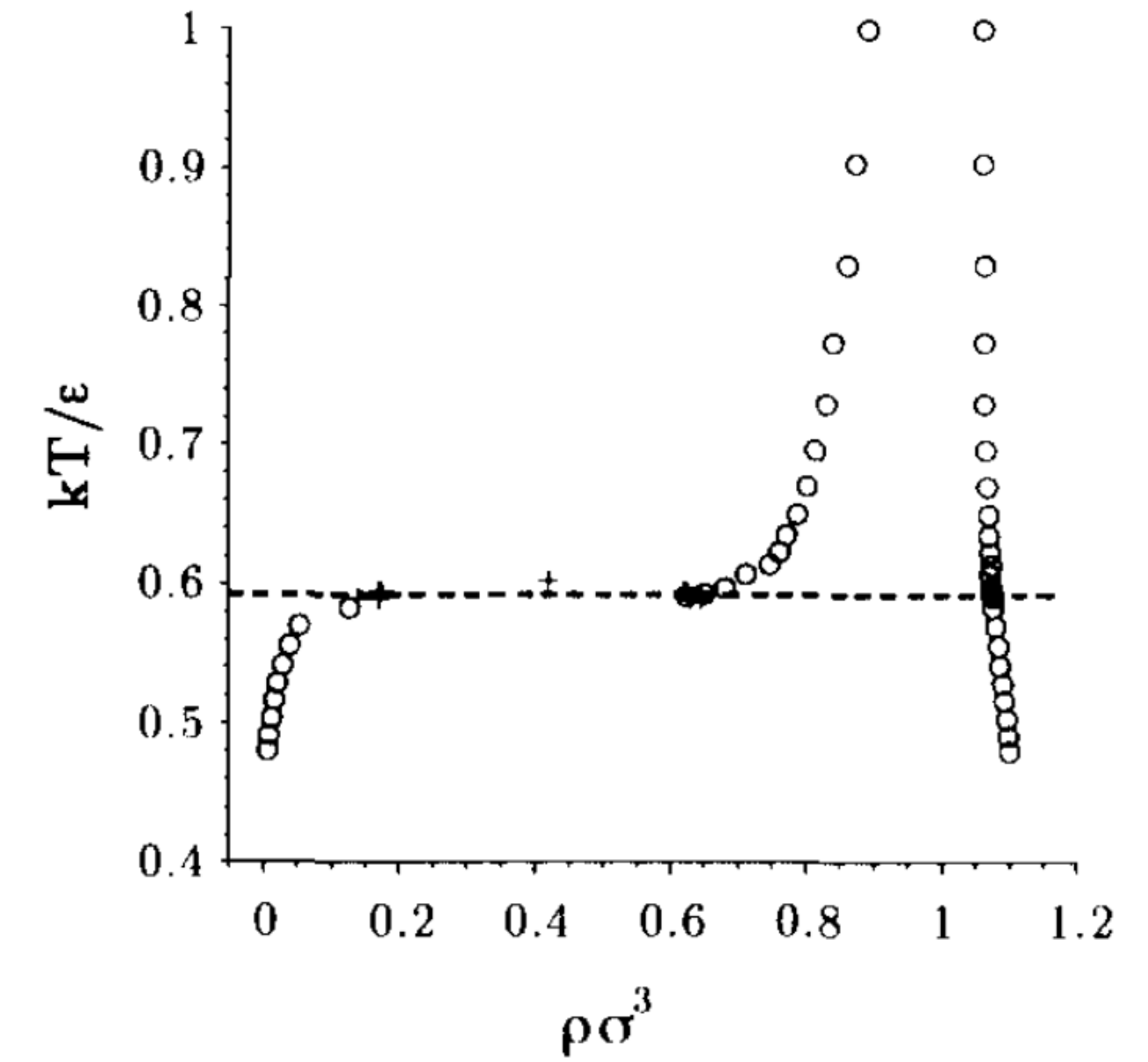
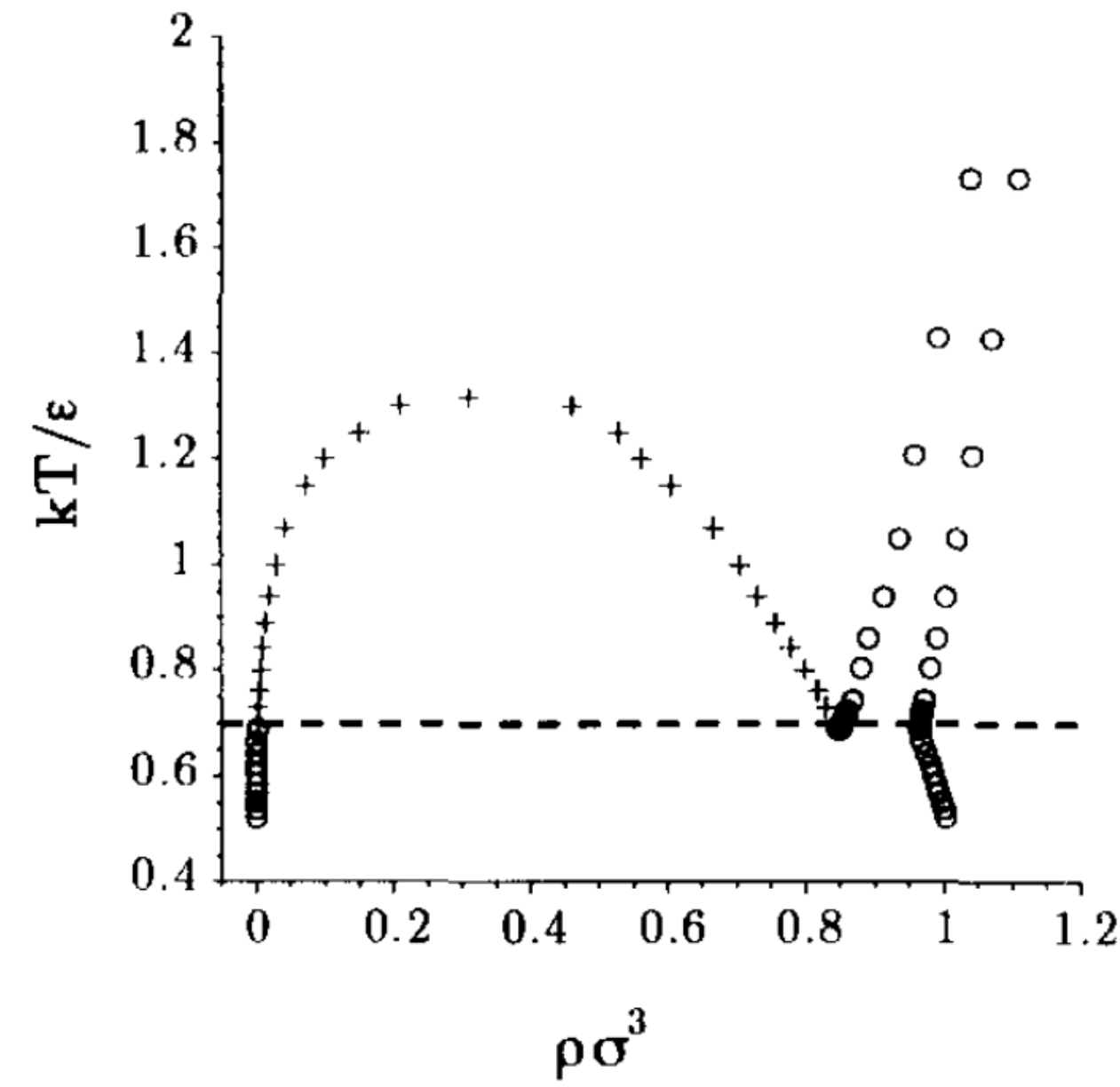
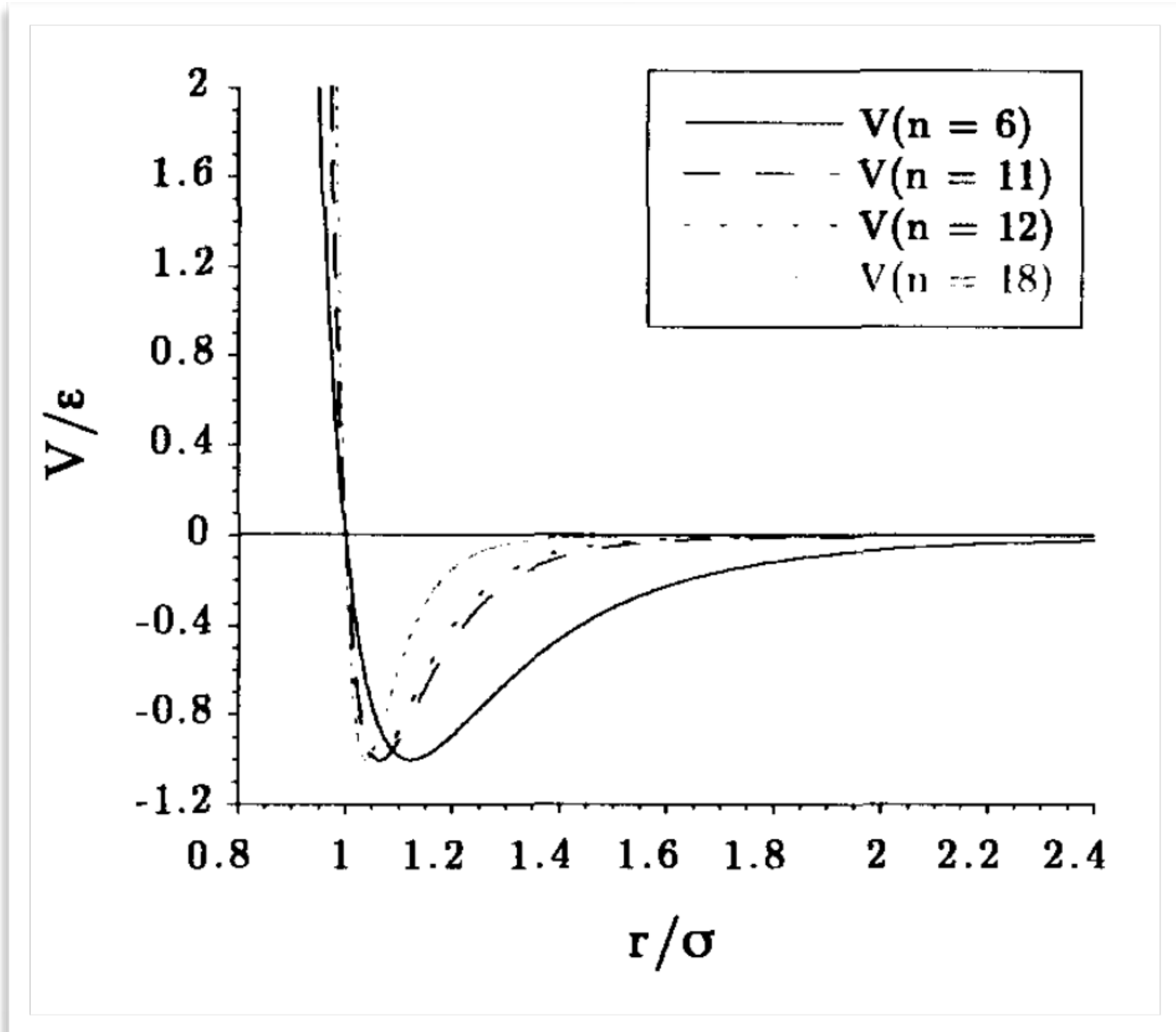


 = non-adsorbing polymer



Volume fraction,  $\phi$

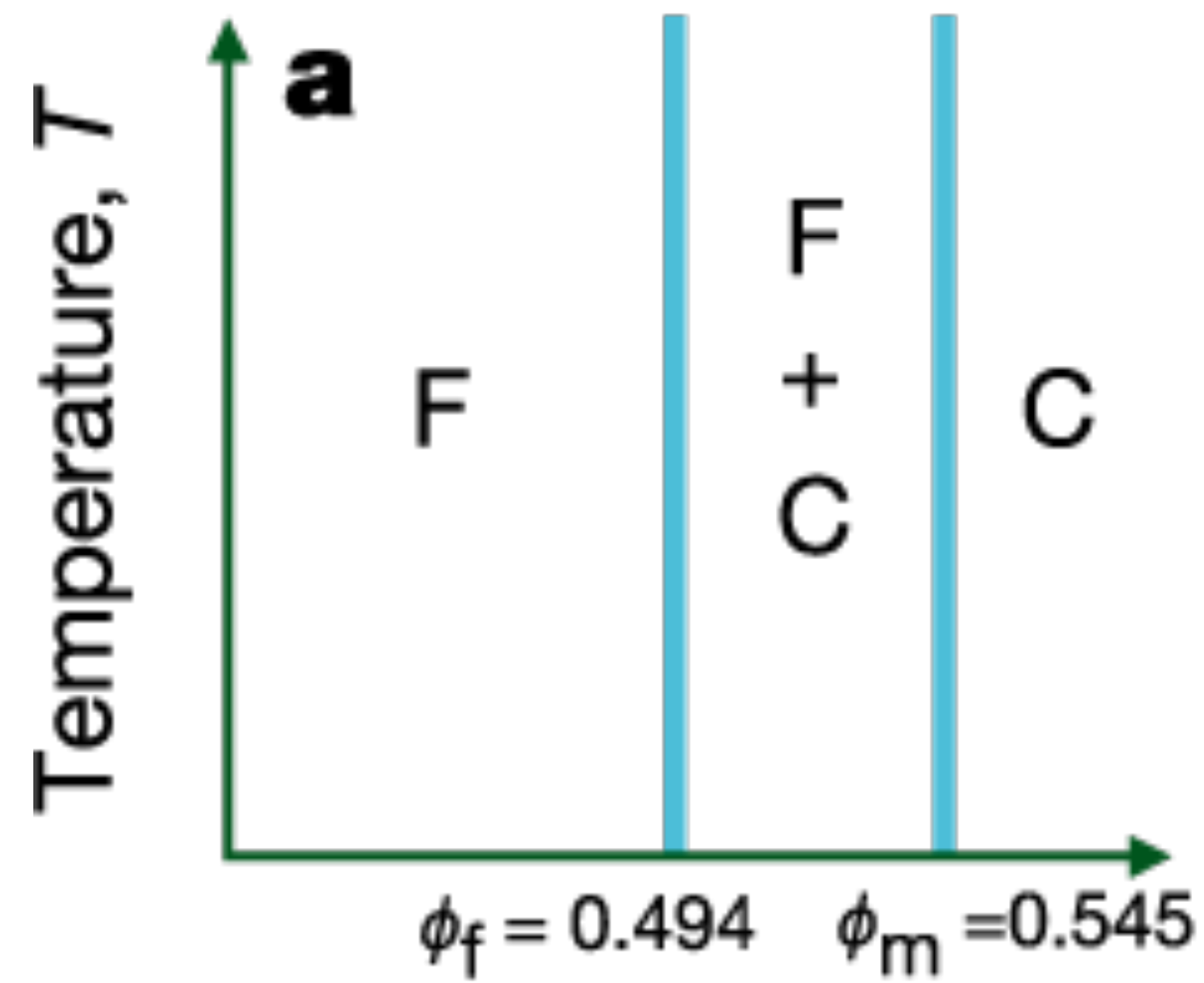
# Effect of attraction range on phase diagram



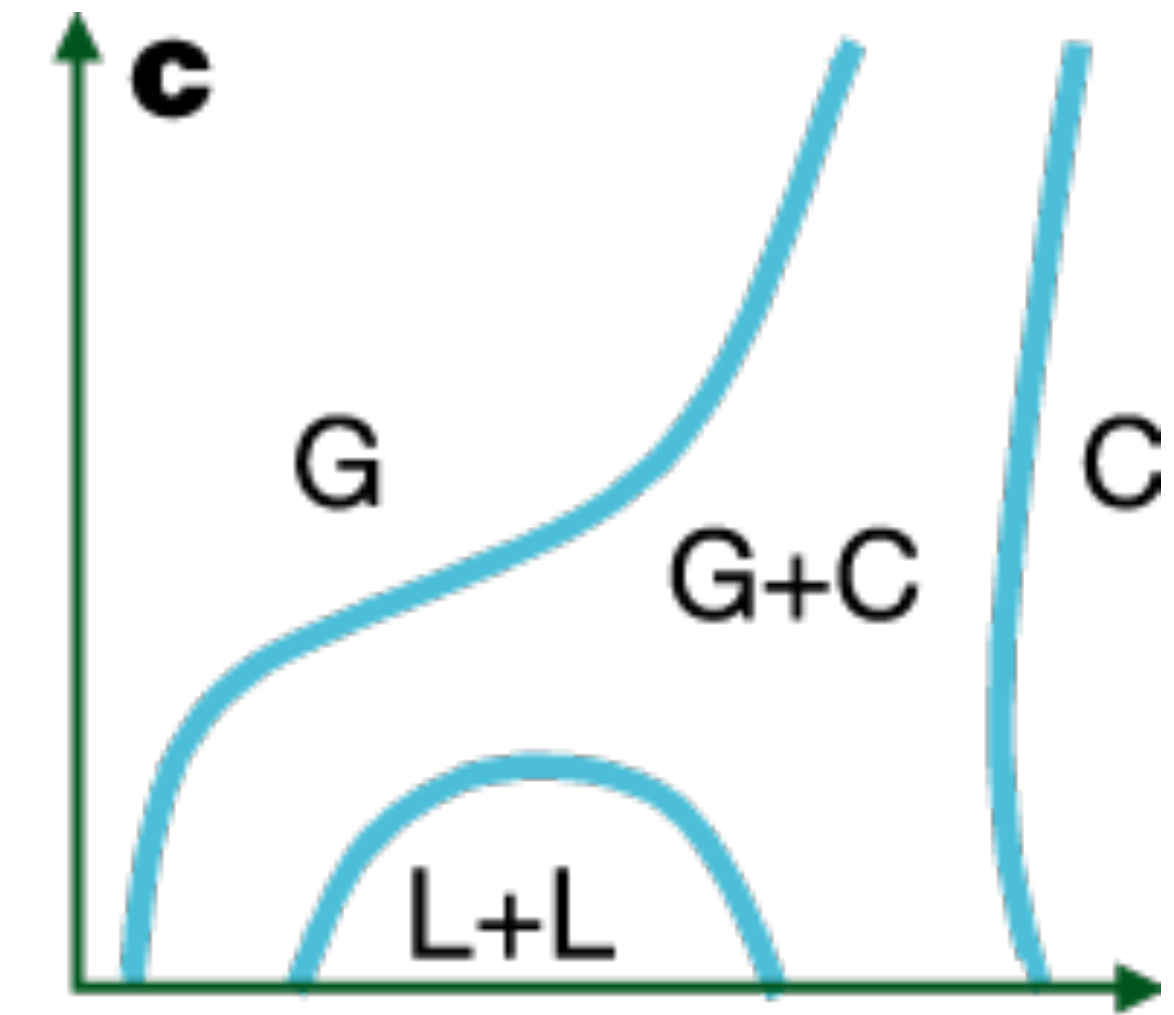
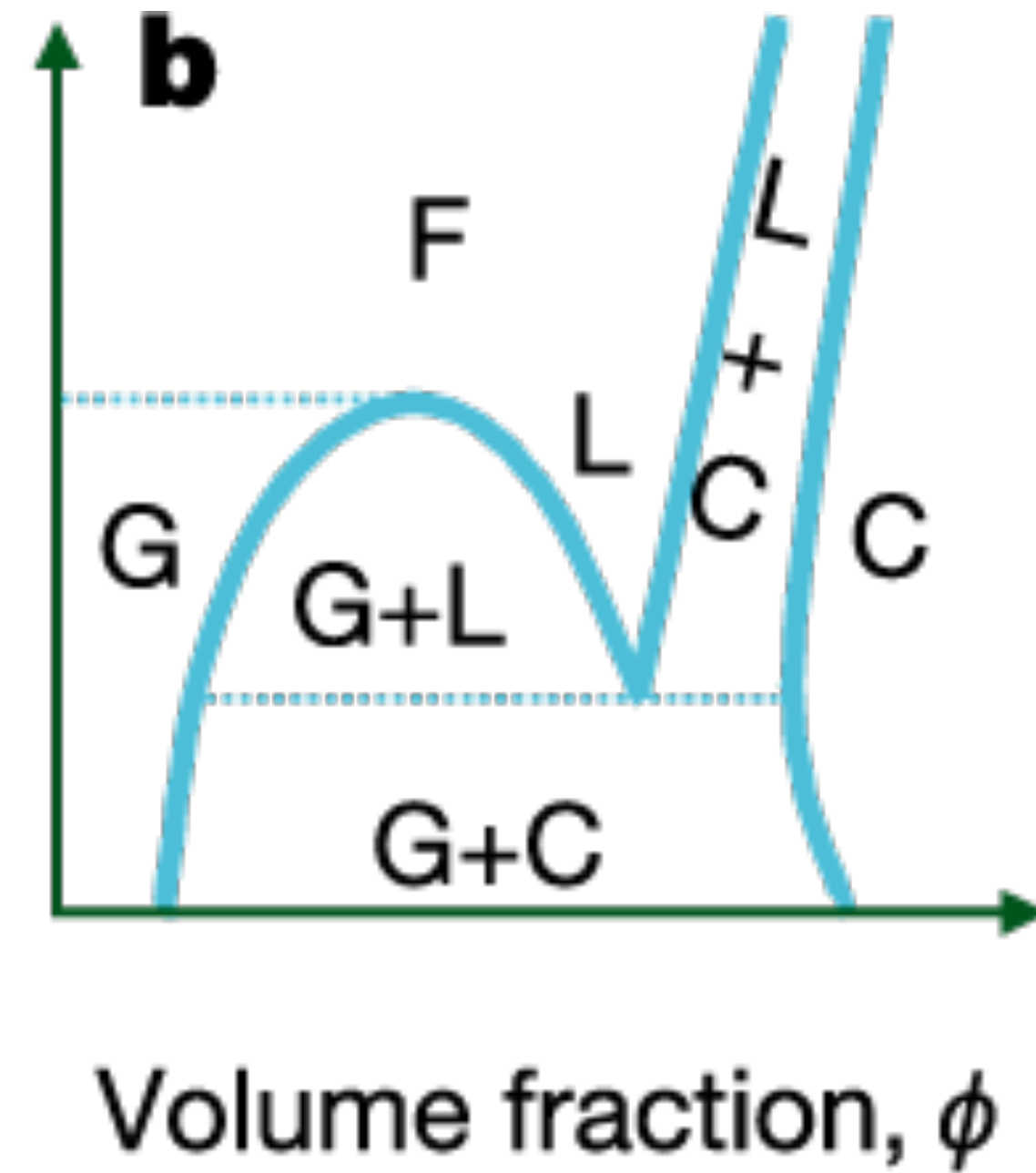
Vliegthart and Lekkerkerker  
 Physica A 1999

# Colloidal phase diagrams

hard spheres



short-range attractions



Anderson and Lekkerkerker  
Nature 2002

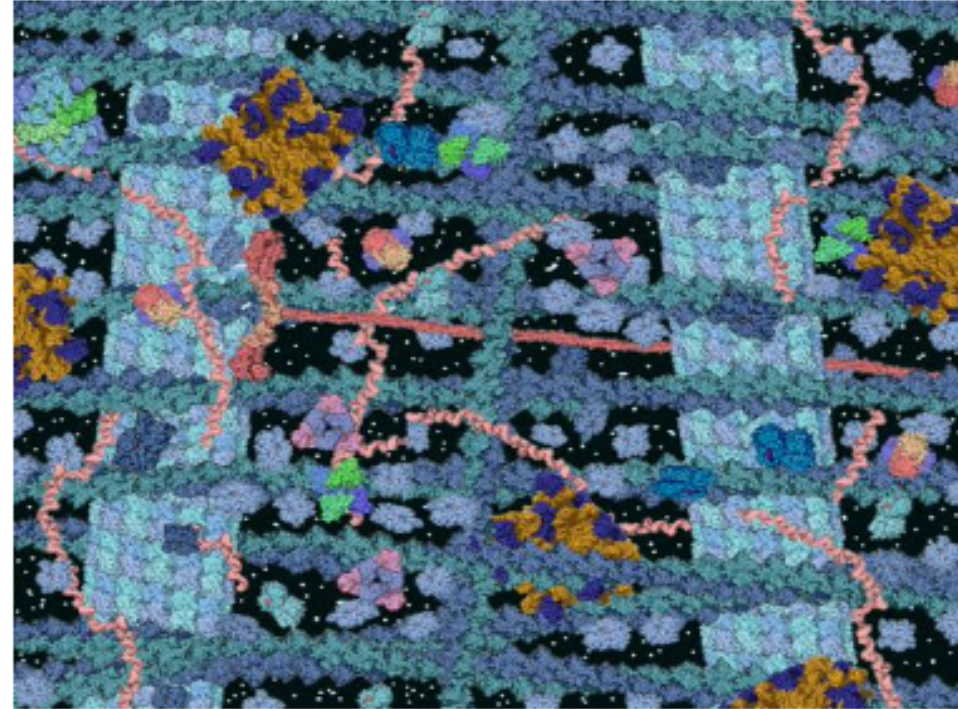
(atomic/molecular liquids)

colloids with long-range attraction

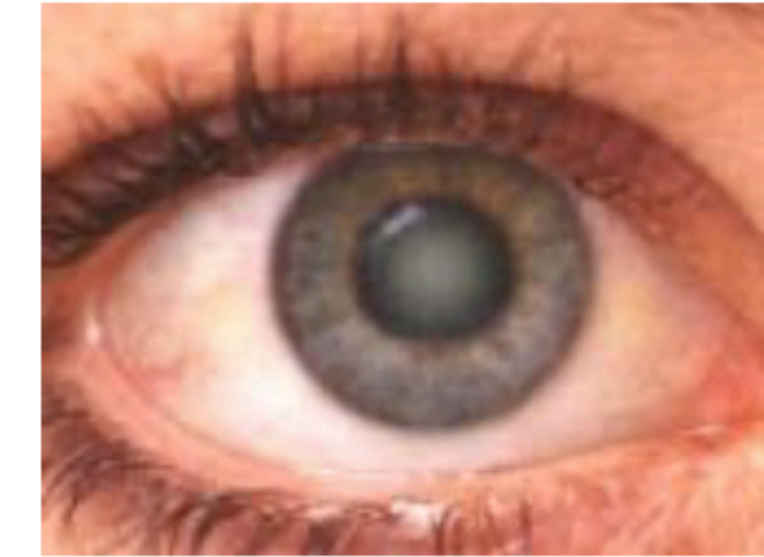


# Why are we interested in phase transitions in proteins?

Protein crowding and the cytosol stability



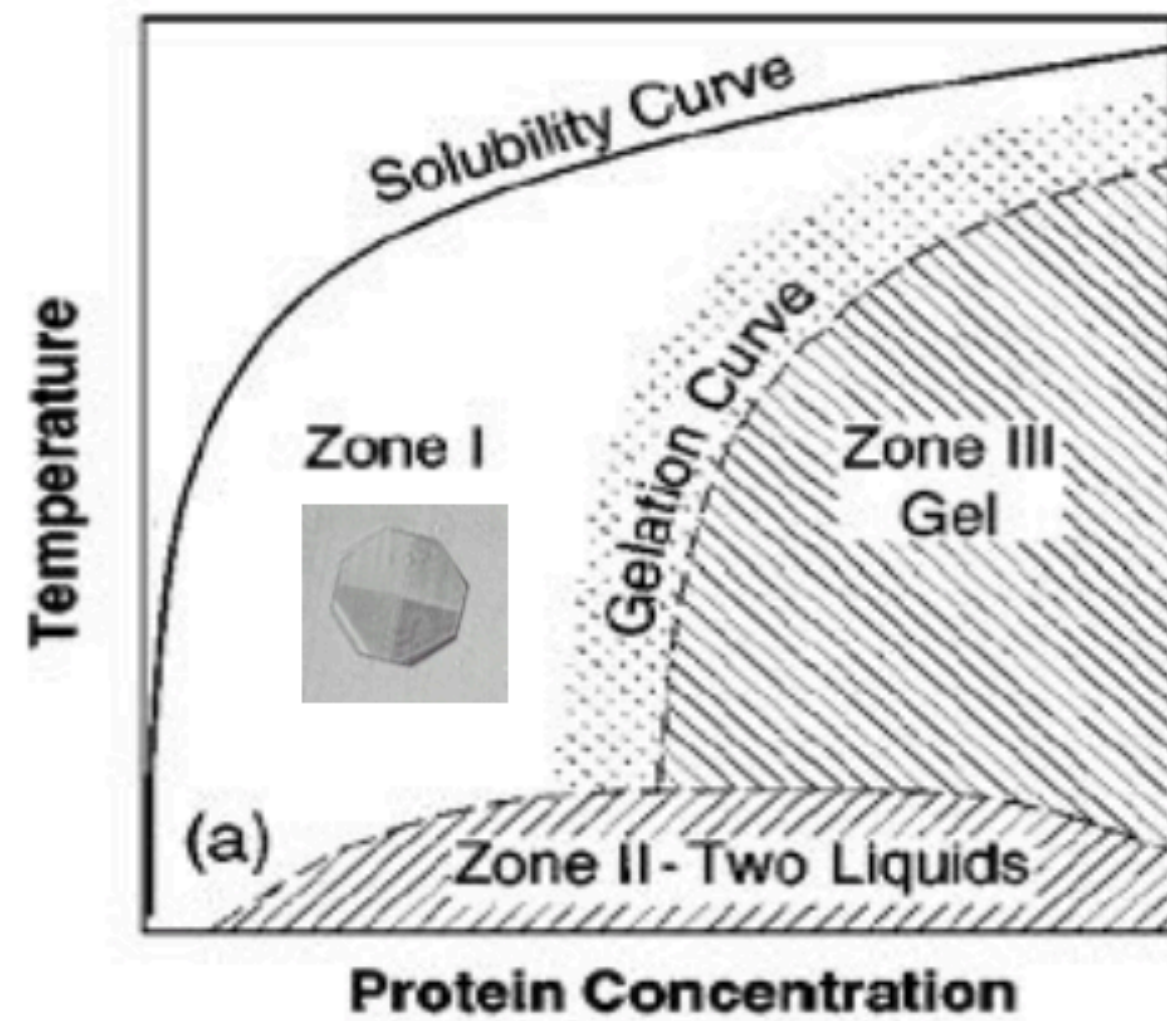
Understanding protein condensation disease



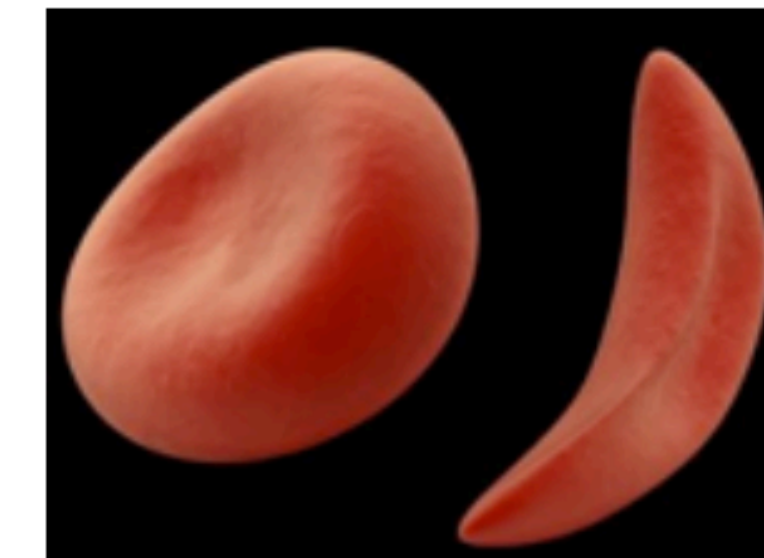
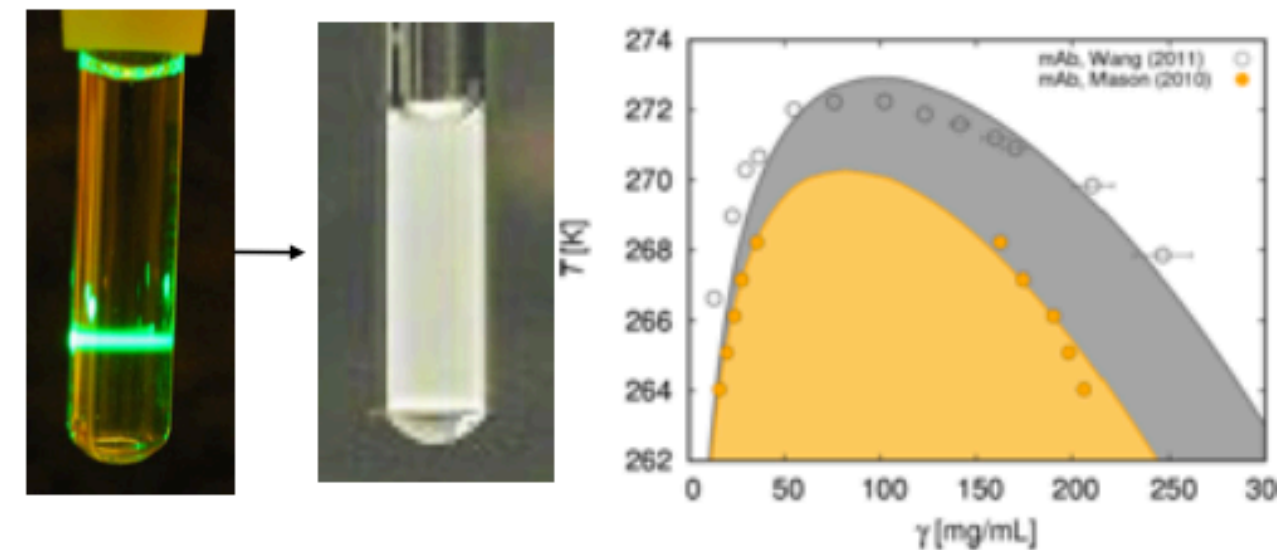
*Cataract*

Phase transitions in proteins

Optimal conditions for crystallization



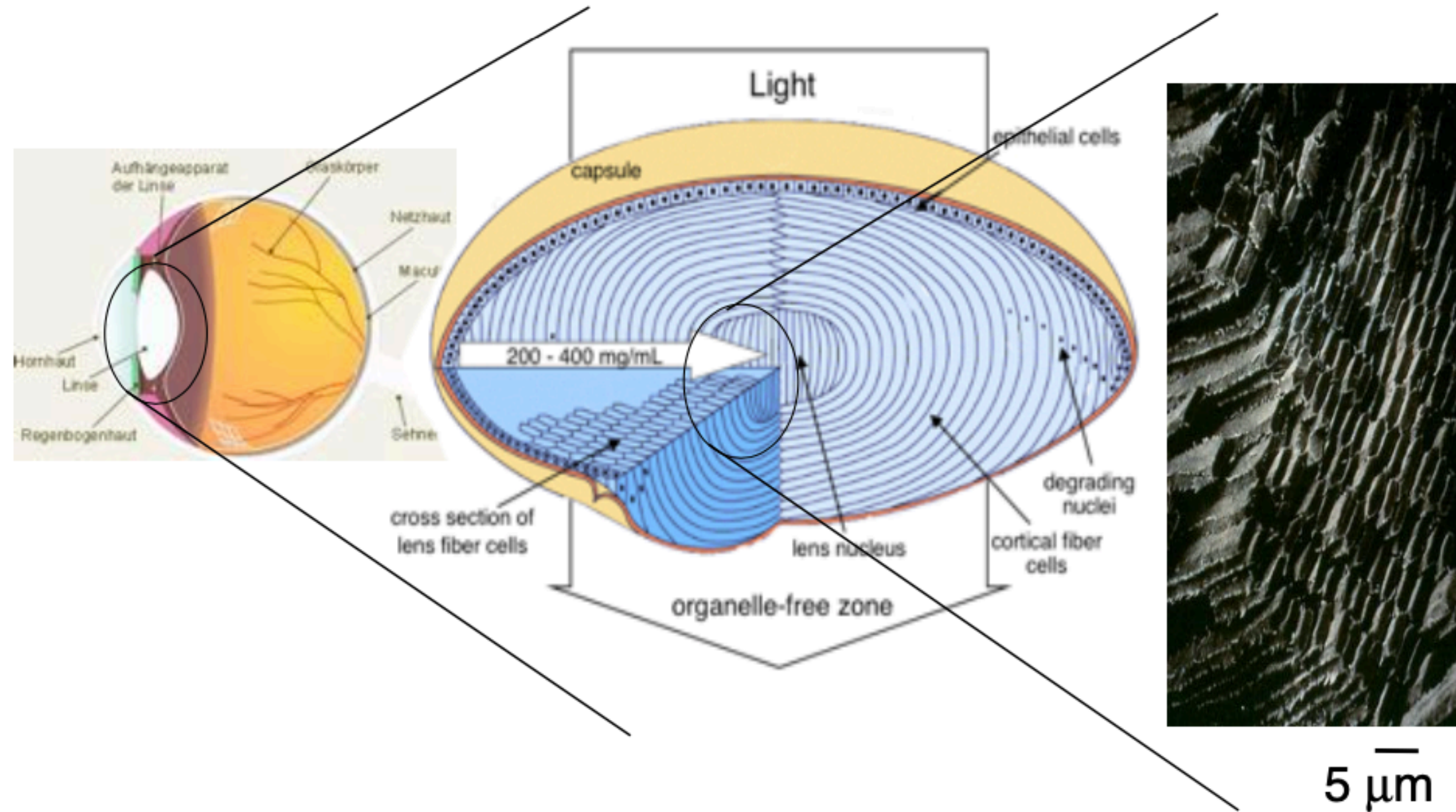
Concentrated formulations



*Sickle Cell Disease*

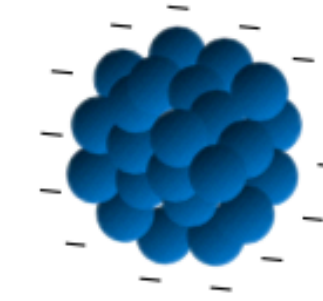
# The eye lens

The fiber cells consist of a highly concentrated protein solution:



## Alpha-crystallins:

~ 800 kDa  
specific volume:  
~ 1.5 - 1.7 mL/g



## Beta-crystallins:

$\beta_H$  ~ 200 kDa;  
 $\beta_L$  ~ 50 kDa

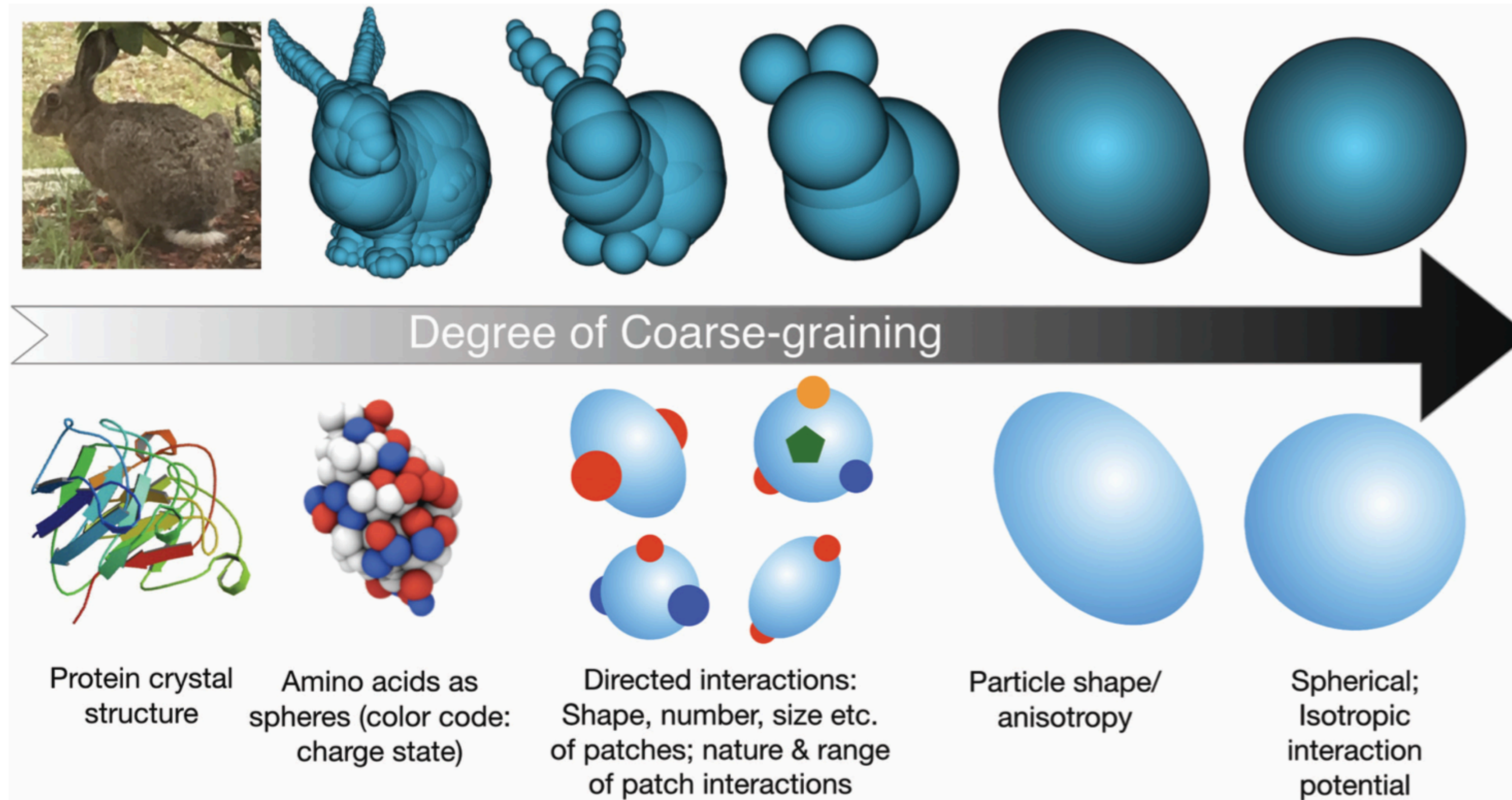


## Gamma-crystallins:

~ 20 kDa  
specific volume:  
~ 0.7 mL/g



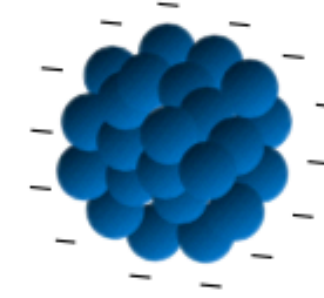
# Coarse-graining approach



# The eye lens

## Alpha-crystallins:

~ 800 kDa  
specific volume:  
~ 1.5 - 1.7 mL/g



## Beta-crystallins:

$\beta_H$  ~ 200 kDa;  
 $\beta_L$  ~ 50 kDa



## Gamma-crystallins:

~ 20 kDa  
specific volume:  
~ 0.7 mL/g



# The eye lens

Biophysical Journal

Article



## Crowding in the Eye Lens: Modeling the Multisubunit Protein $\beta$ -Crystallin with a Colloidal Approach

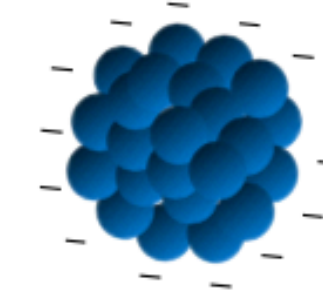
Felix Roosen-Runge,<sup>1,\*</sup> Alessandro Gulotta,<sup>1</sup> Saskia Bucciarelli,<sup>1</sup> Lucía Casal-Dujat,<sup>1</sup> Tommy Gating,<sup>1</sup> Nicholas Skar-Gislinge,<sup>1</sup> Marc Obiols-Rabasa,<sup>1</sup> Bela Farago,<sup>2</sup> Emanuela Zaccarelli,<sup>3,4</sup> Peter Schurtenberger,<sup>1</sup> and Anna Stradner<sup>1,\*</sup>

<sup>1</sup>Division of Physical Chemistry, Lund University, Lund, Sweden; <sup>2</sup>Institut Laue-Langevin, Grenoble, France; <sup>3</sup>Institute for Complex Systems, National Research Council, Uos Sapienza, Rome, Italy; and <sup>4</sup>Department of Physics, Sapienza Università di Roma, Rome, Italy

**ABSTRACT** We present a multiscale characterization of aqueous solutions of the bovine eye lens protein  $\beta_H$  crystallin from dilute conditions up to dynamical arrest, combining dynamic light scattering, small-angle x-ray scattering, tracer-based microrheology, and neutron spin echo spectroscopy. We obtain a comprehensive explanation of the observed experimental signatures from a model of polydisperse hard spheres with additional weak attraction. In particular, the model predictions quantitatively describe the multiscale dynamical results from microscopic nanometer cage diffusion over mesoscopic micrometer gradient diffusion up to macroscopic viscosity. Based on a comparative discussion with results from other crystallin proteins, we suggest an interesting common pathway for dynamical arrest in all crystallin proteins, with potential implications for the understanding of crowding effects in the eye lens.

### Alpha-crystallins:

~ 800 kDa  
specific volume:  
~ 1.5 - 1.7 mL/g



### Beta-crystallins:

$\beta_H$  ~ 200 kDa;  
 $\beta_L$  ~ 50 kDa



### Gamma-crystallins:

~ 20 kDa  
specific volume:  
~ 0.7 mL/g



Biophysical Journal 2020

# The eye lens

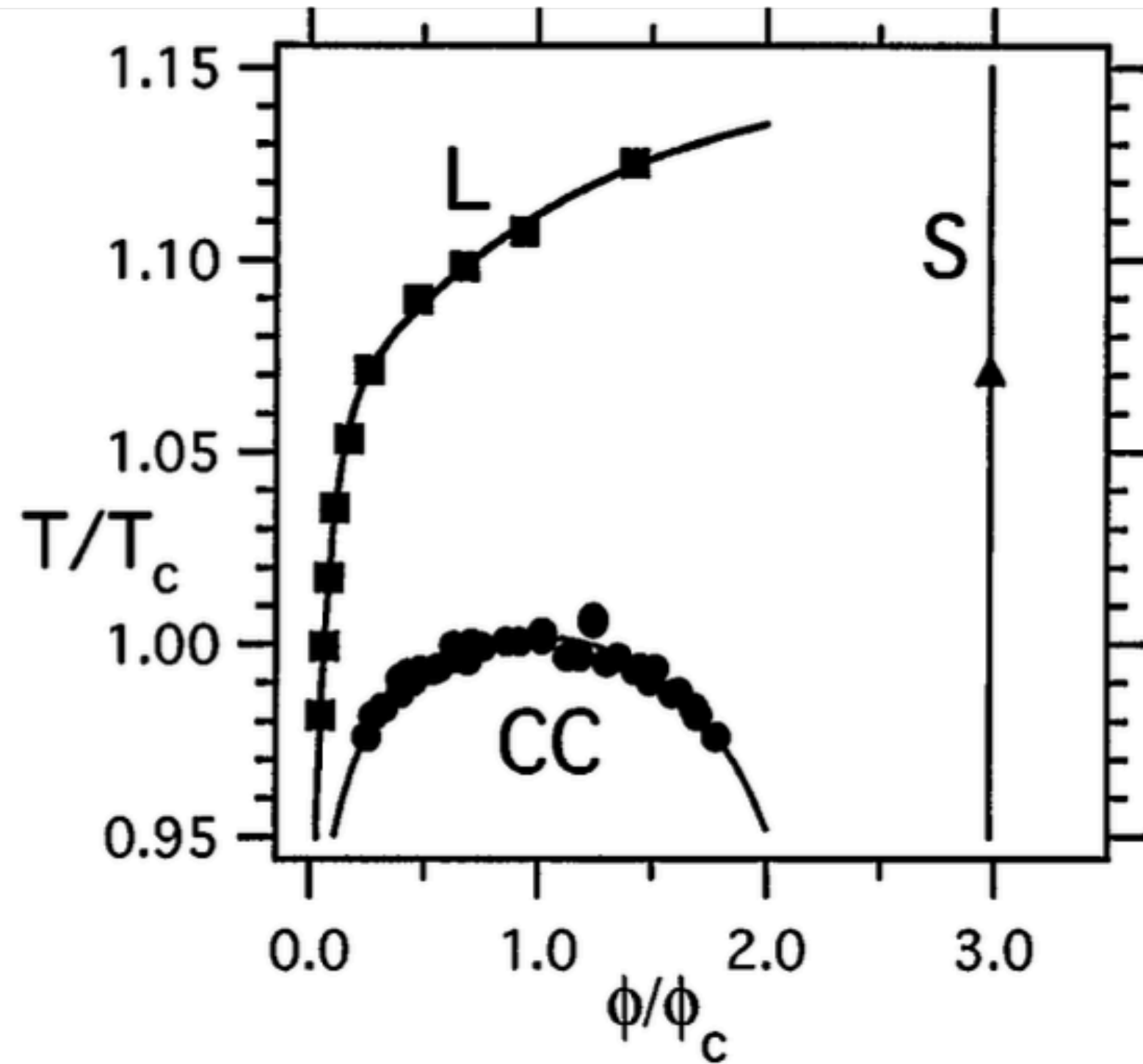
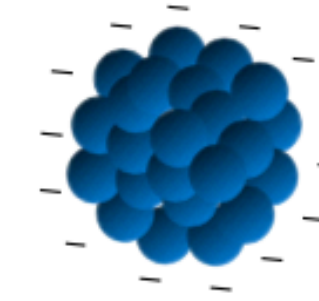


FIG. 1. The phase diagram of  $\gamma_{II}$ -crystallin [3–5]. The circles are points on the liquid-liquid coexistence curve (CC). The squares are points on the liquidus line (L). The triangle is a point on the solidus line (S). The lines are guides to the eye. The critical temperature is  $T_c = 278.4$  K. The critical volume fraction is  $\phi_c = 0.21$ .

Asherie, Lomakin & Benedek PRL 1996

## Alpha-crystallins:

$\sim 800$  kDa  
specific volume:  
 $\sim 1.5 - 1.7$  mL/g



## Beta-crystallins:

$\beta_H \sim 200$  kDa;  
 $\beta_L \sim 50$  kDa



## Gamma-crystallins:

$\sim 20$  kDa  
specific volume:  
 $\sim 0.7$  mL/g



# Colloidal approach to globular protein phase diagram: isotropic potentials

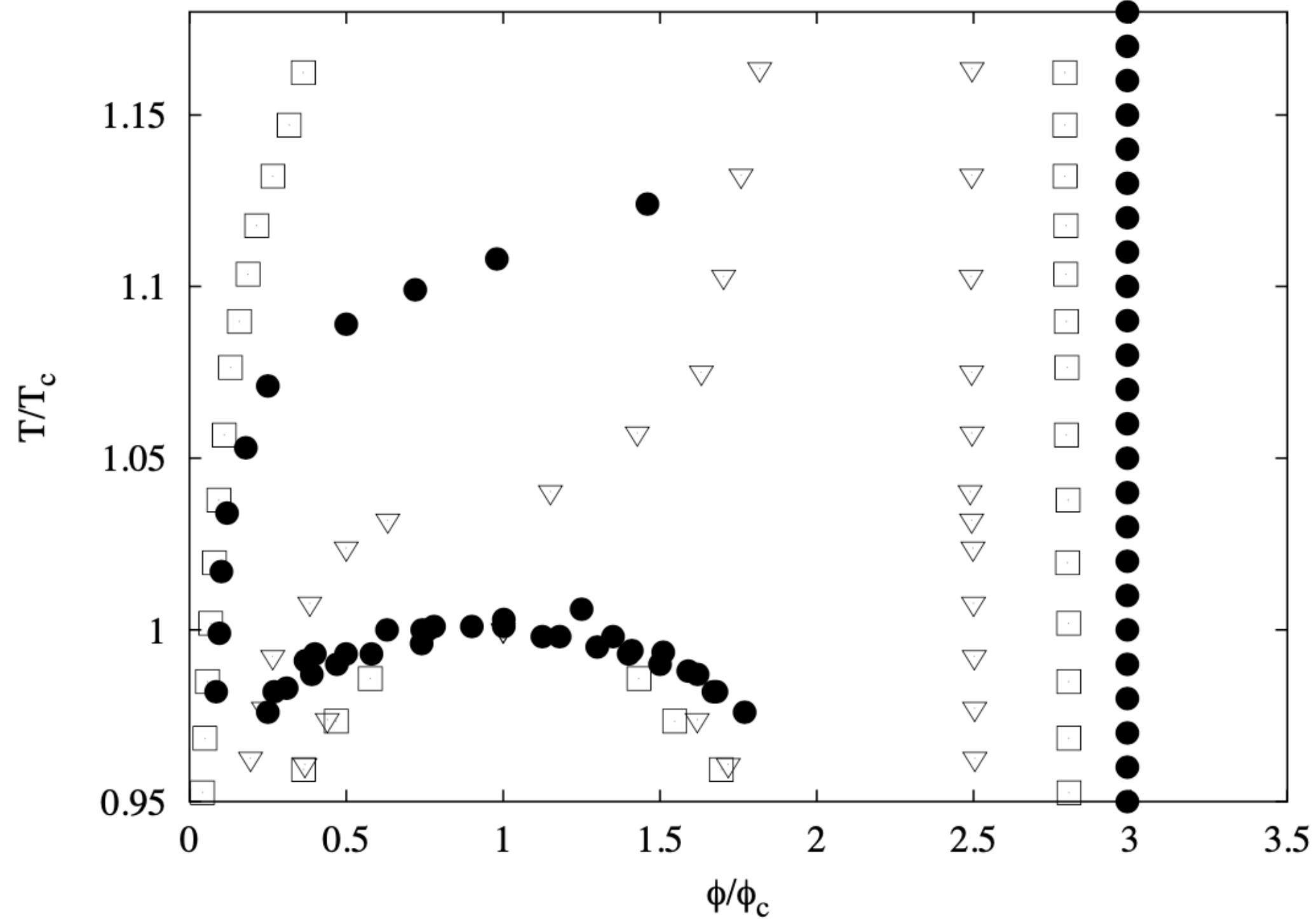
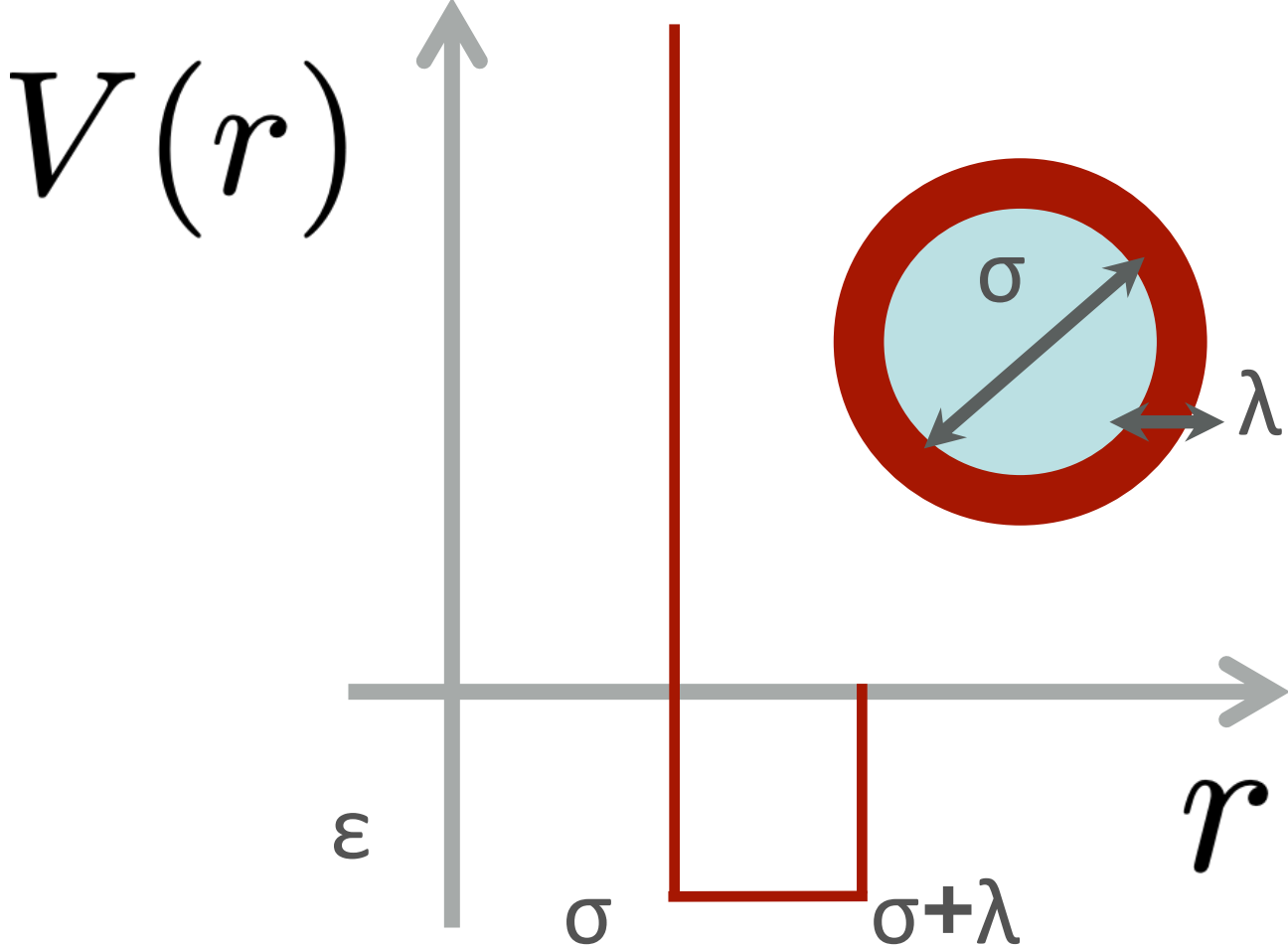


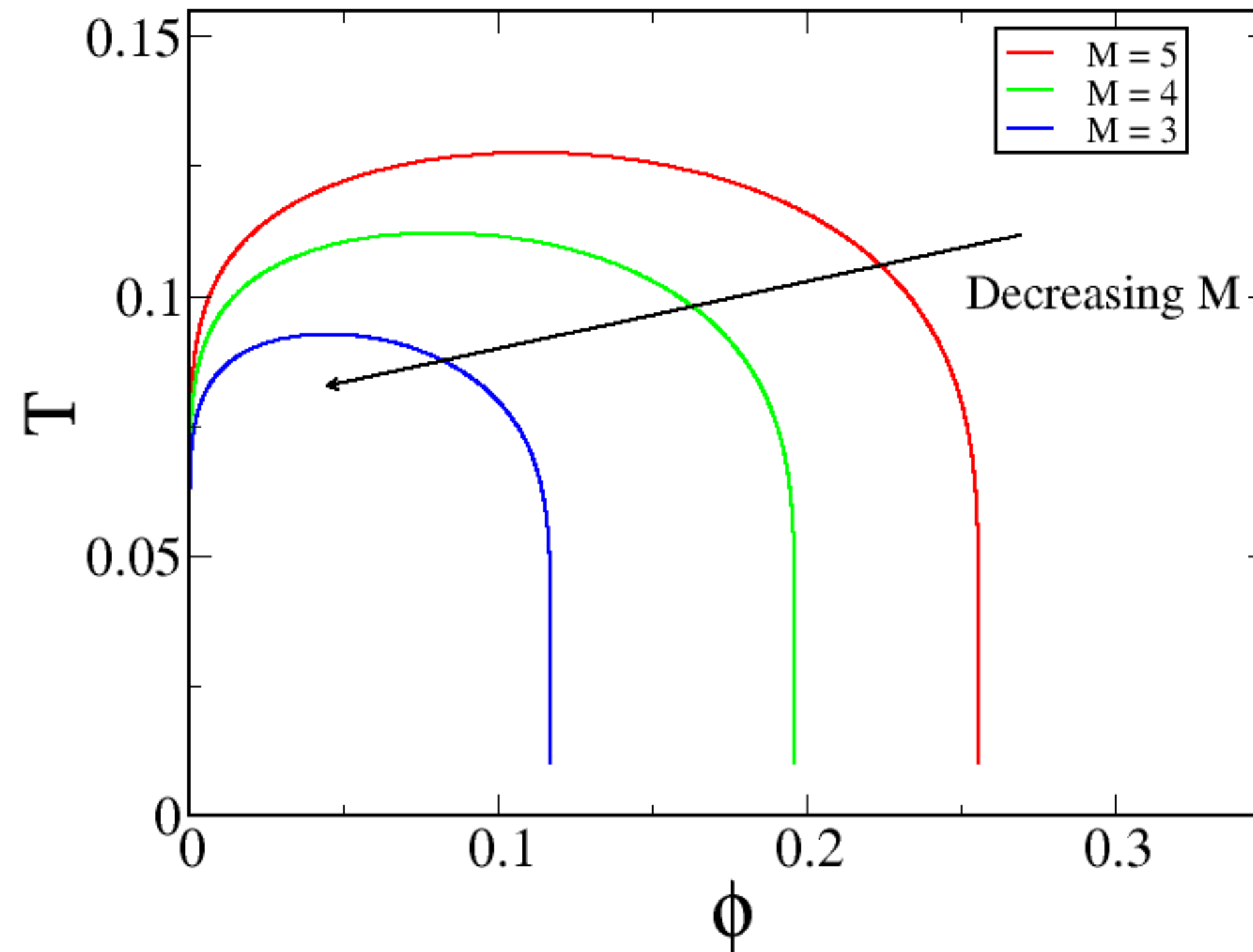
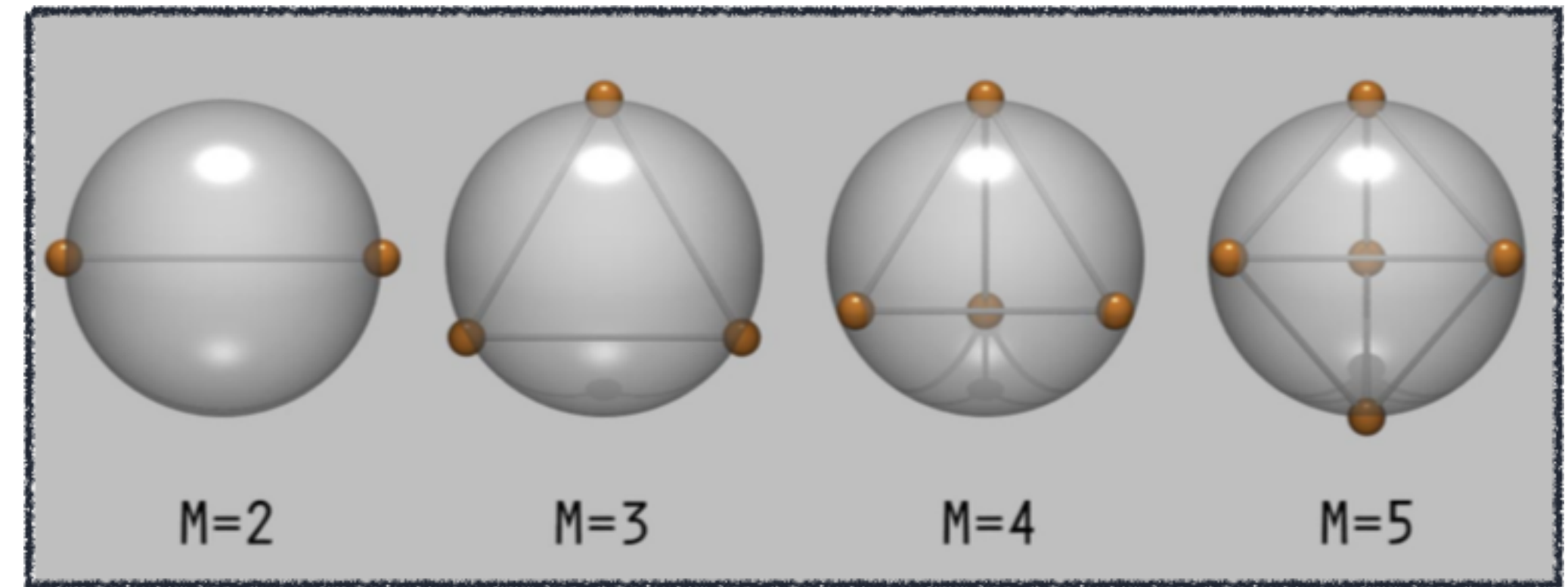
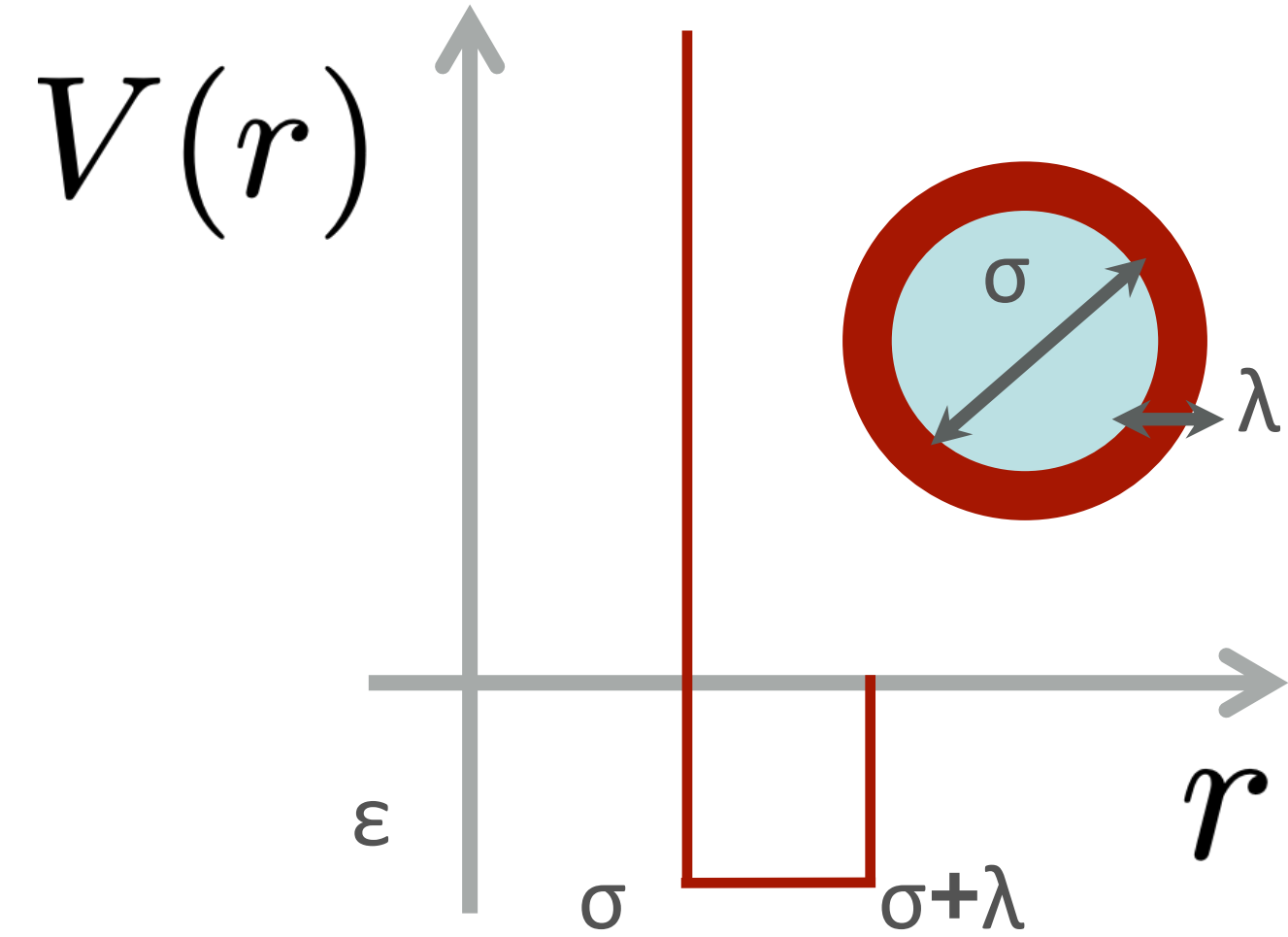
FIG. 8: Comparison of our Monte Carlo results for both  $\lambda = 1.15$  ( $\square$ ) and  $\lambda = 1.25$  ( $\nabla$ ), respectively, to the gamma-II crystallin ( $\bullet$ ).

Pagan & Gunton J. Chem. Phys. 2005



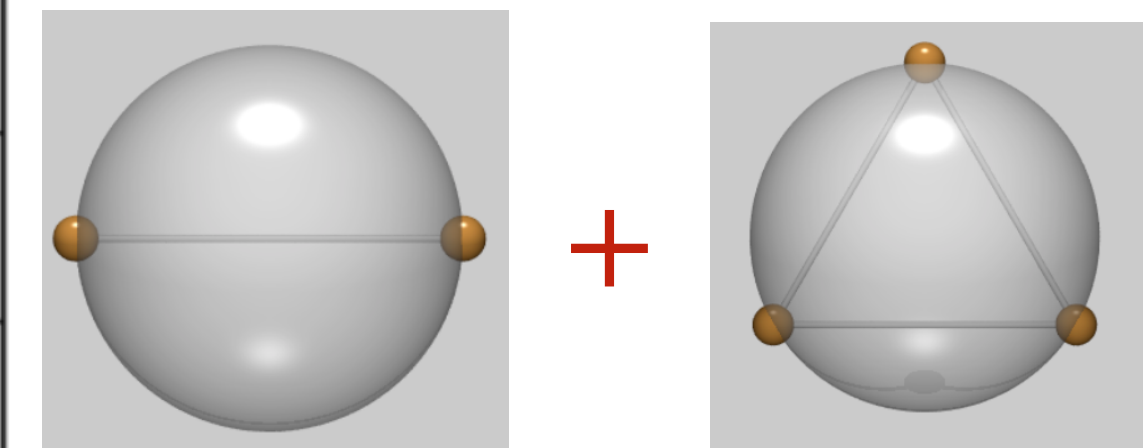
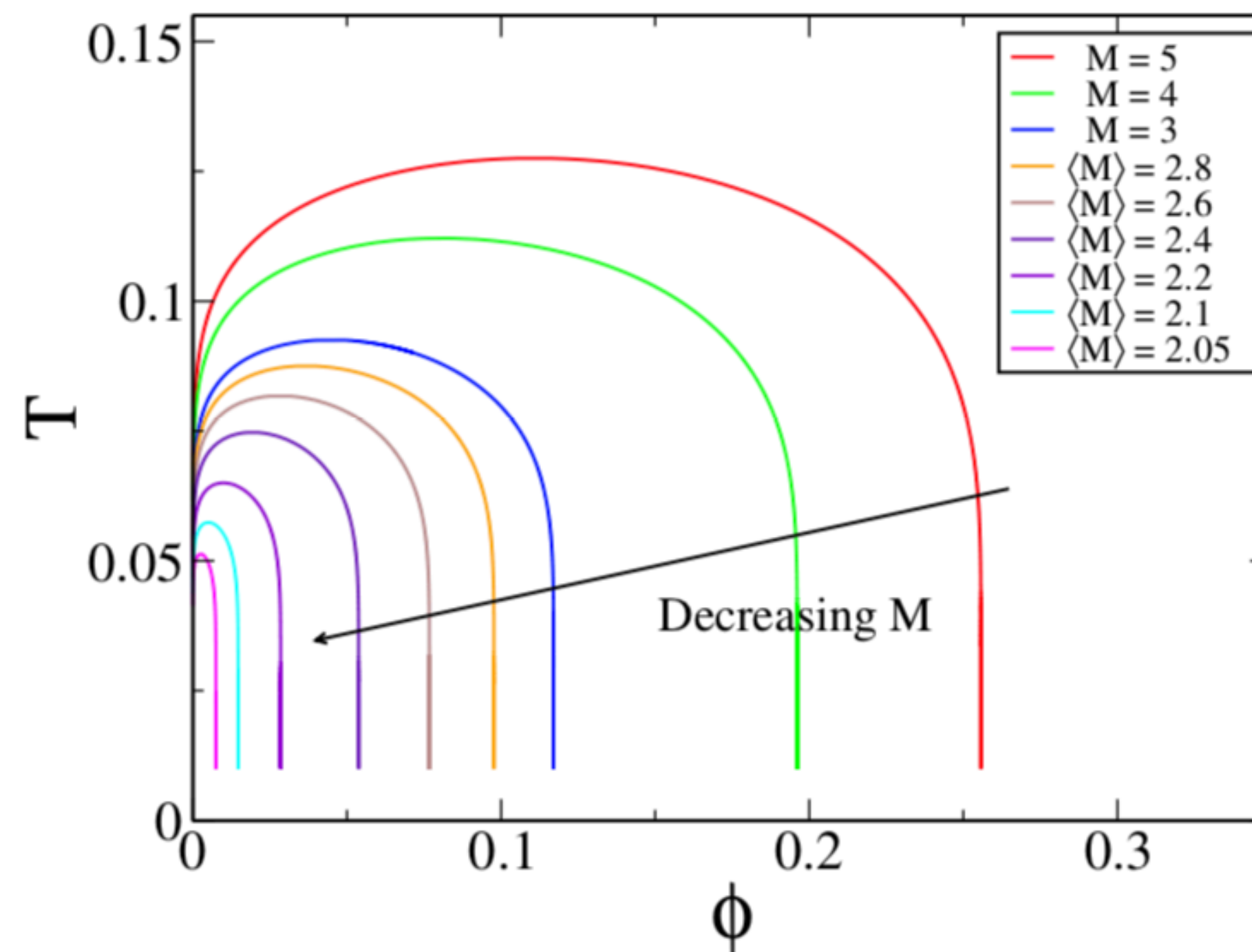
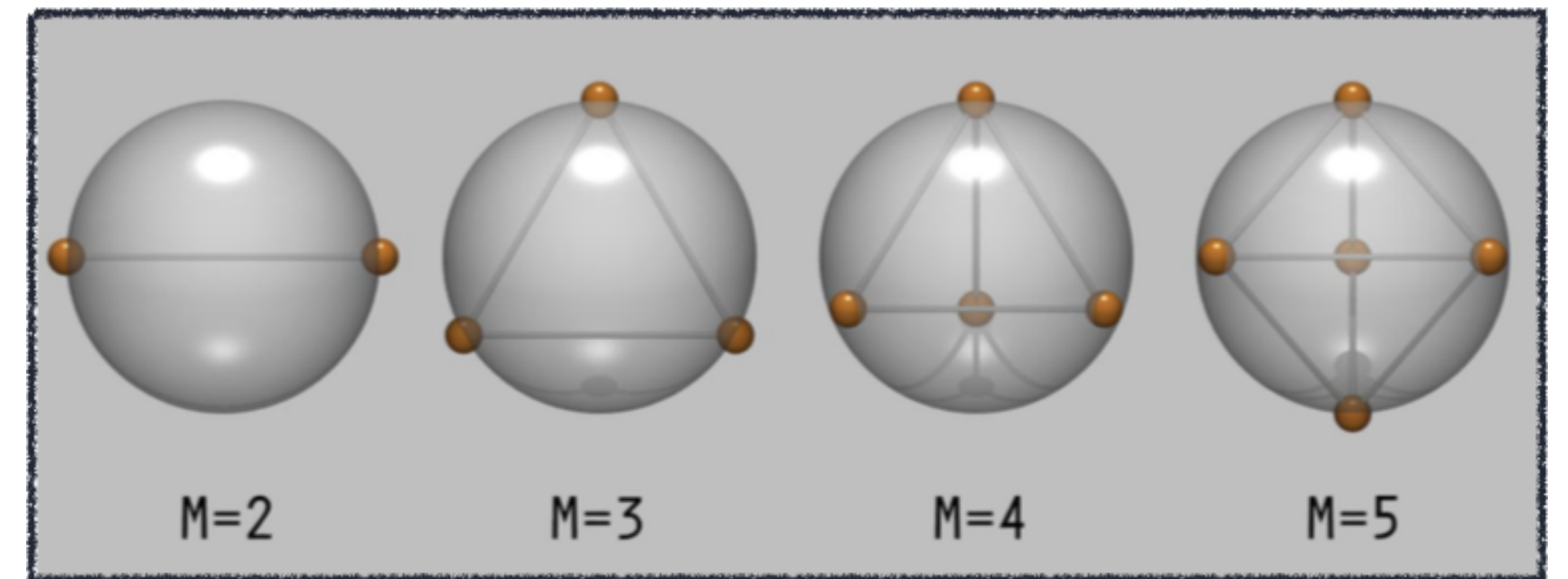
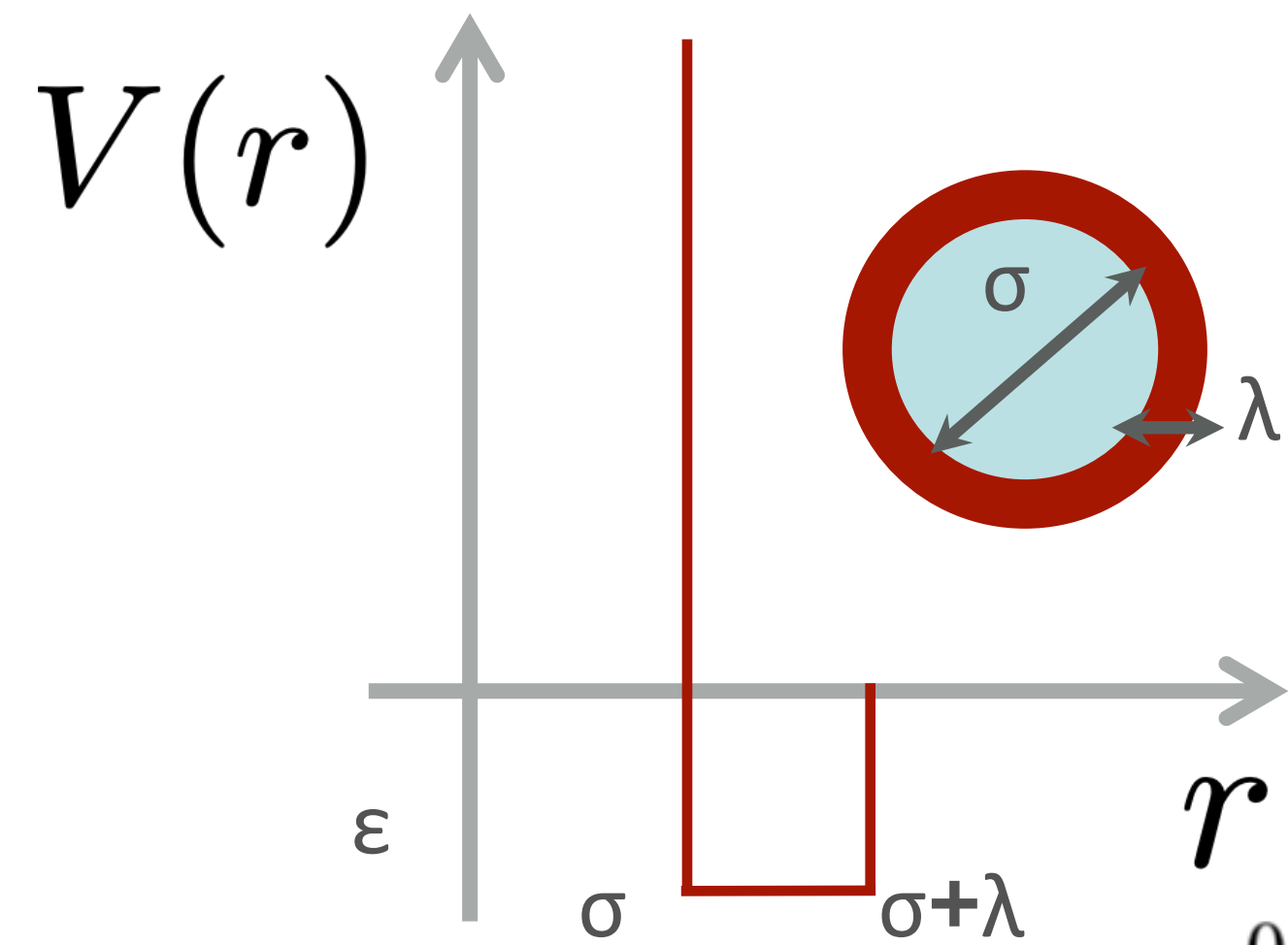
square-well potential

# The need of anisotropic models: patchy models

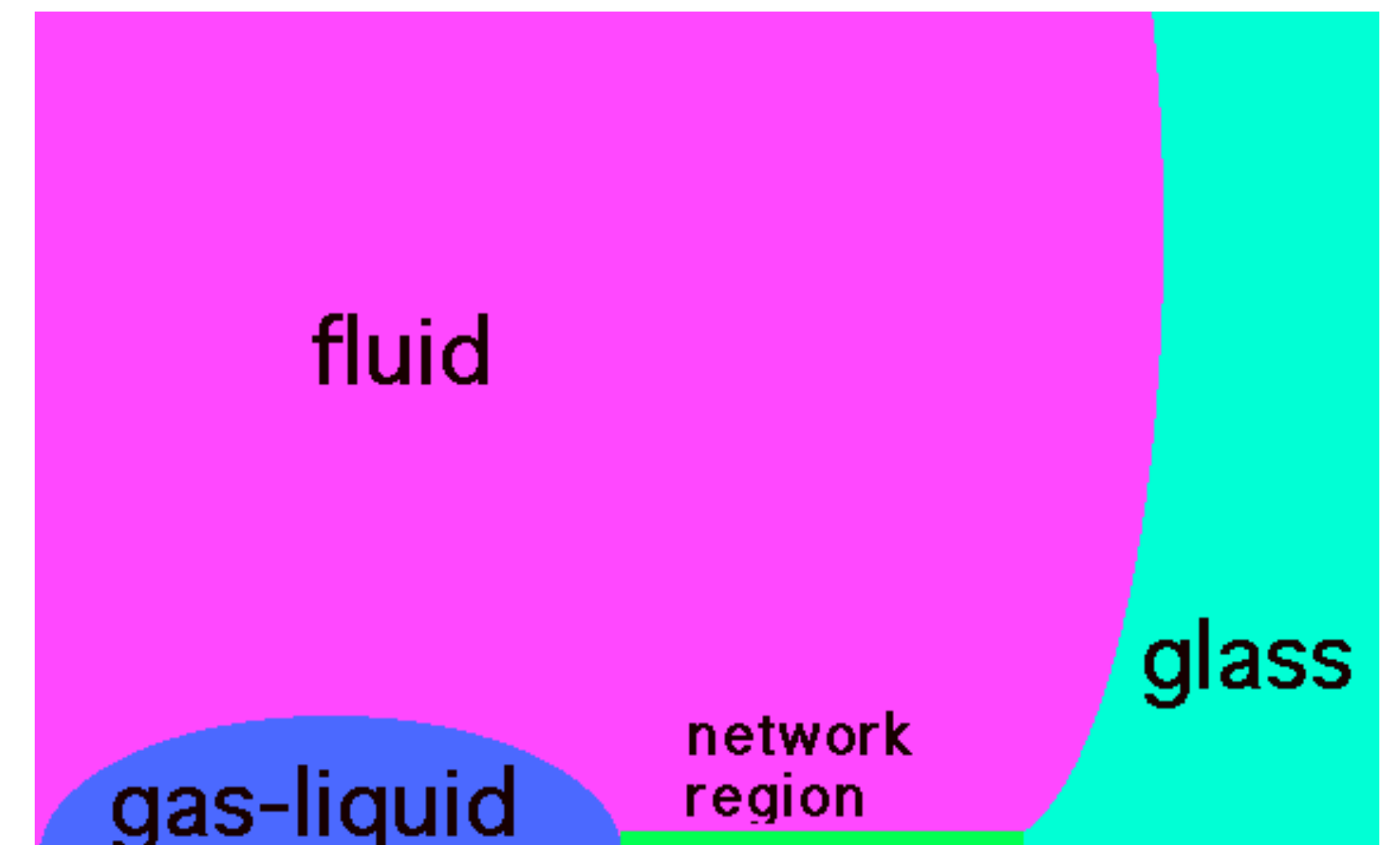
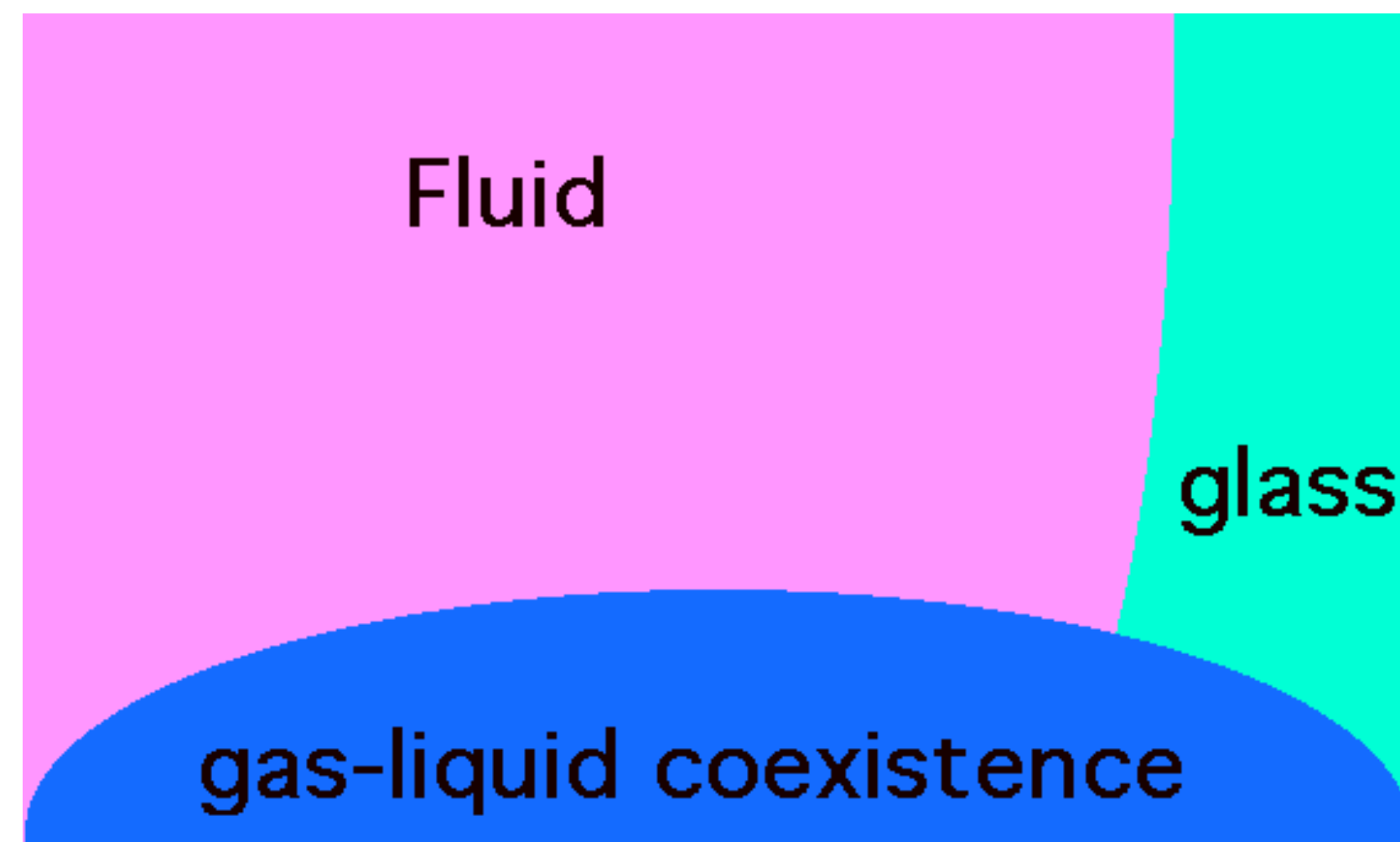
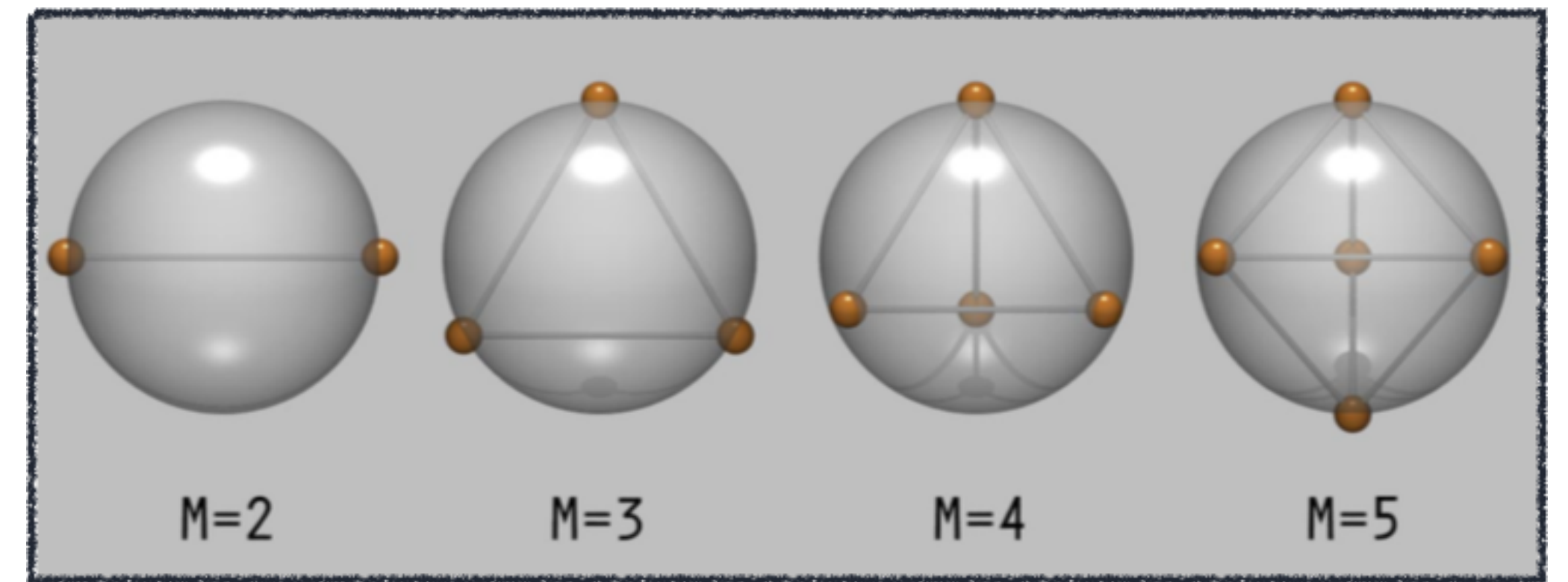
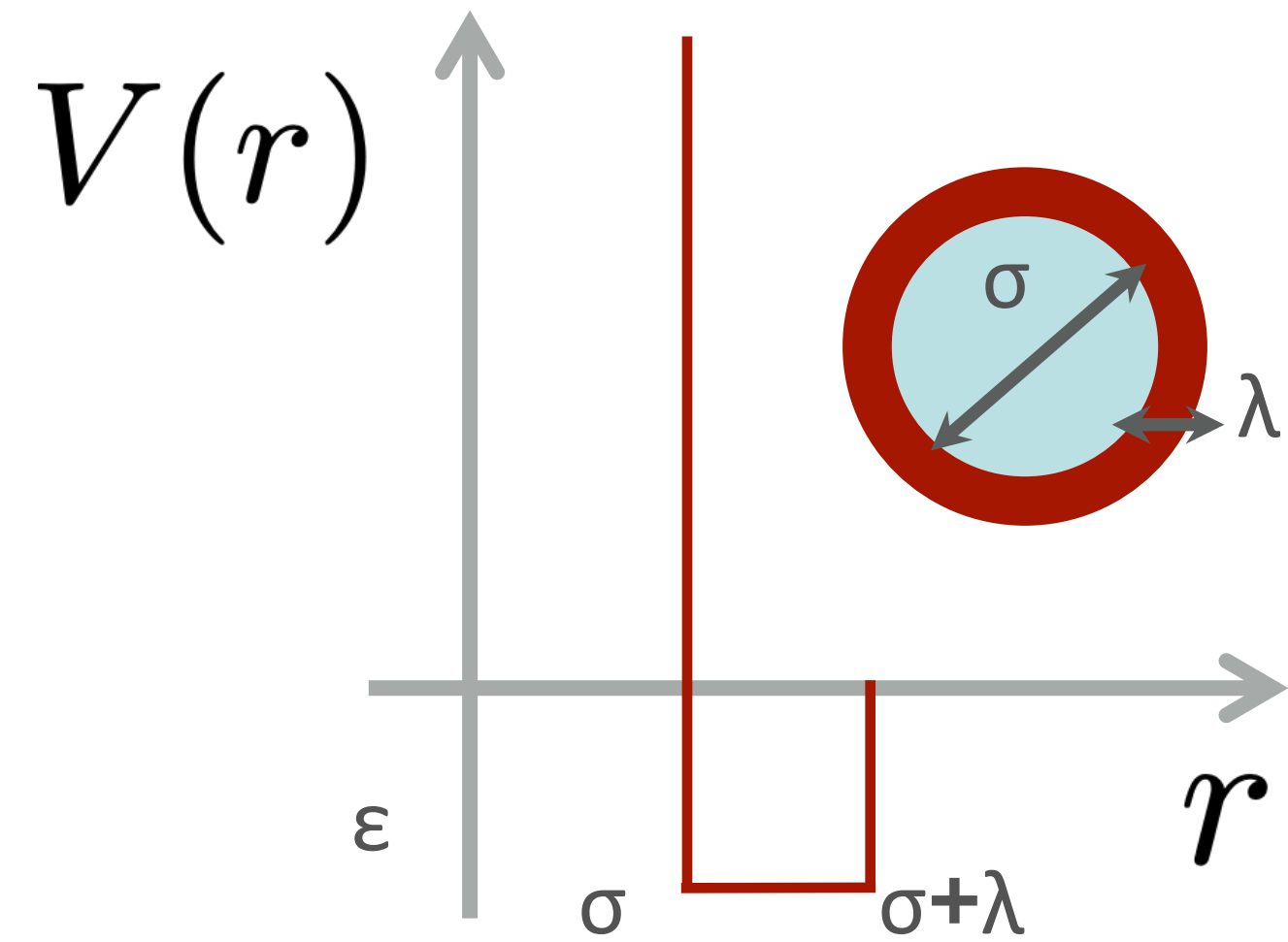




# The need of anisotropic models: patchy models



# The need of anisotropic models: patchy models



# Back to proteins

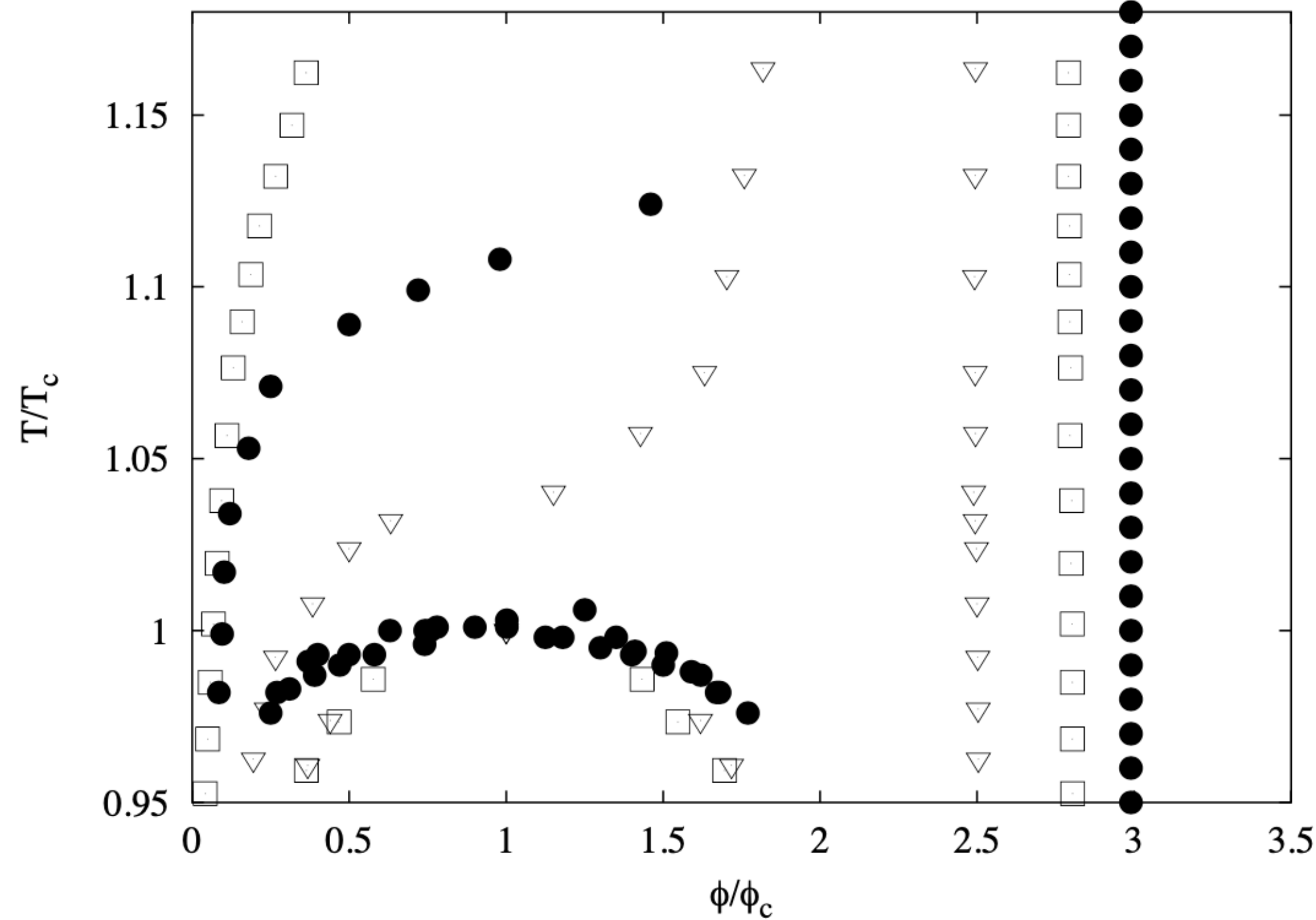


FIG. 8: Comparison of our Monte Carlo results for both  $\lambda = 1.15$  ( $\square$ ) and  $\lambda = 1.25$  ( $\nabla$ ), respectively, to the gamma-II crystallin ( $\bullet$ ).

Pagan & Gunton J. Chem. Phys. 2005

square-well potential

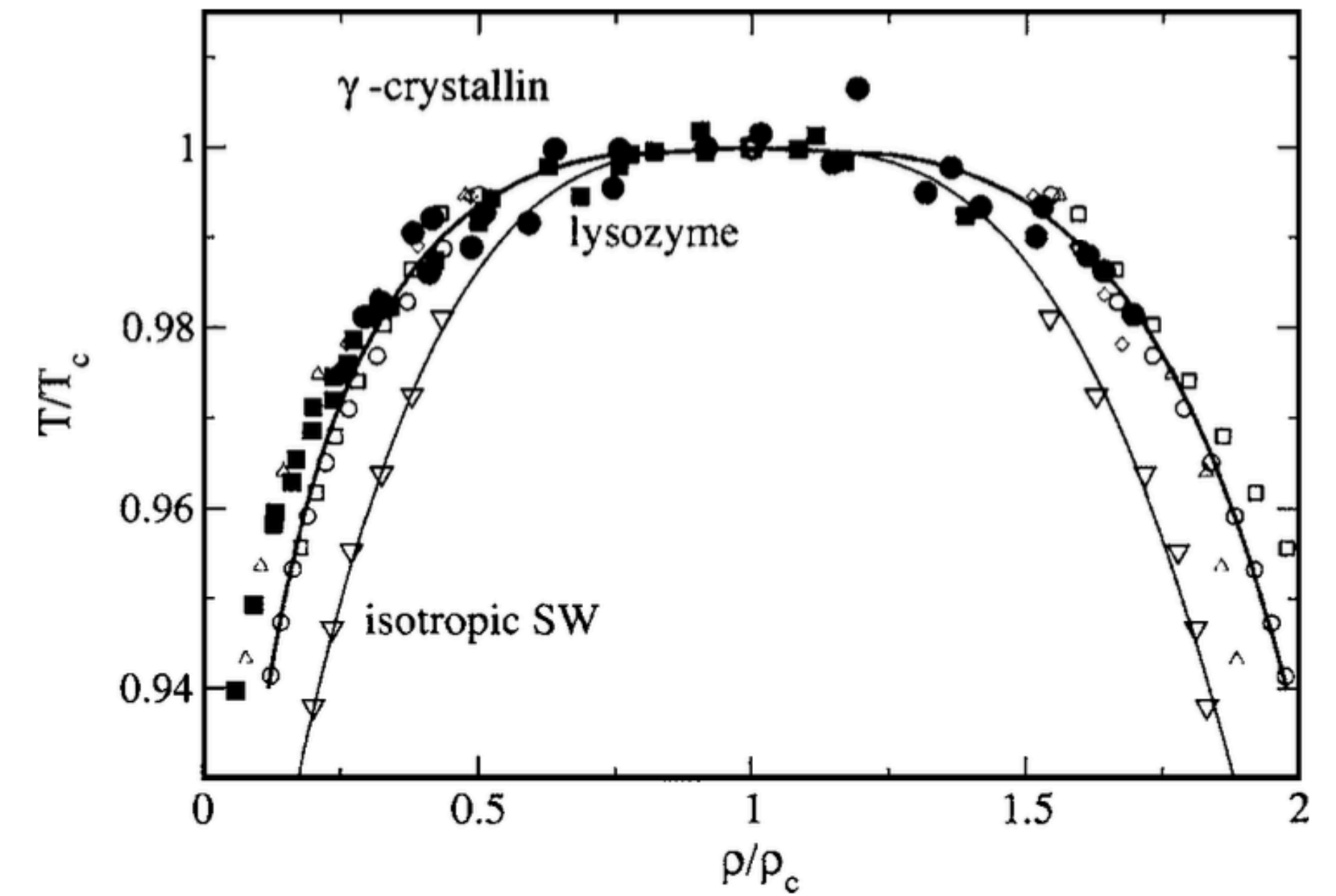


FIG. 4. Vapor-liquid coexistence curves in terms of the reduced temperatures  $T/T_c$  and the reduced density  $\rho/\rho_c$  for the studied patchy models:  $M=4$  (open squares),  $M=5$  (open circles), and  $M=7$  (open diamonds). The isotropic only (Ref. 27) (inverted open triangles) and PMW (Ref. 16) (open triangles) data are shown for comparison. The experimental data for  $\gamma$ -crystallin (full circles) and lysozyme (full squares) are taken from Refs. 28 and 29, respectively. The lines are the fit to the standard critical scaling law used to describe coexistence curves.

Liu, Kumar & Sciortino J. Chem. Phys. 2007

square-well + patchy

# Single-point modification of protein interactions

## How fluorescent labelling alters the solution behaviour of proteins

M. K. Quinn,<sup>a</sup> N. Gnan,<sup>b</sup> S. James,<sup>a</sup> A. Ninarello,<sup>c</sup> F. Sciortino,<sup>bc</sup> E. Zaccarelli<sup>bc</sup> and J. J. McManus<sup>\*a</sup>

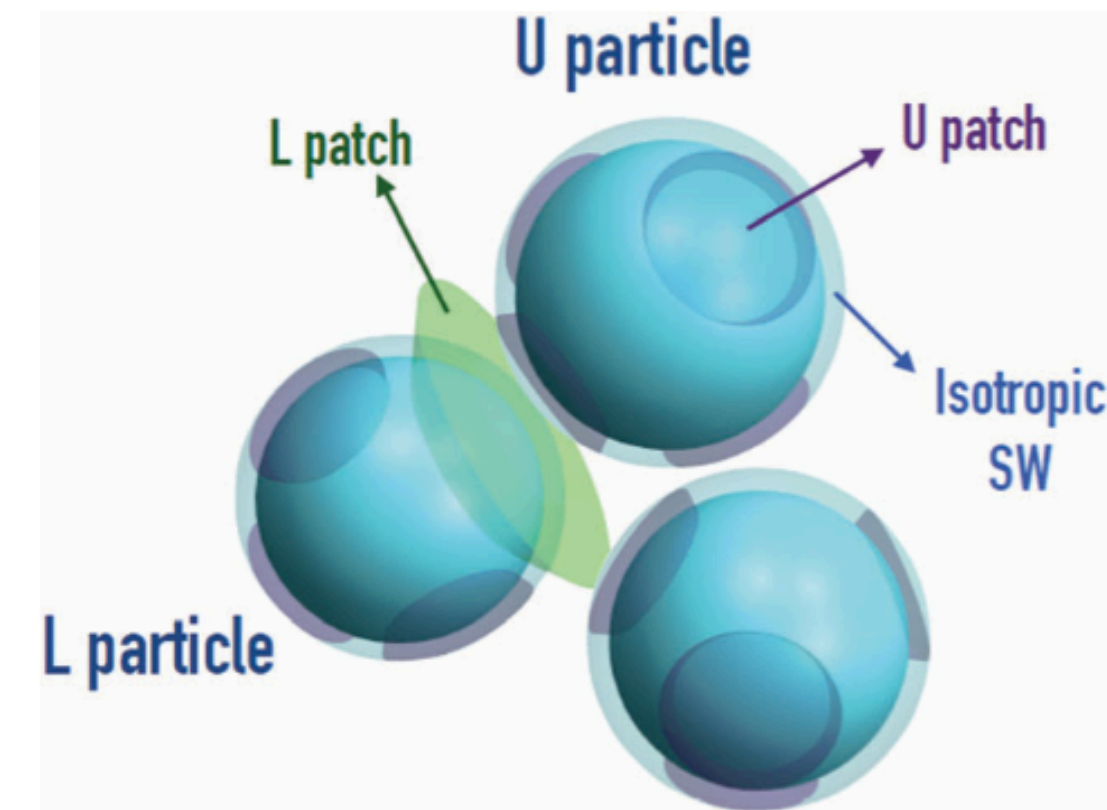
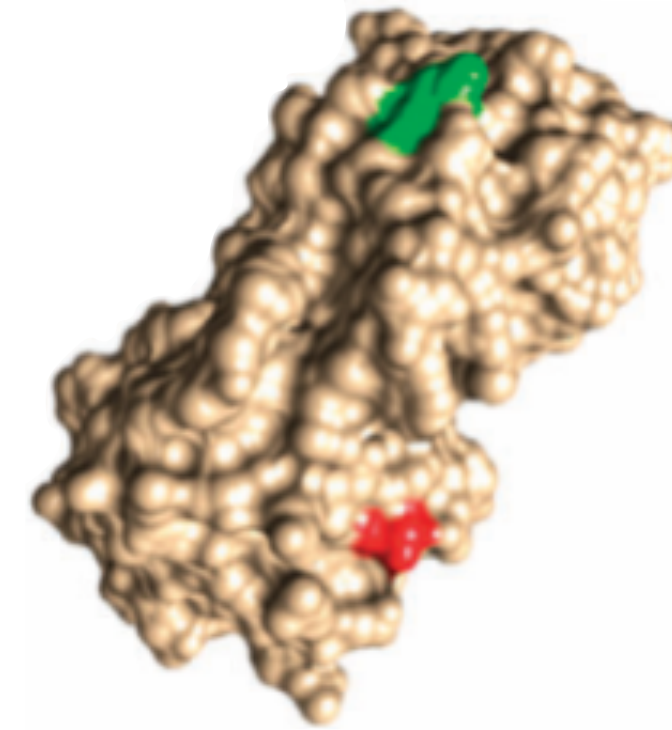
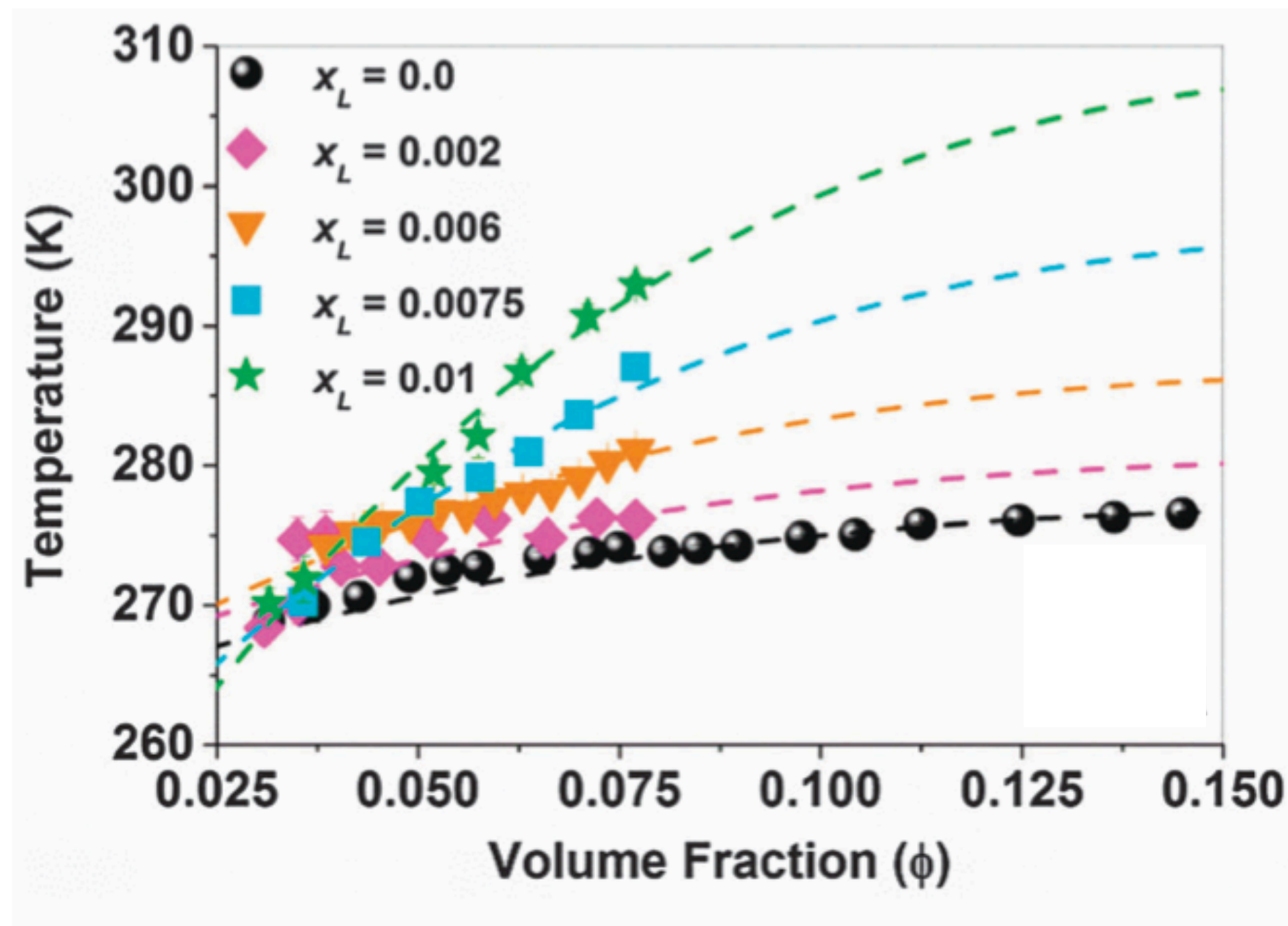
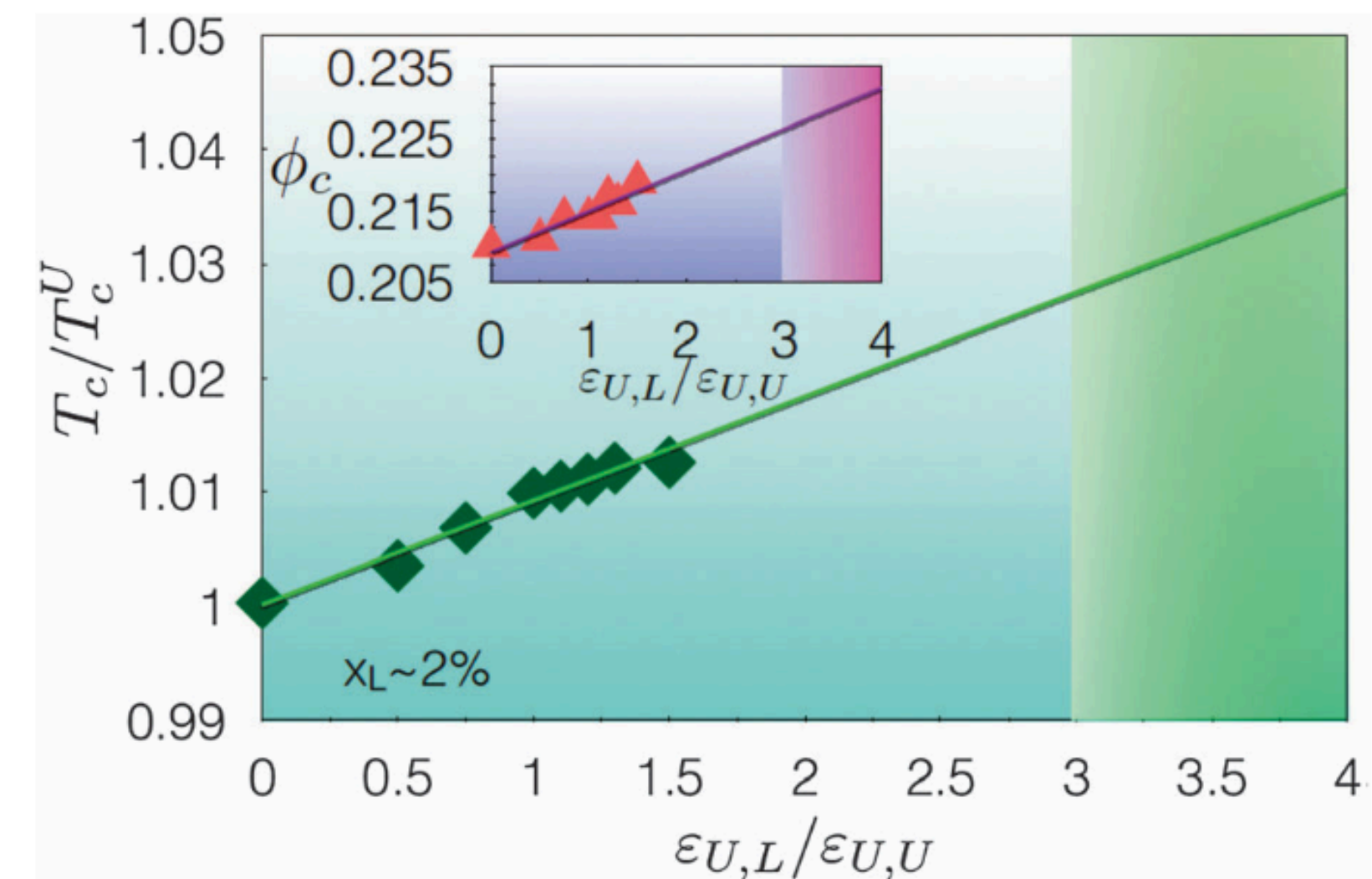


Fig. 1 Patchy particles used for modelling HGD proteins. Particles with four patches (U-patches) are unlabelled proteins (U-type), while the particle with green (wider) patch (L-patch) corresponds to a fluorescently labelled protein (L-type).

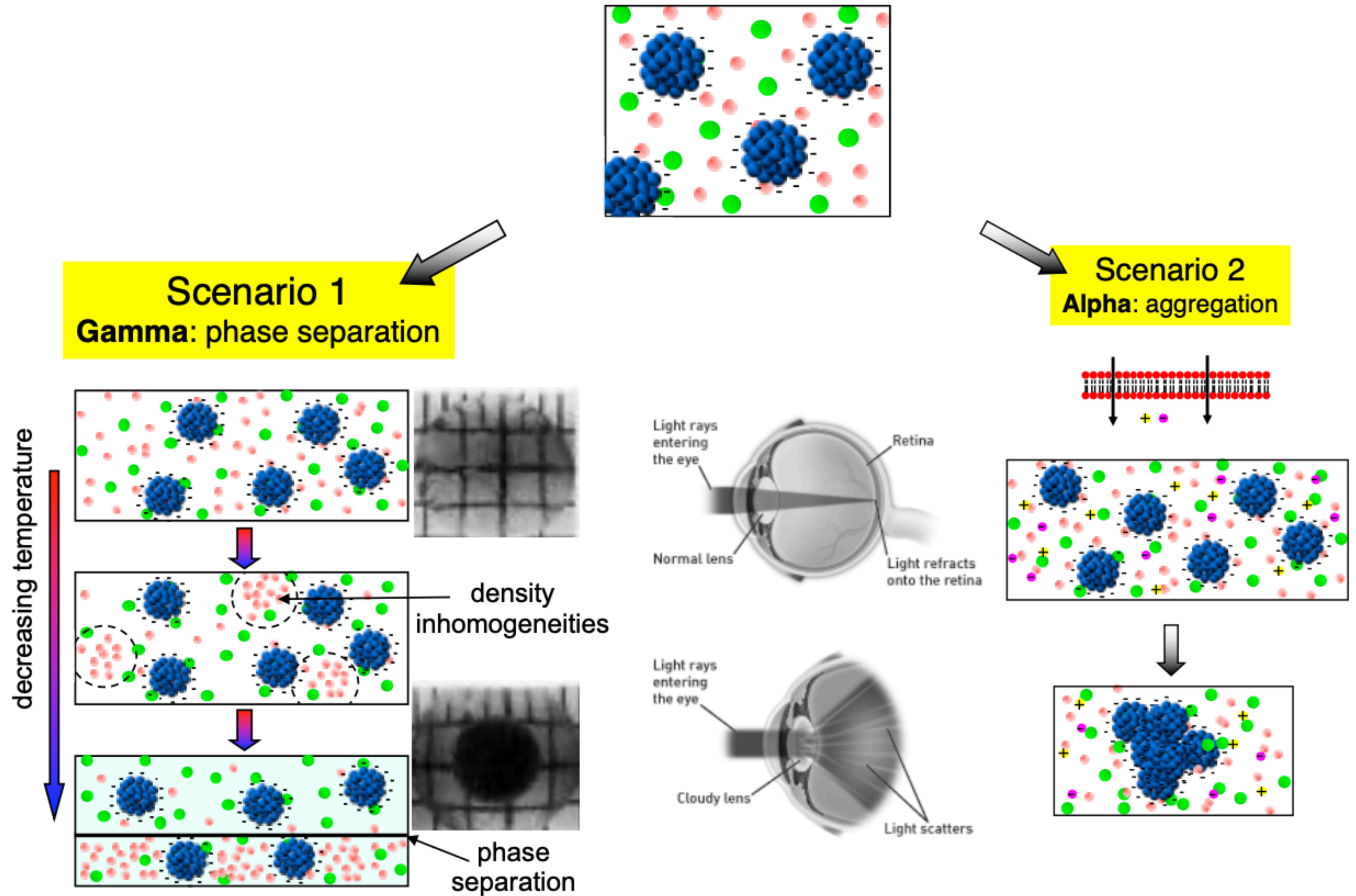
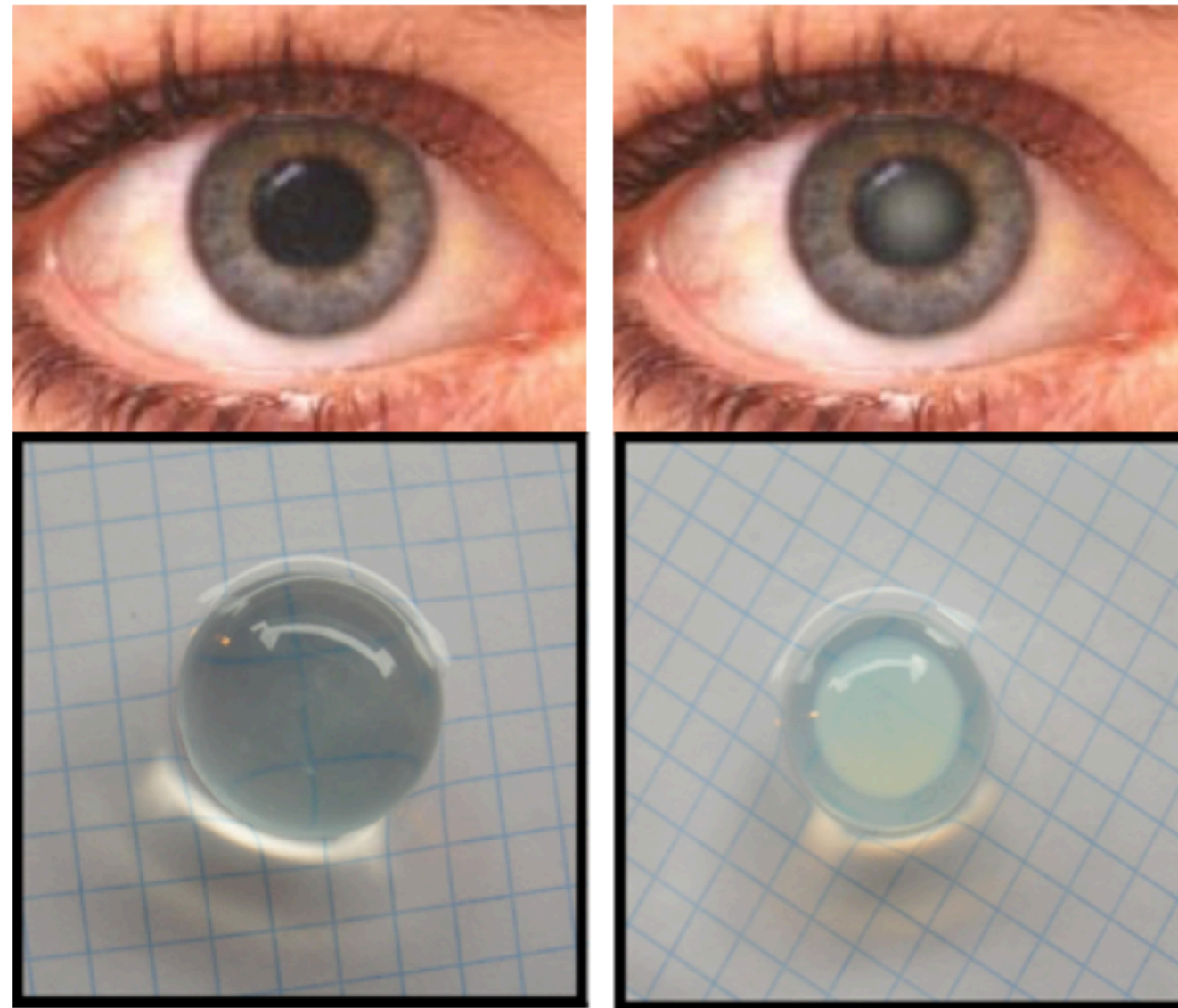


Phys. Chem. Chem. Phys. (2015)

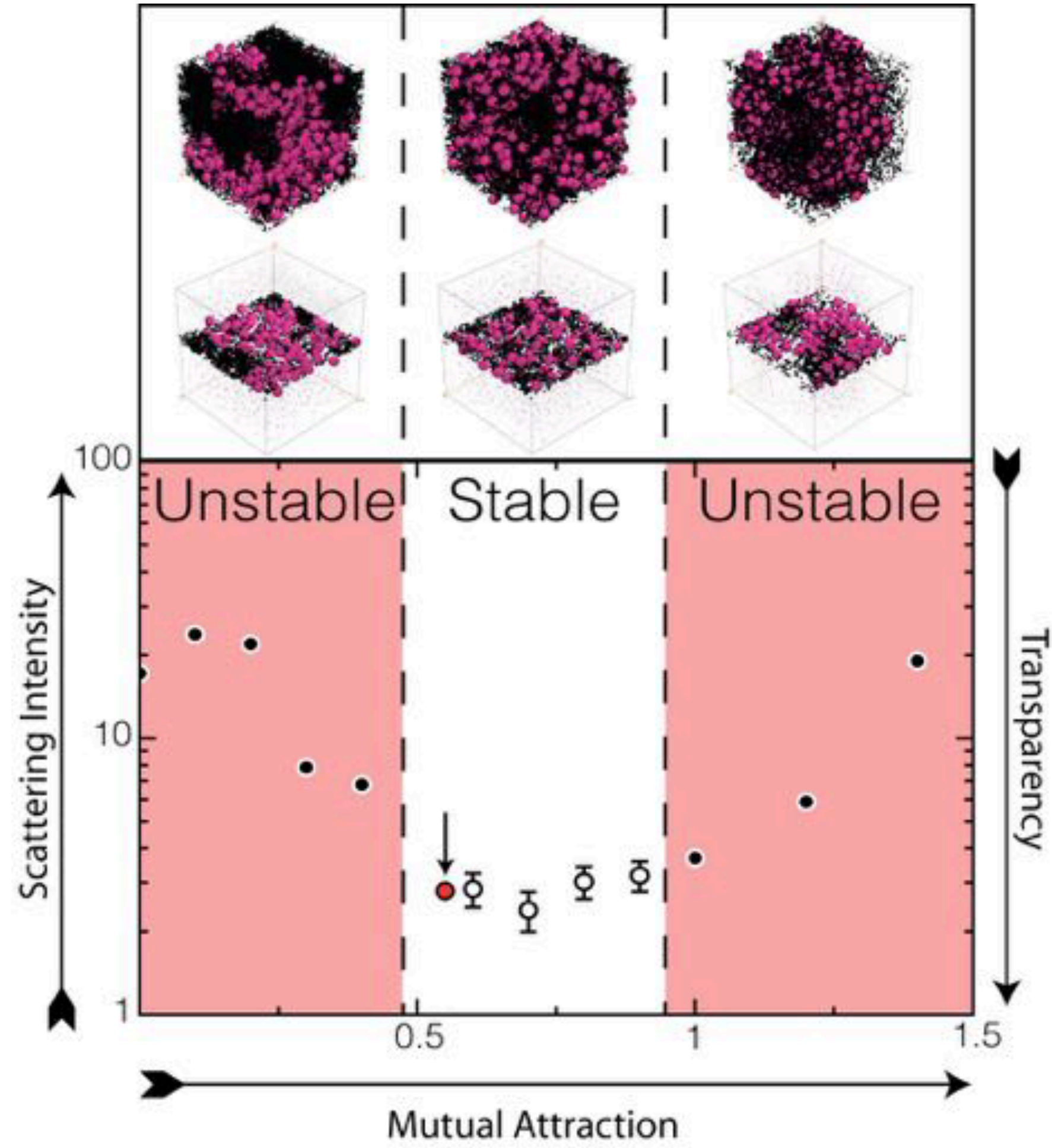


# Back to the eye-lens

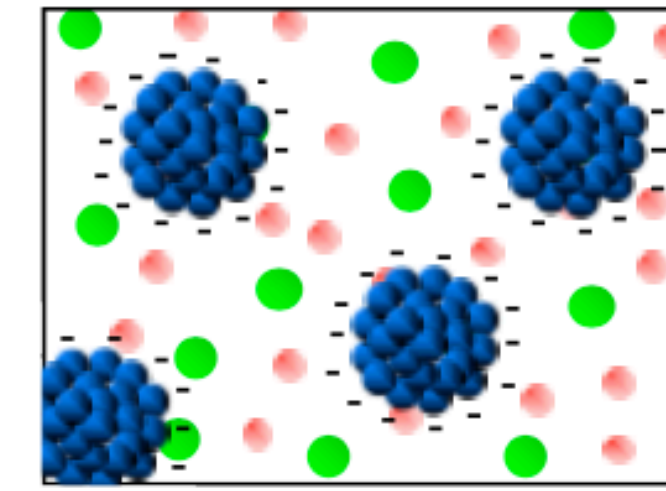
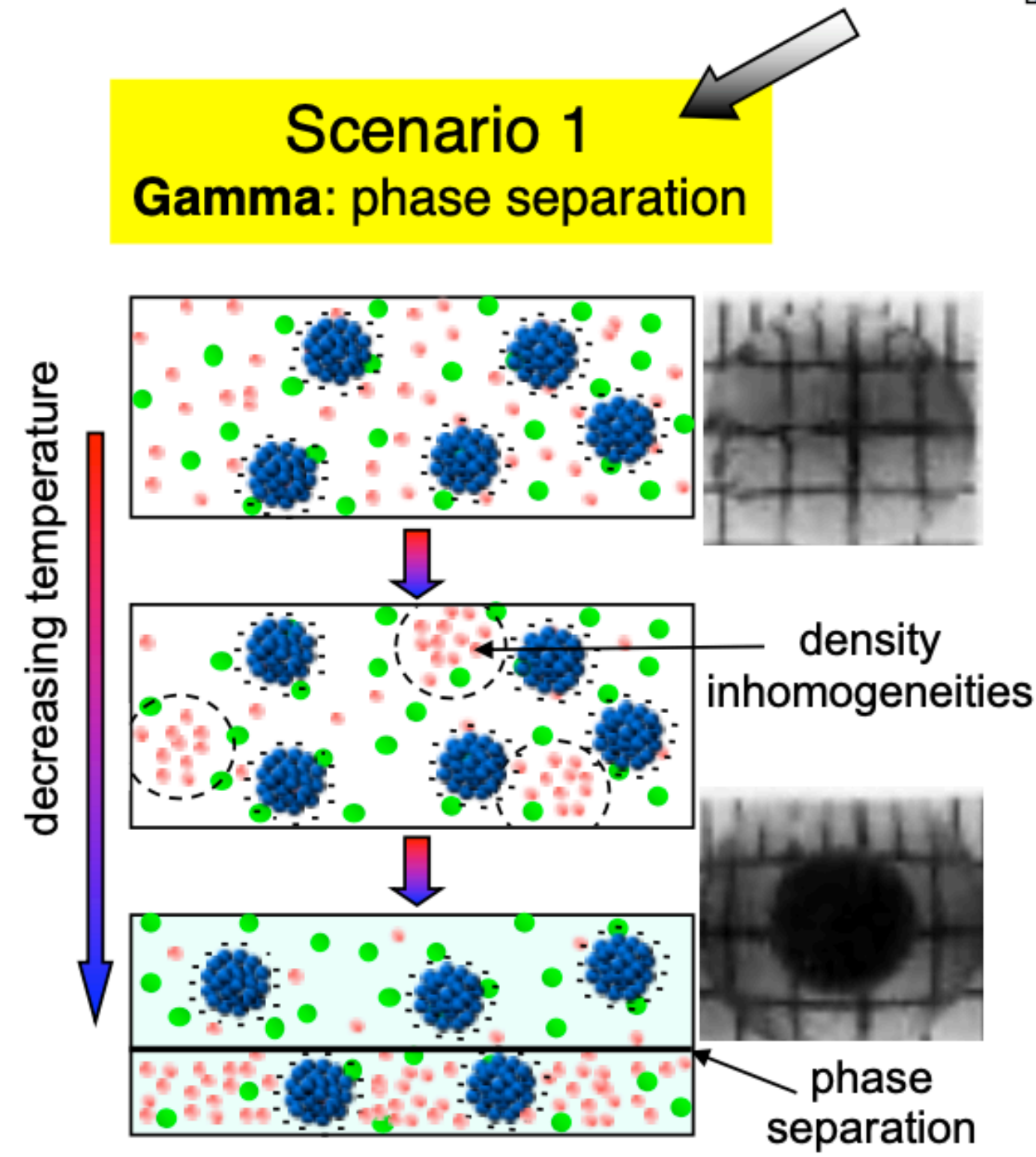
**Cataract:** protein condensation disease, where protein aggregation and phase separation lead to a *clouding of the eye lens*; cataract is still the leading cause of blindness worldwide



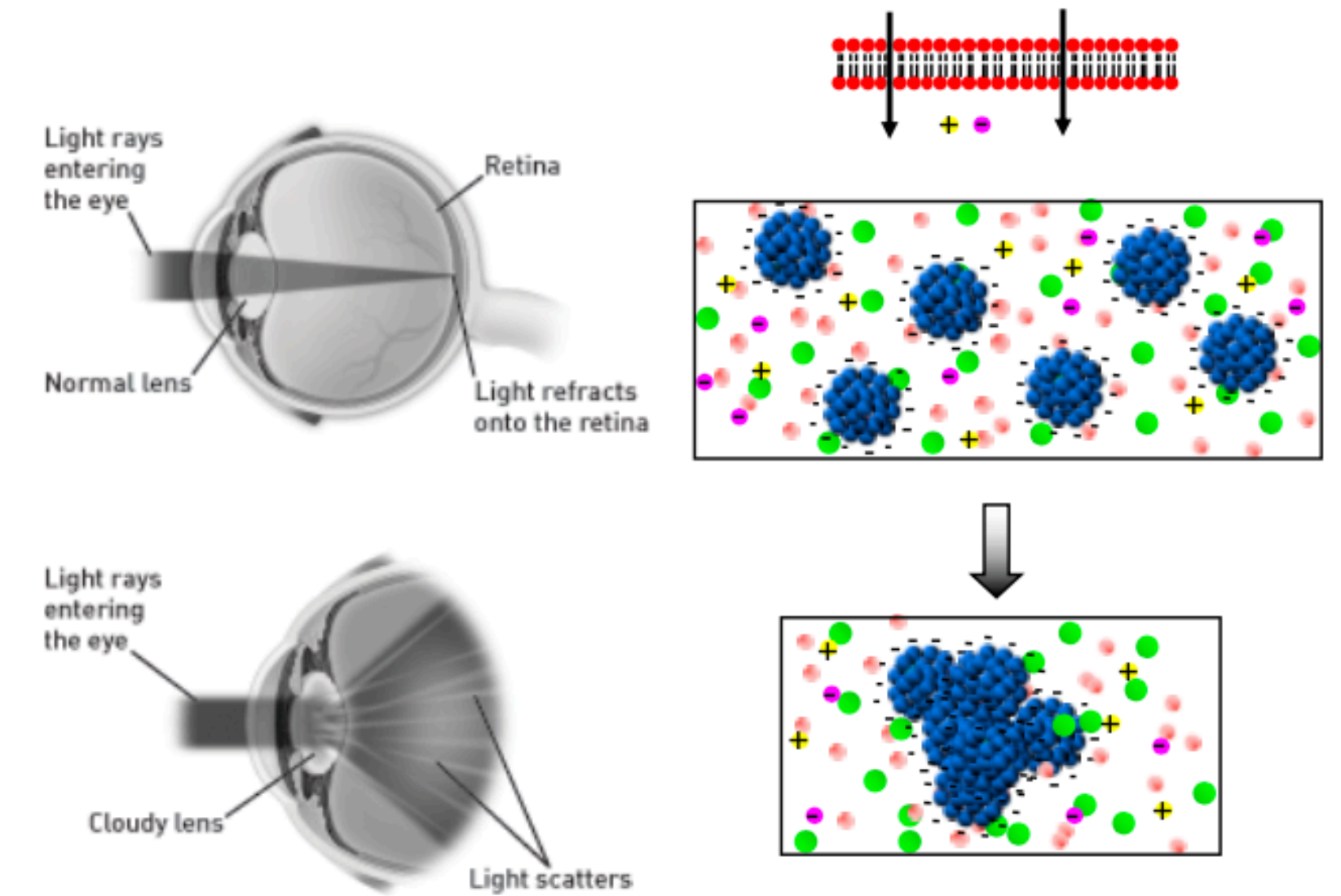
# Back to the eye-lens



Stradner et al  
Phys. Rev. Lett. 2007

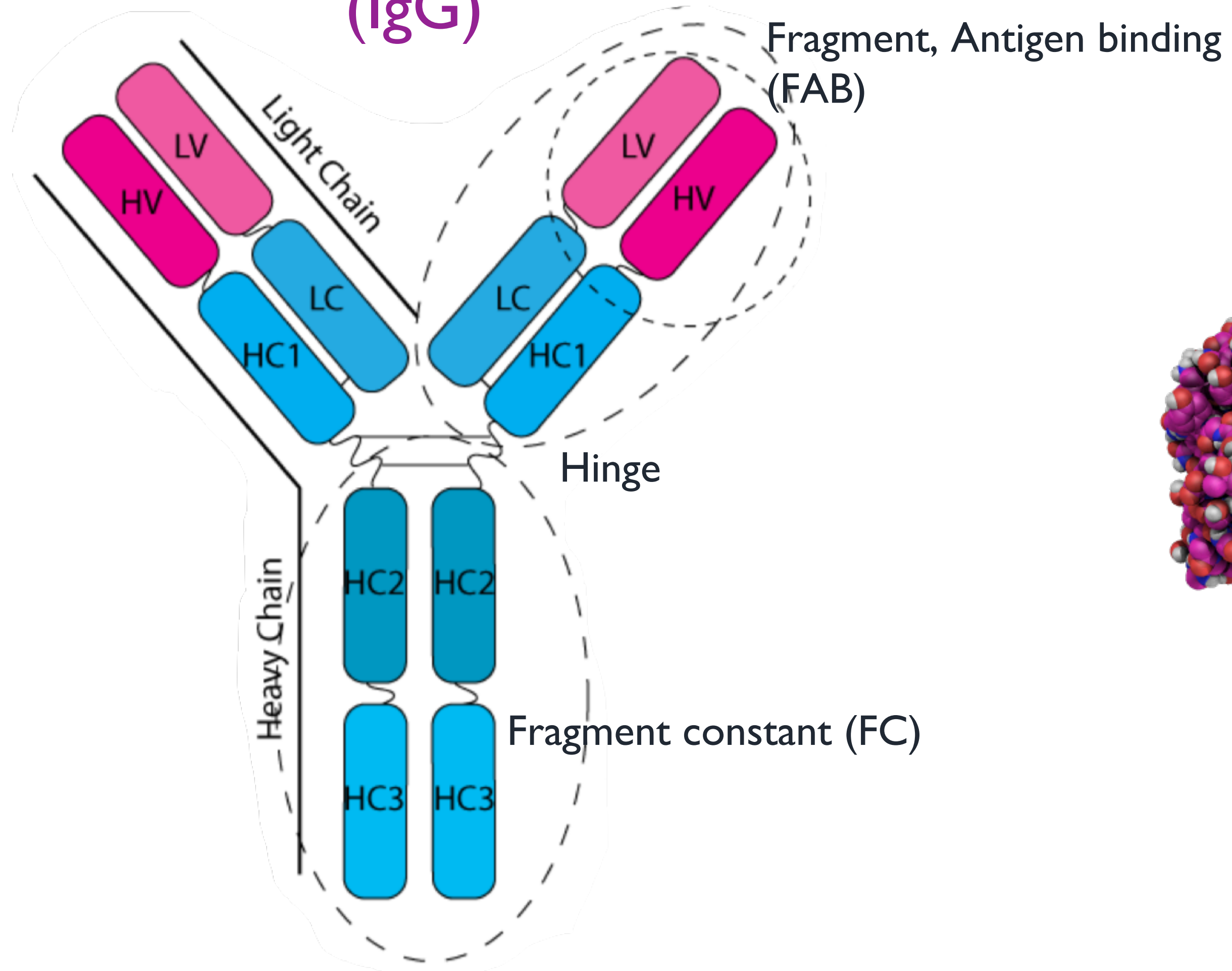


**Scenario 2**  
**Alpha: aggregation**

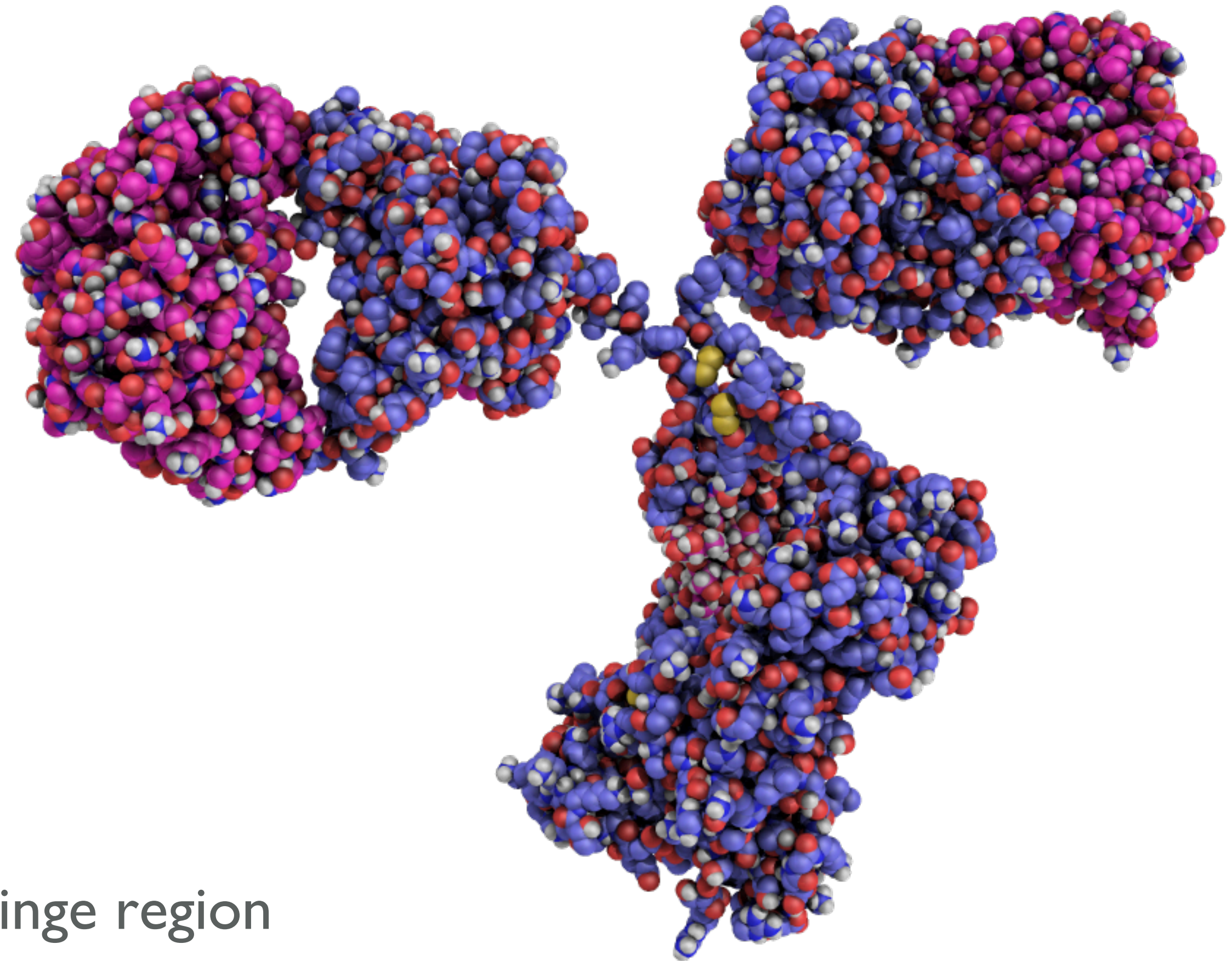


# The last example: monoclonal antibodies

## Immunoglobuline Gamma (IgG)

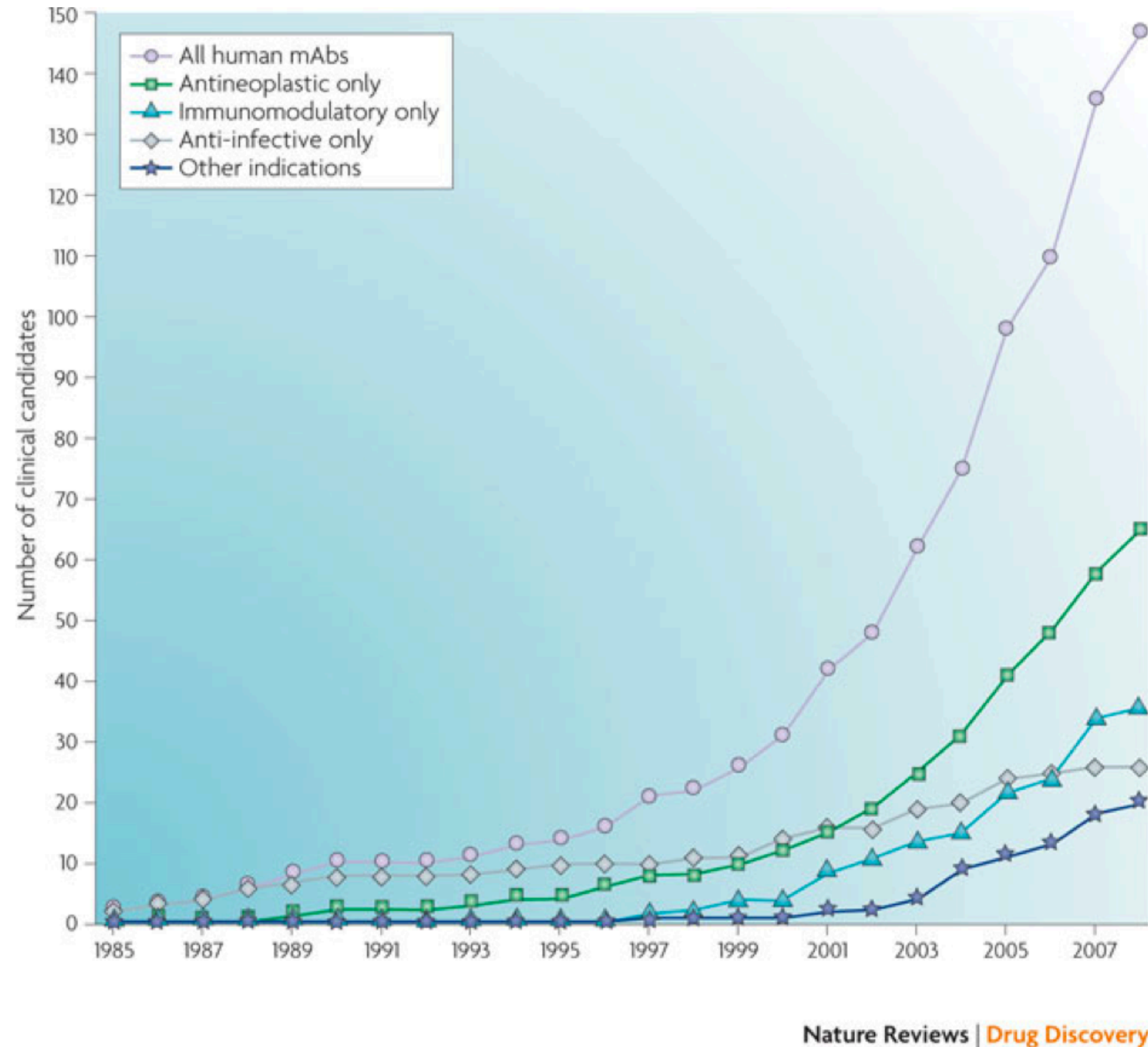


Antibodies are **large** proteins (~150 kDa) employed by the immune system for binding targets



three structured domains connected by a flexible hinge region

# Antibodies as pharmaceutical drugs



widely investigated in biopharmaceutical industry due to

- i) large flexibility in molecular recognition
- ii) long half-life in the body
- iii) possibility of humanization with low risk of immunogenicity

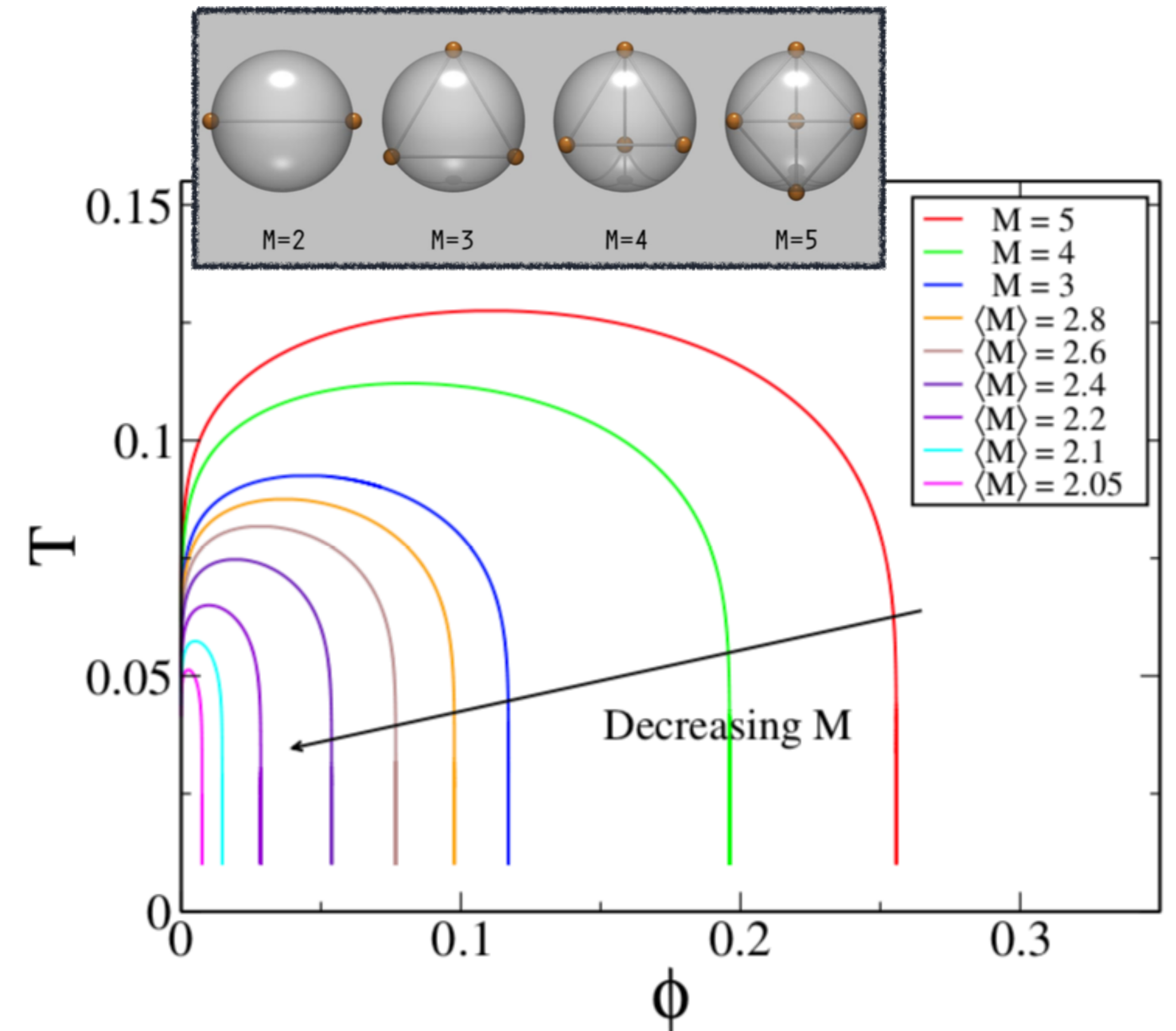
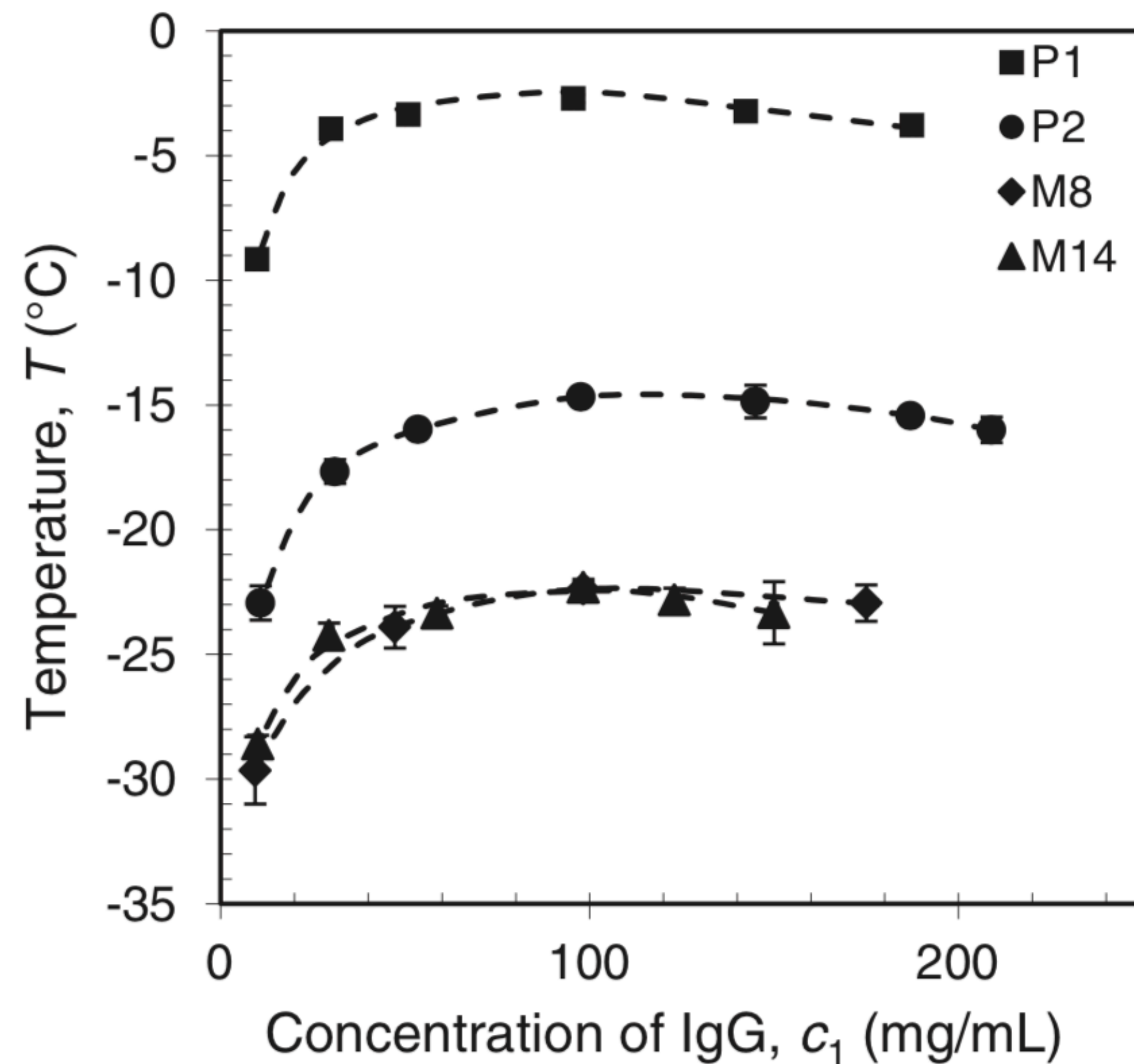
**NEED FOR HIGH CONCENTRATION FORMULATIONS  
> 100 MG/ML  
OFTEN RESULTING IN TOO HIGH VISCOSITIES**



# Phase separation of antibodies solutions

## Phase transitions in human IgG solutions

Ying Wang, Aleksey Lomakin, Ramil F. Latypov, Jacob P. Laubach, Teru Hideshima, Paul G. Richardson, Nikhil C. Munshi, Kenneth C. Anderson, and George B. Benedek



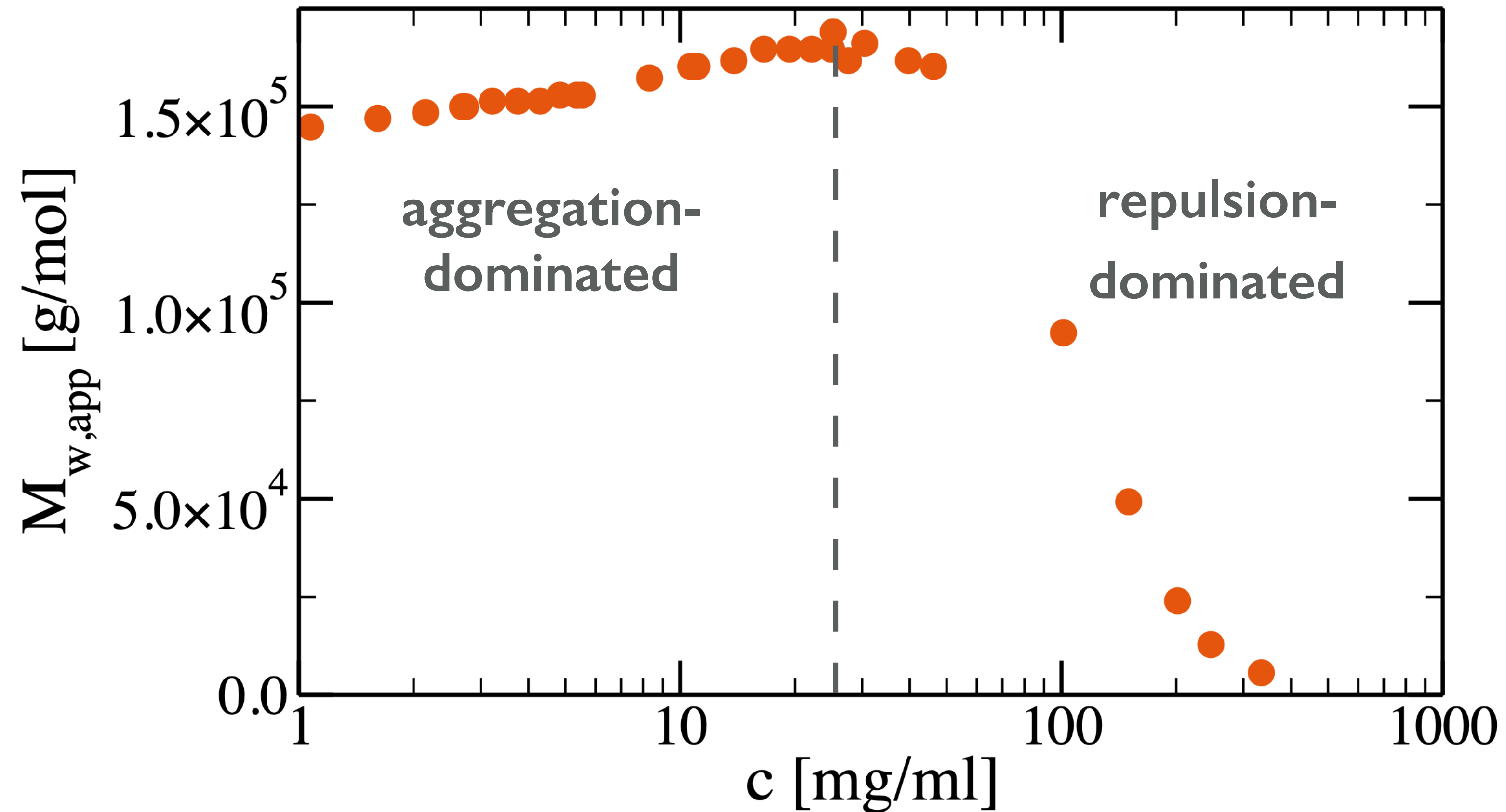
exceptionally low critical volume fraction  $\Phi \sim 0.07$

analogy with patchy particles

# Experimental results for IgG4: Static light scattering

added 10mM NaCl, pH=6.5, T=25°C

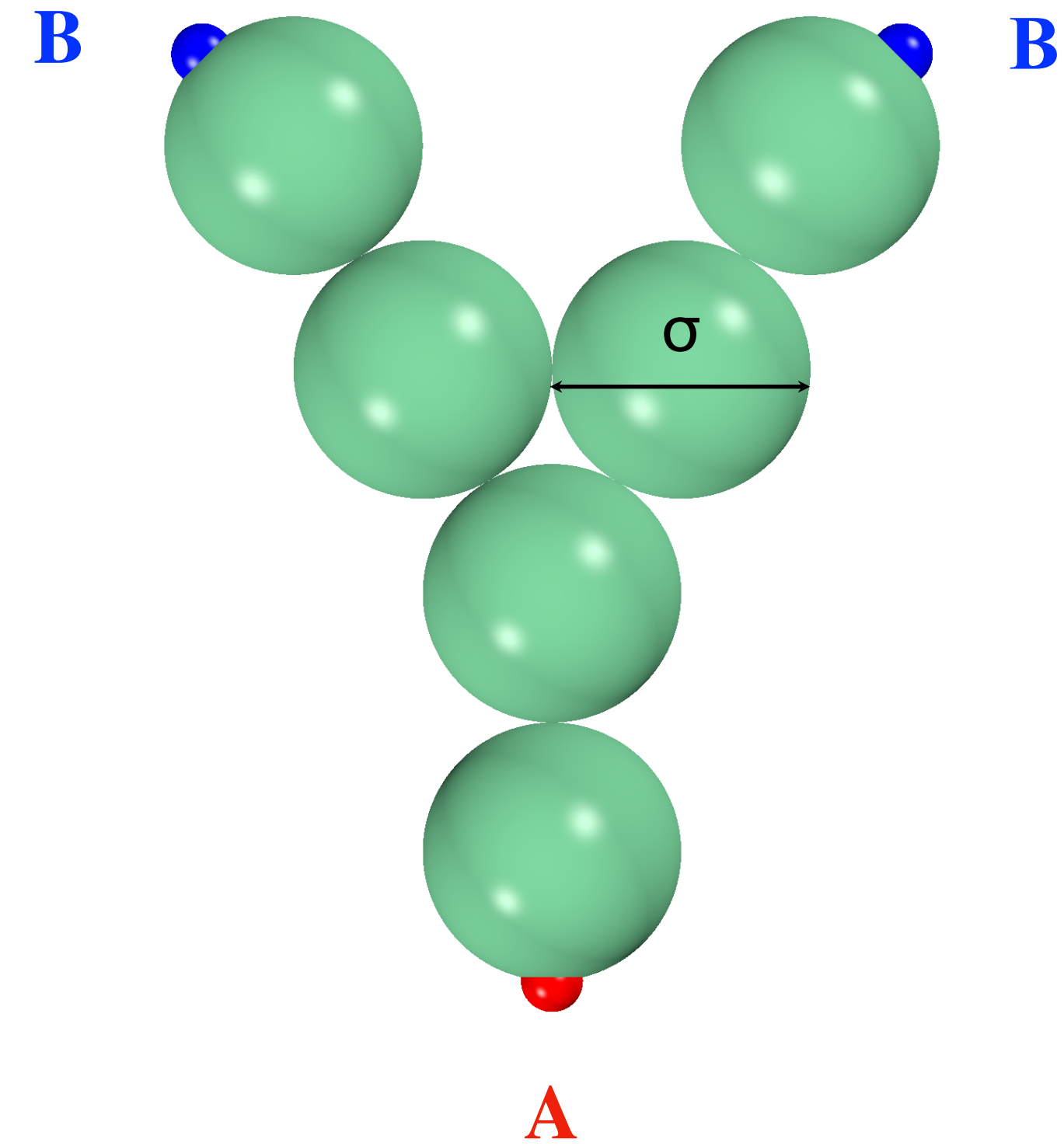
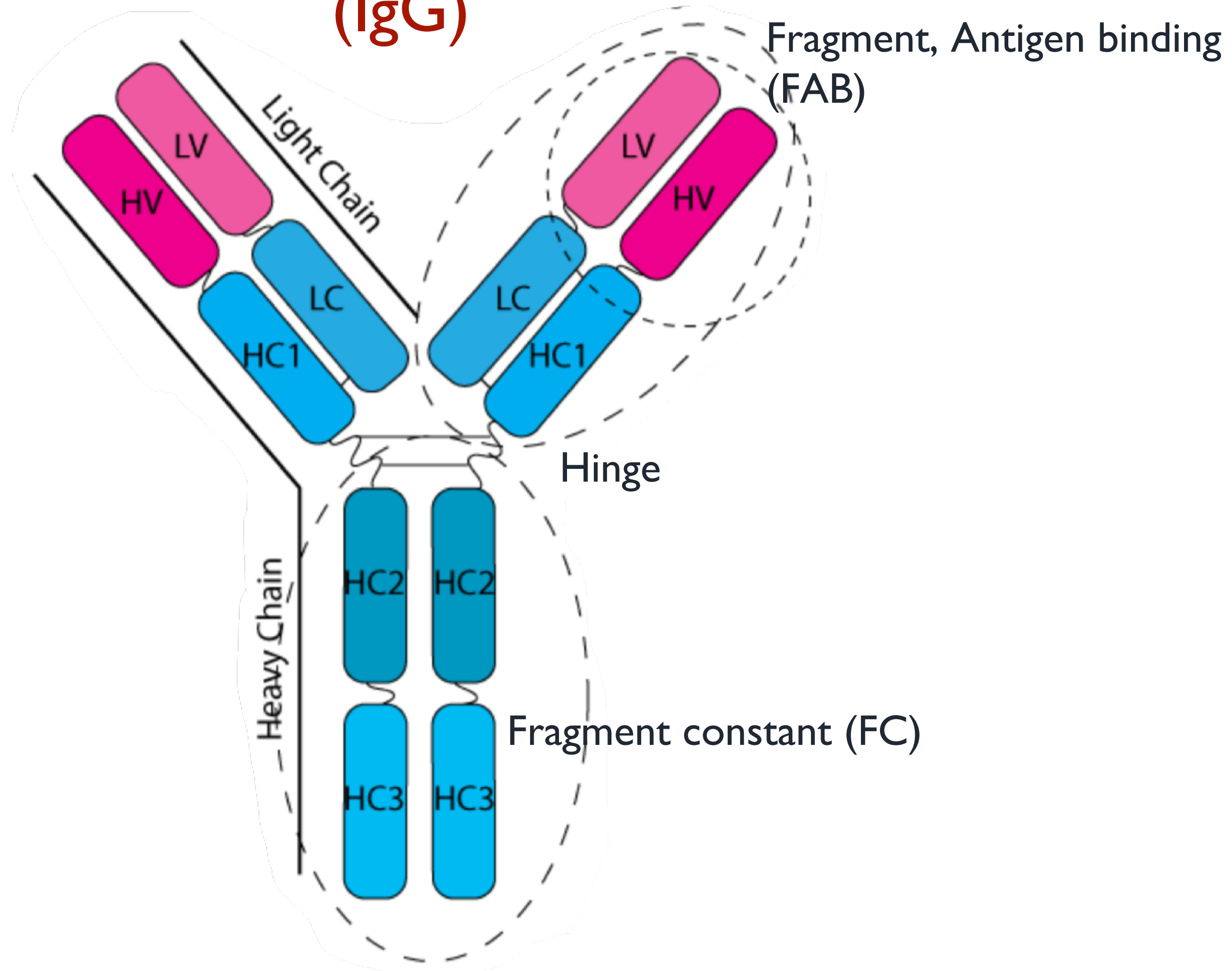
NO PHASE SEPARATION OBSERVED



Apparent molecular weight shows a maximum at  $c \sim 25$  mg/ml

# A patchy model for antibodies

## Immunoglobuline Gamma (IgG)



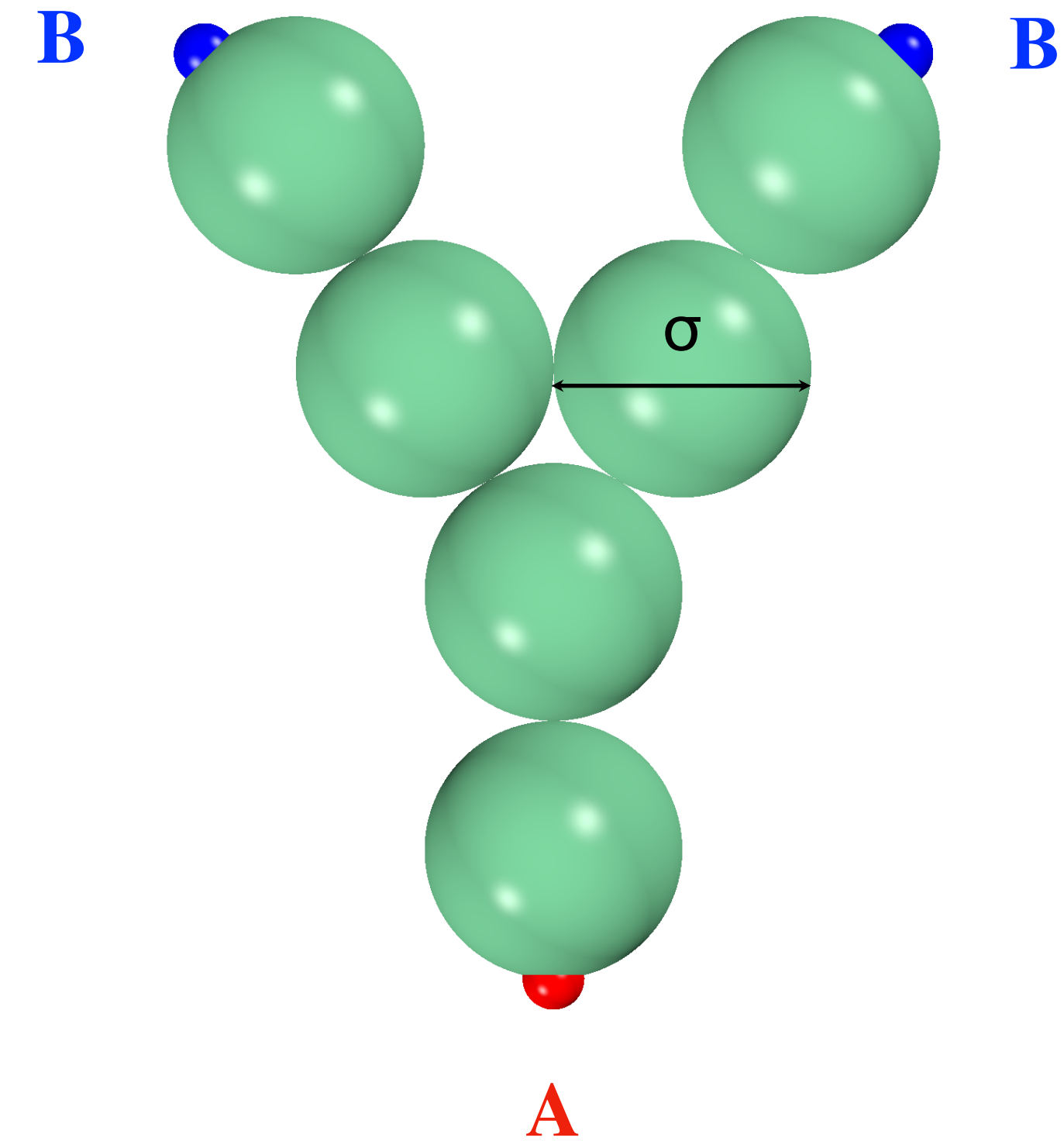
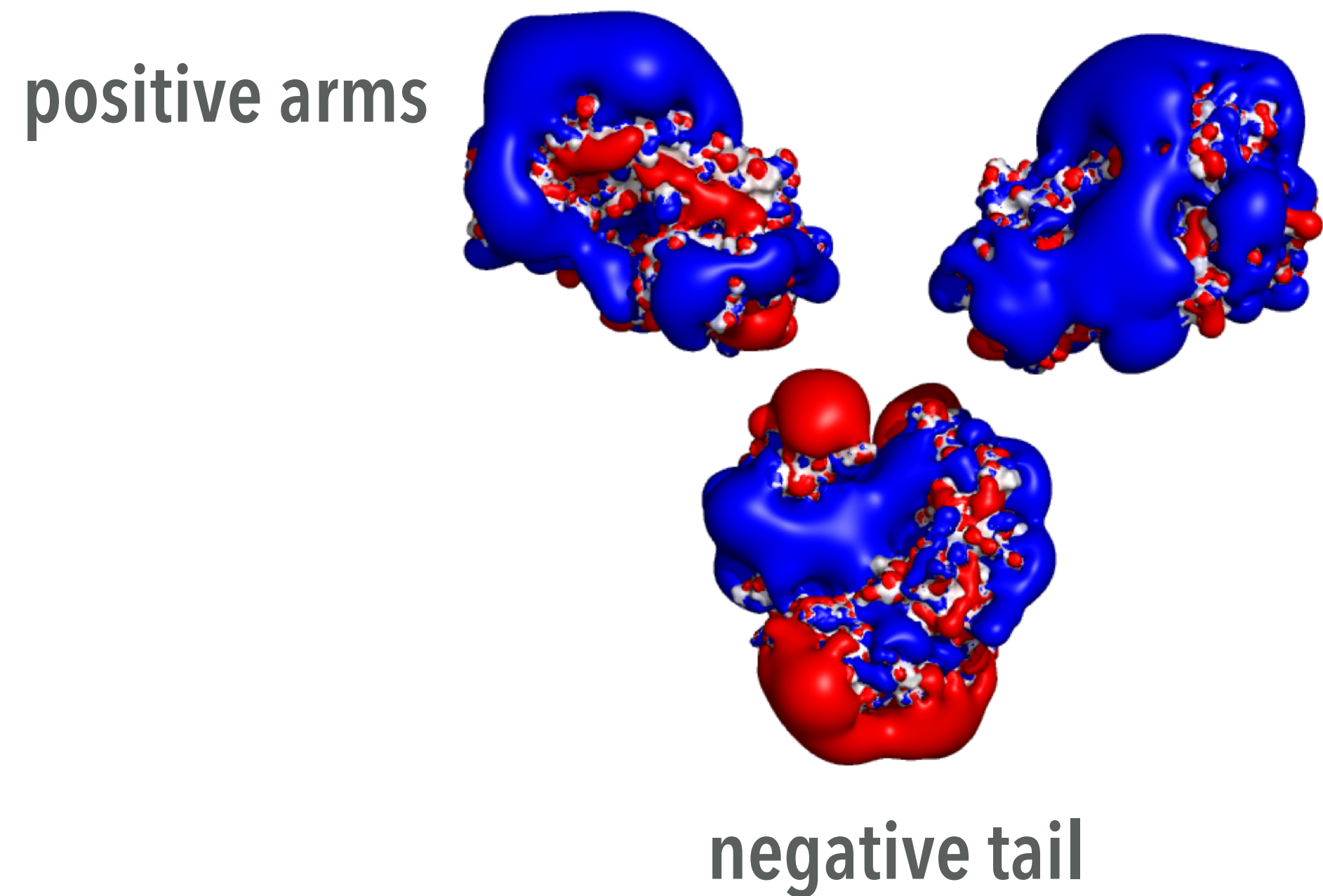
Main ingredients:

- 6 hard spheres in a symmetric Y-shape;
- rigid structure;
- decorated with three attractive patches:

**2 patches of type A and 1 patch of type B**

# A patchy model for antibodies

electrostatic iso-surface calculations



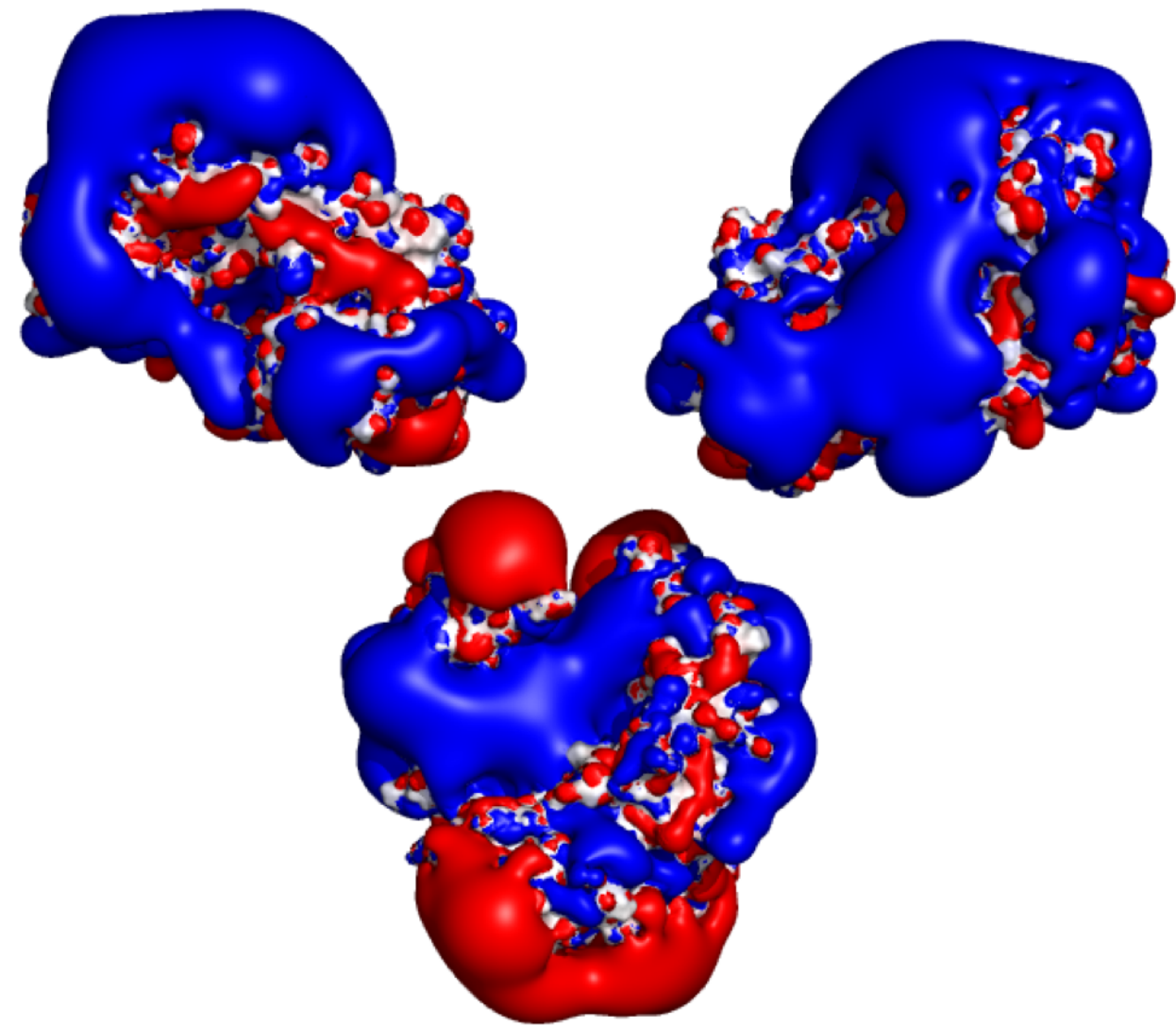
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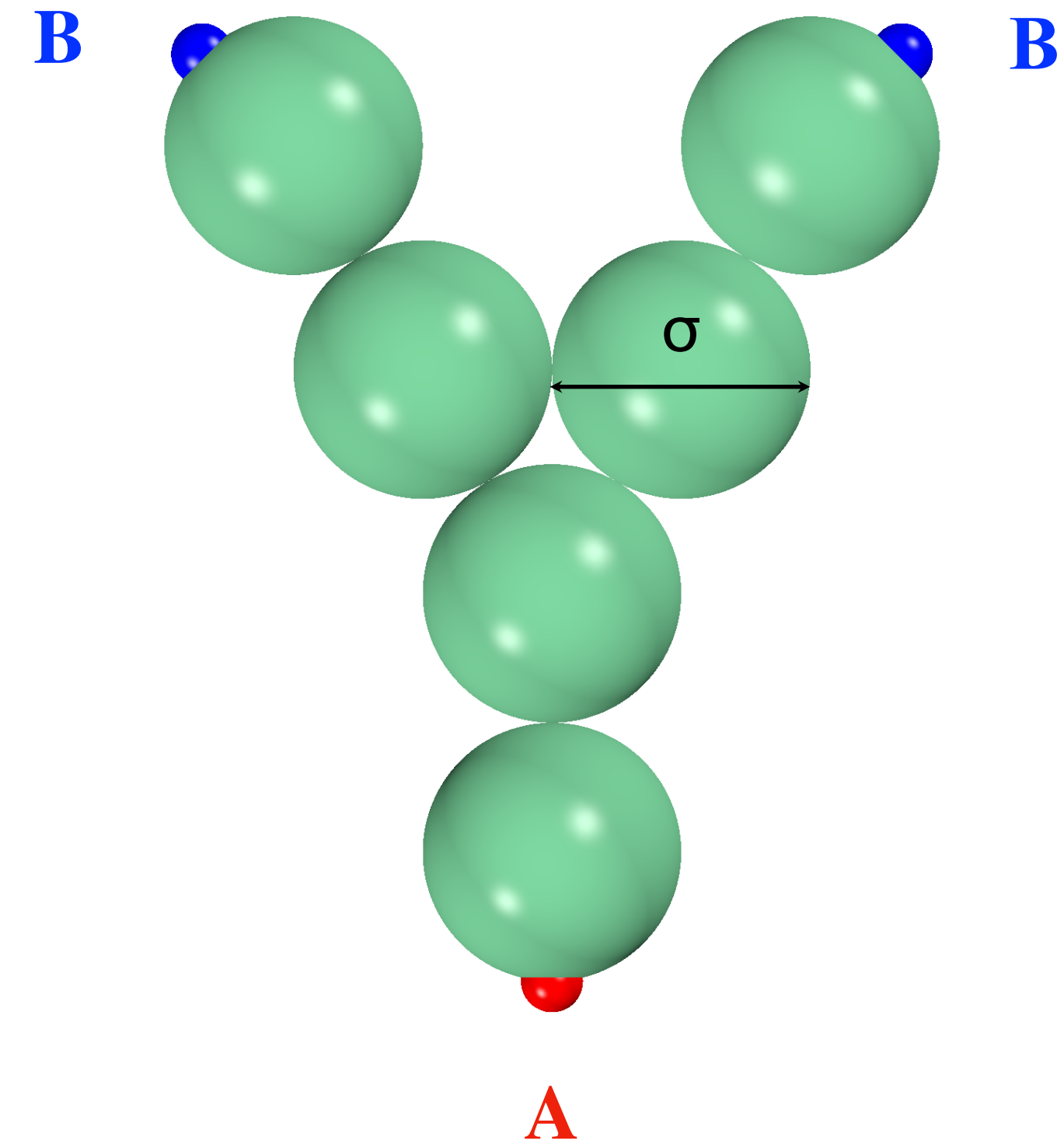
# A patchy model for antibodies

electrostatic iso-surface calculations

positive arms



negative tail



we consider only

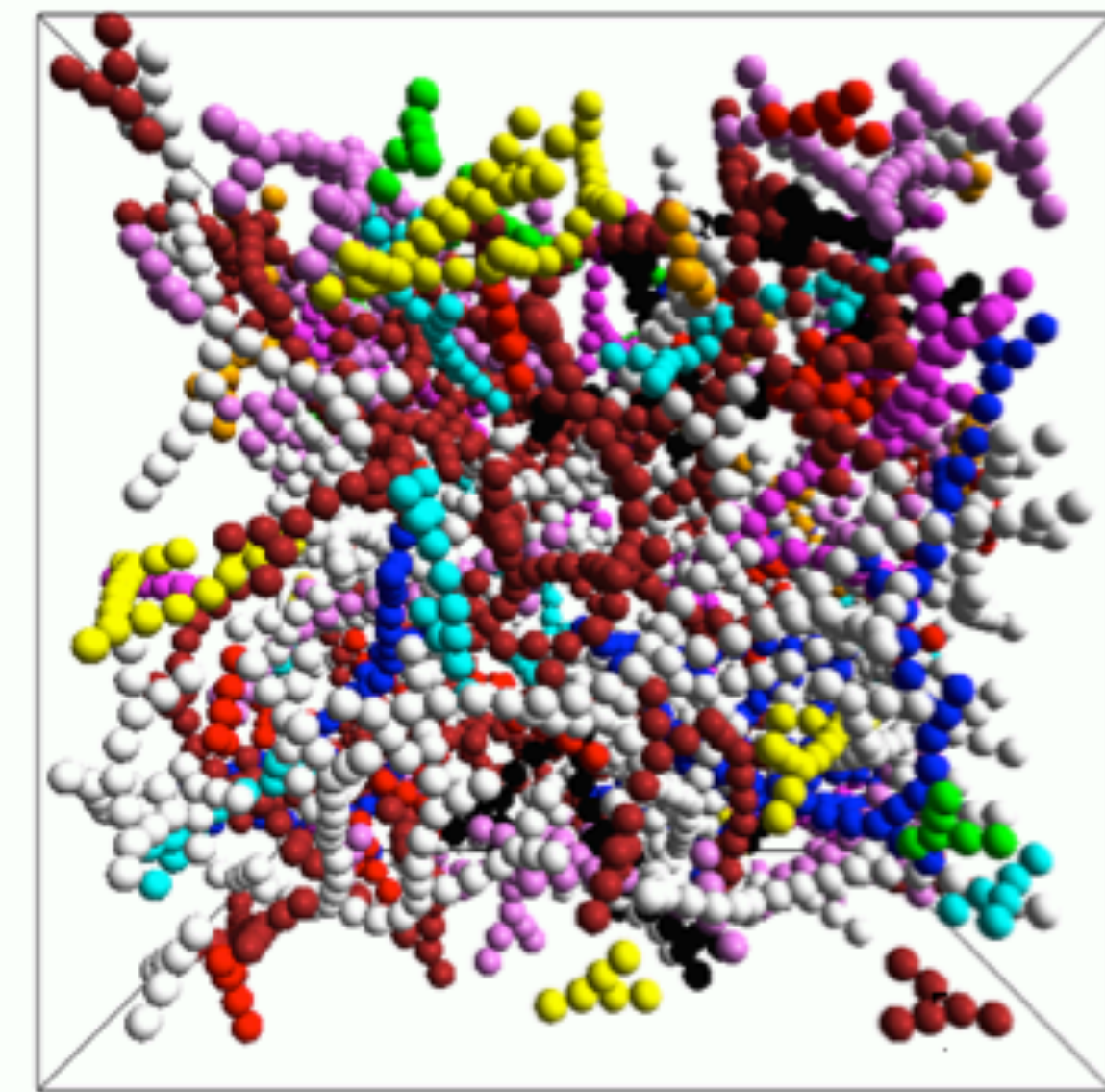
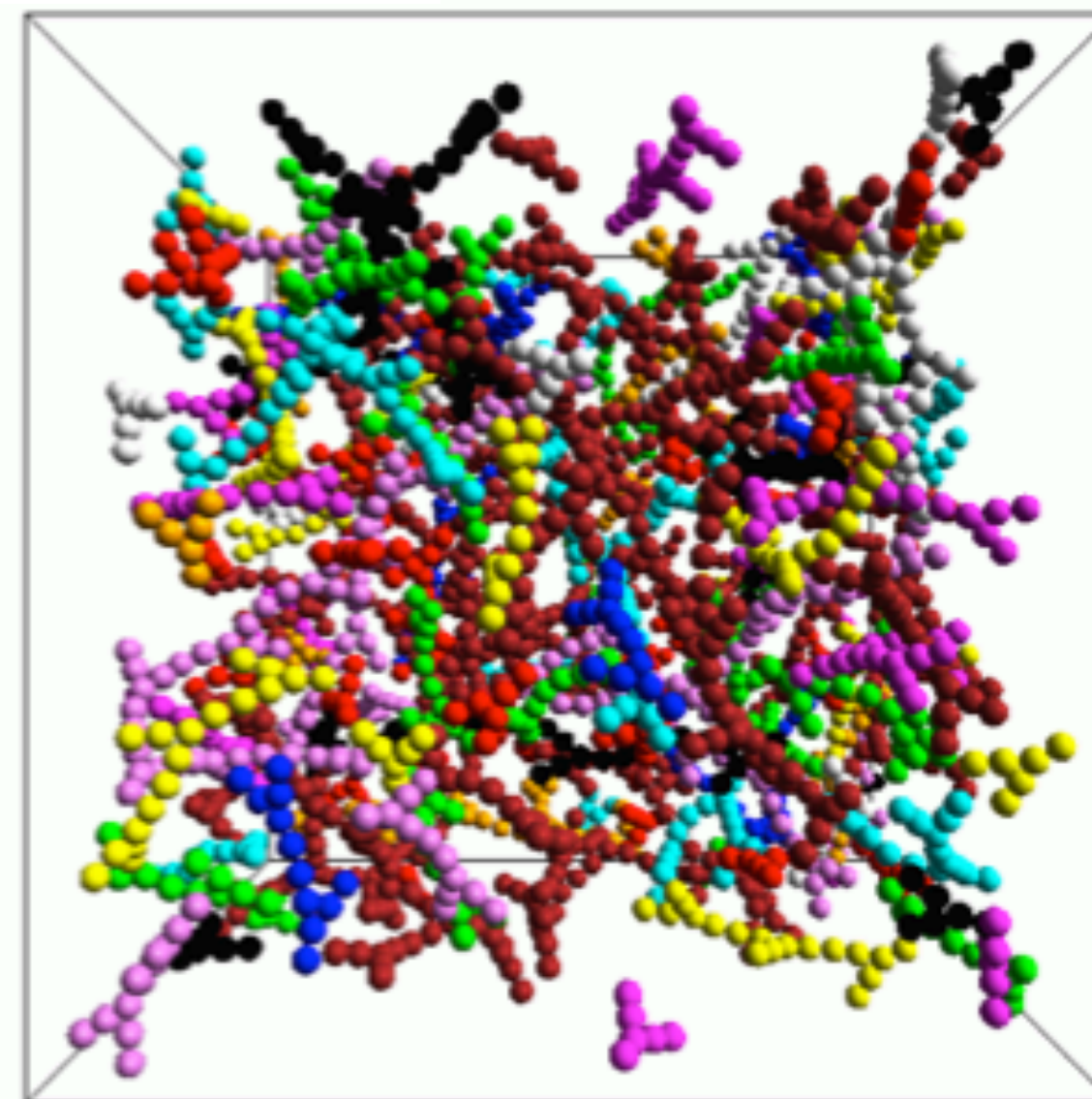
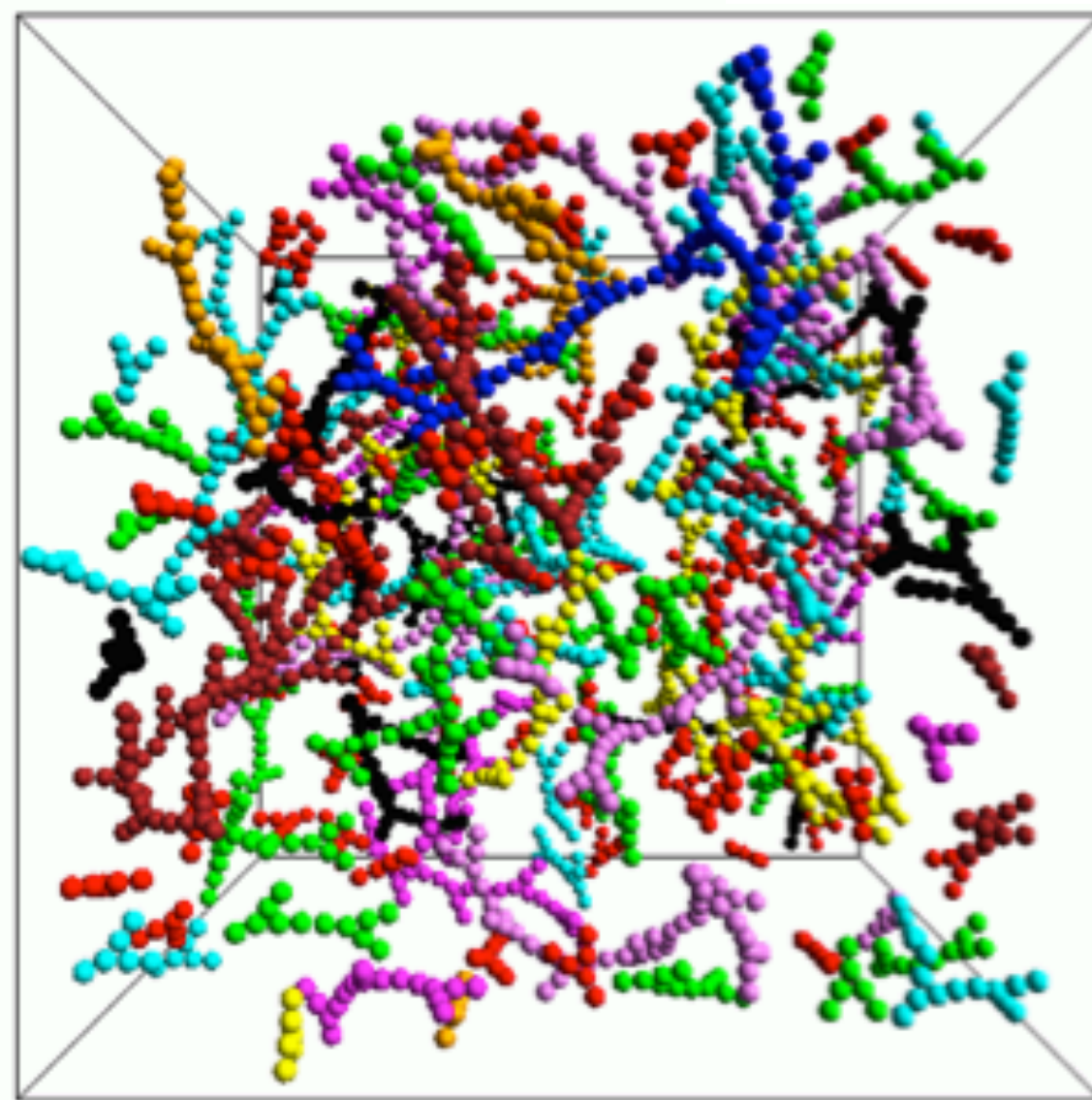
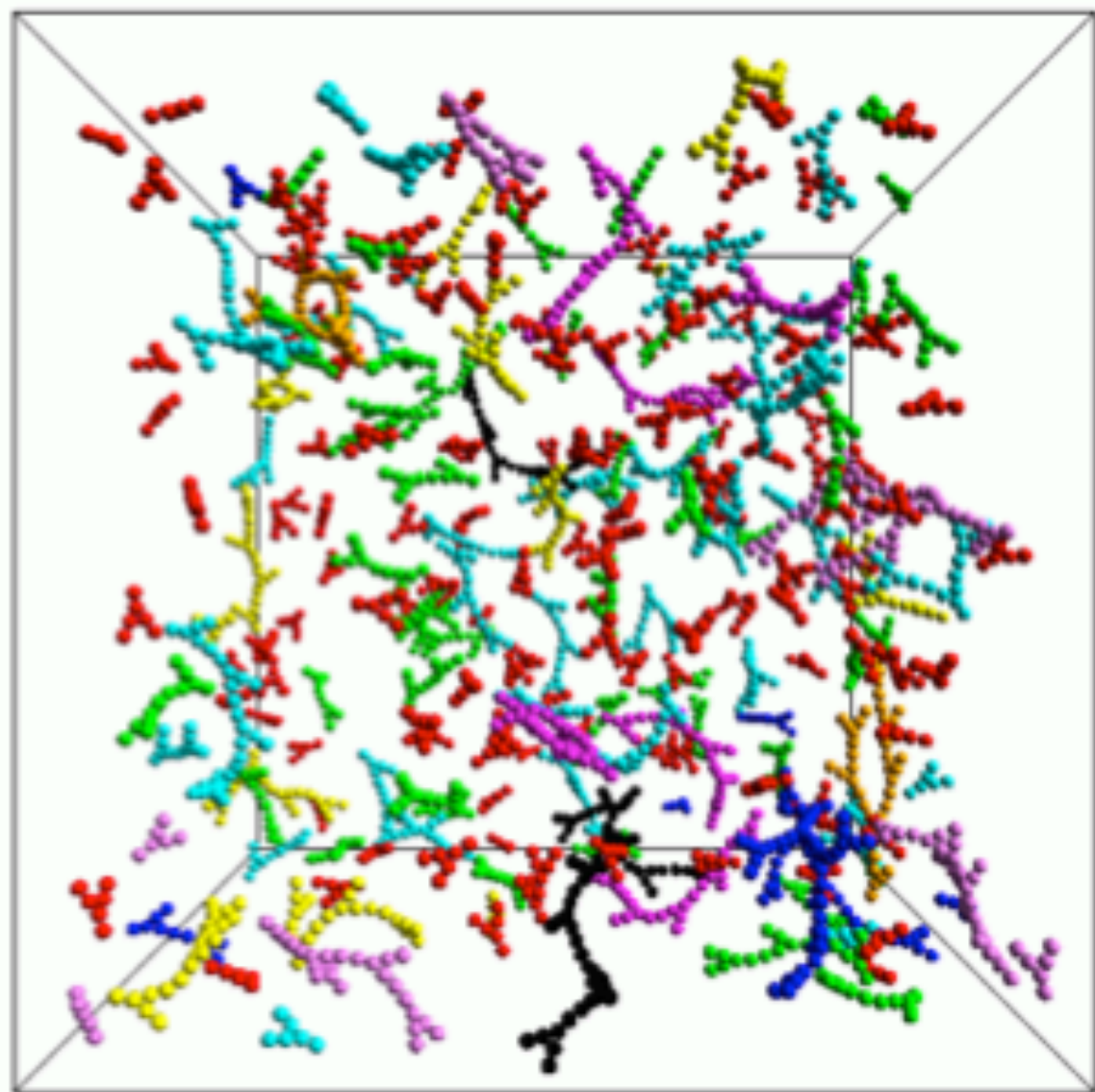
**AB (head-tail) bonds**

**square-well** attraction

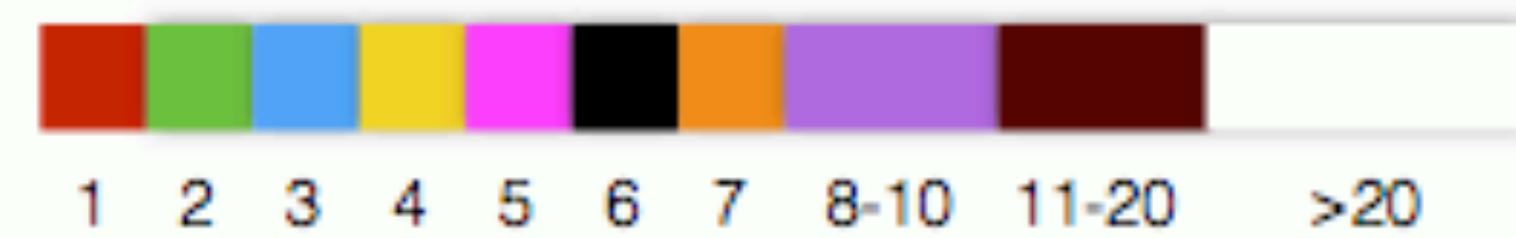
fulfilling *one-bond-per-patch* condition

# Monte Carlo Simulations

$$k_B T/\epsilon=0.0775$$



increasing concentration



cluster size colour bar

- finite-size clusters only
- weak aggregation also at very low temperatures
- no percolation is found; no phase separation

# A patchy model for antibodies

Hyperbranched polymer model

**ANALYTICALLY SOLVABLE:**

thermodynamic properties



*Wertheim theory*

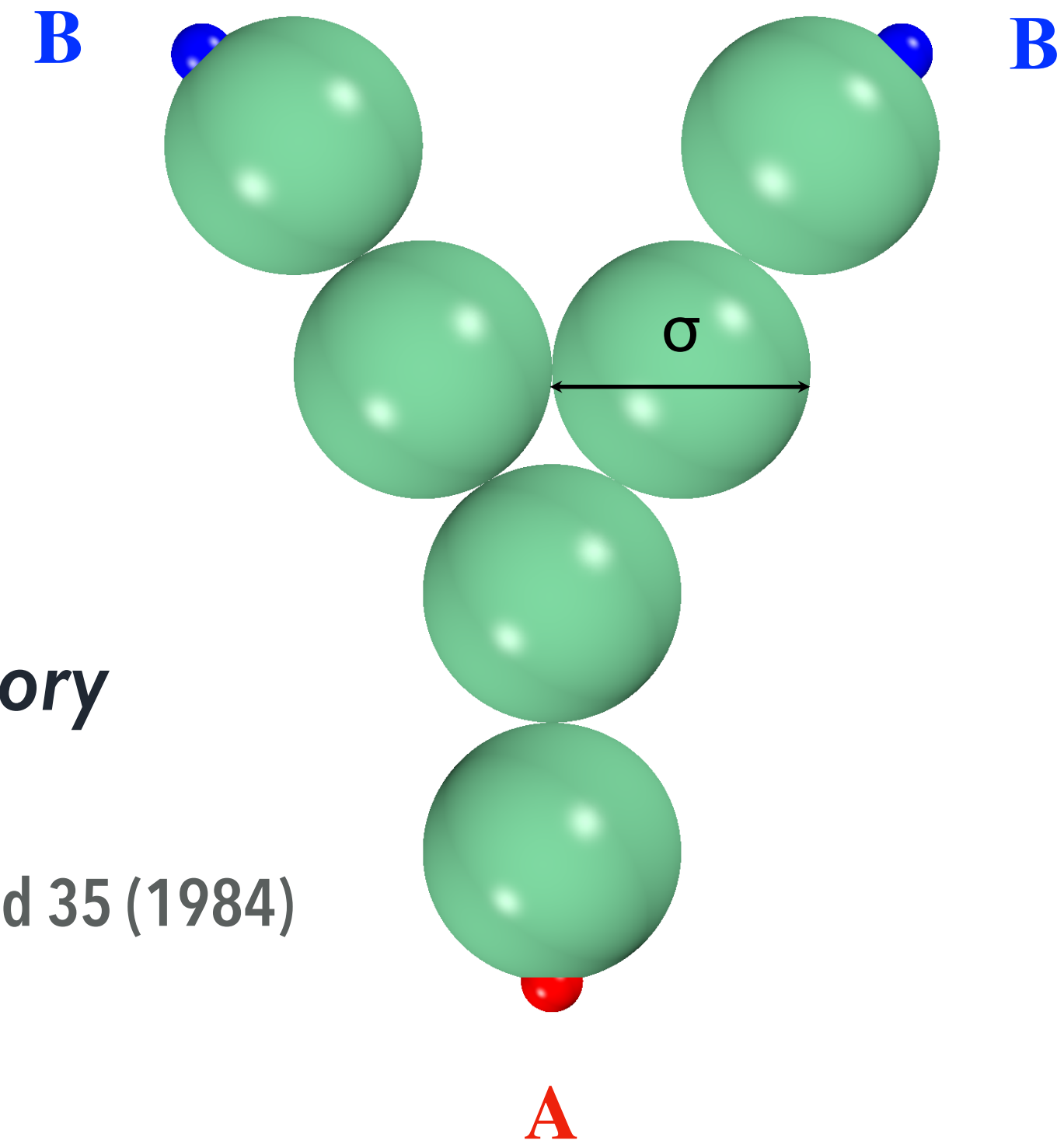
M.S. Wertheim, J. Stat. Phys. 35, 19 and 35 (1984)

connectivity properties  
*polymers theory*



**hyperbranched**

M. Rubinstein and R.H. Colby Polymer Physics (Oxford University Press), Oxford, 2003



**AB (head-tail) bonds**

**square-well** attraction

fulfilling *one-bond-per-patch* condition

# Comparison between Wertheim theory and Monte Carlo Simulations

Thermodynamic perturbative approach

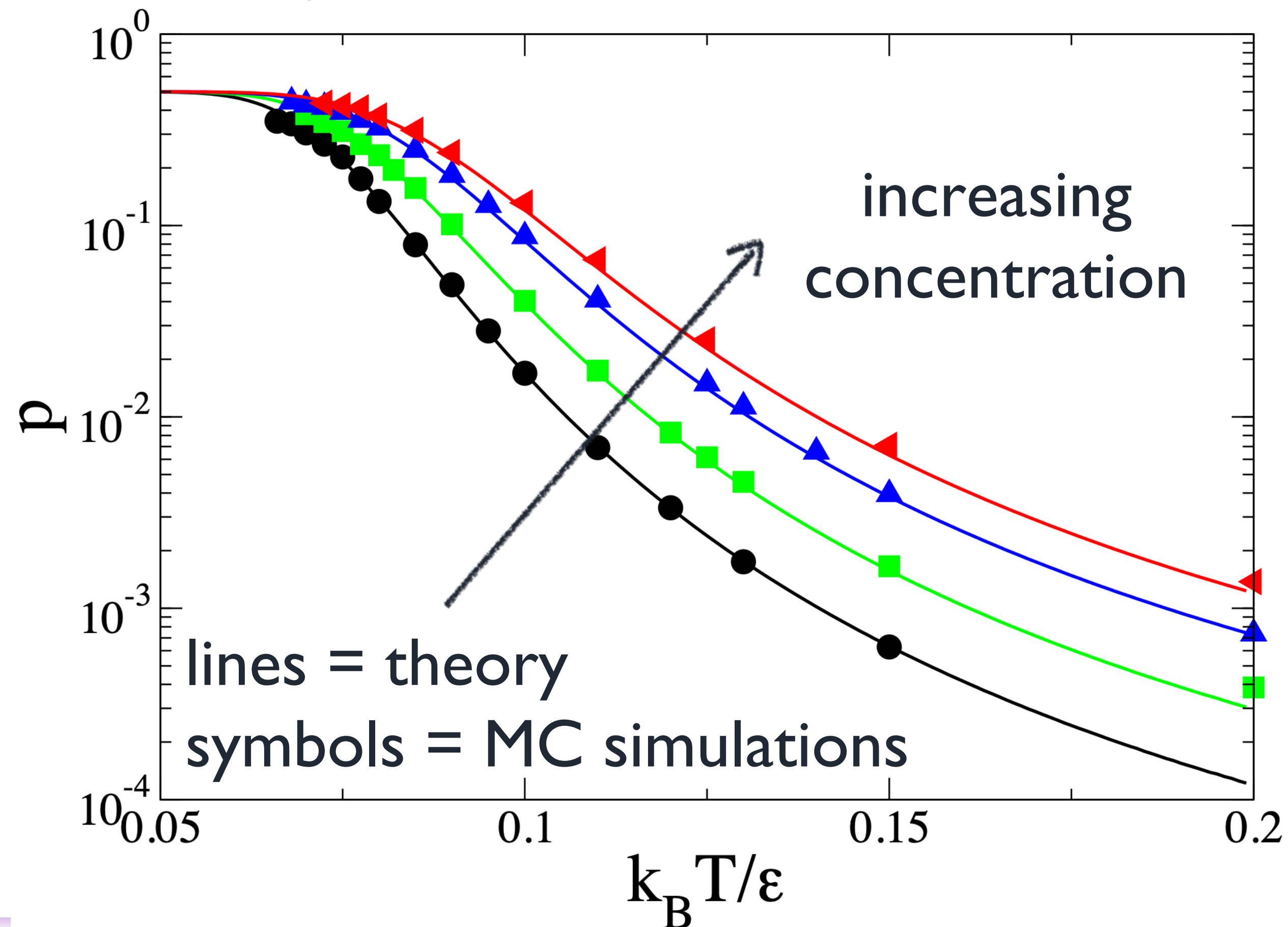
free energy  $F = F_{HS} + F_b$  bond contribution

hard sphere  
reference term

bond probability  $p \equiv p_B = 1 - X_B$

**very good agreement between theory  
and simulations at all temperatures  
and concentrations**

$$\beta \frac{F_b}{N} = 2 \ln X_B + \ln X_A - X_B - \frac{X_A}{2} + \frac{3}{2}$$

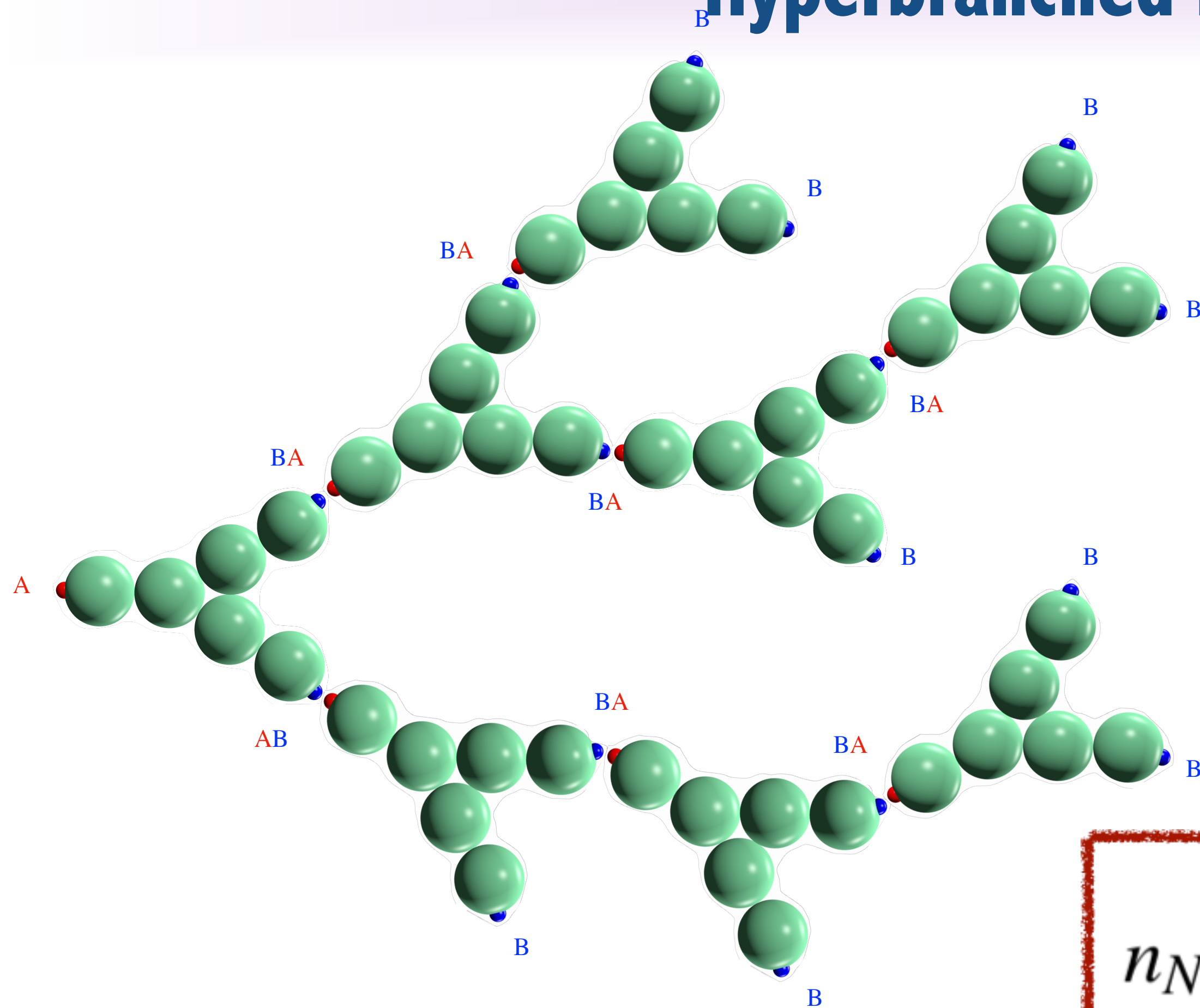




# Hyperbranched Polymers (Flory-Stockmayer)

$AB_{f-1}$  with  $f \geq 2$  and only AB bonds

**Model for branching without gelation**



**cluster size distribution**

$$n_N(p) = \frac{[(f-1)N]!}{N! [(f-2)N+1]!} p^{N-1} (1-p)^{(f-2)N+1}$$

for our IgG4 model

$f = 2$

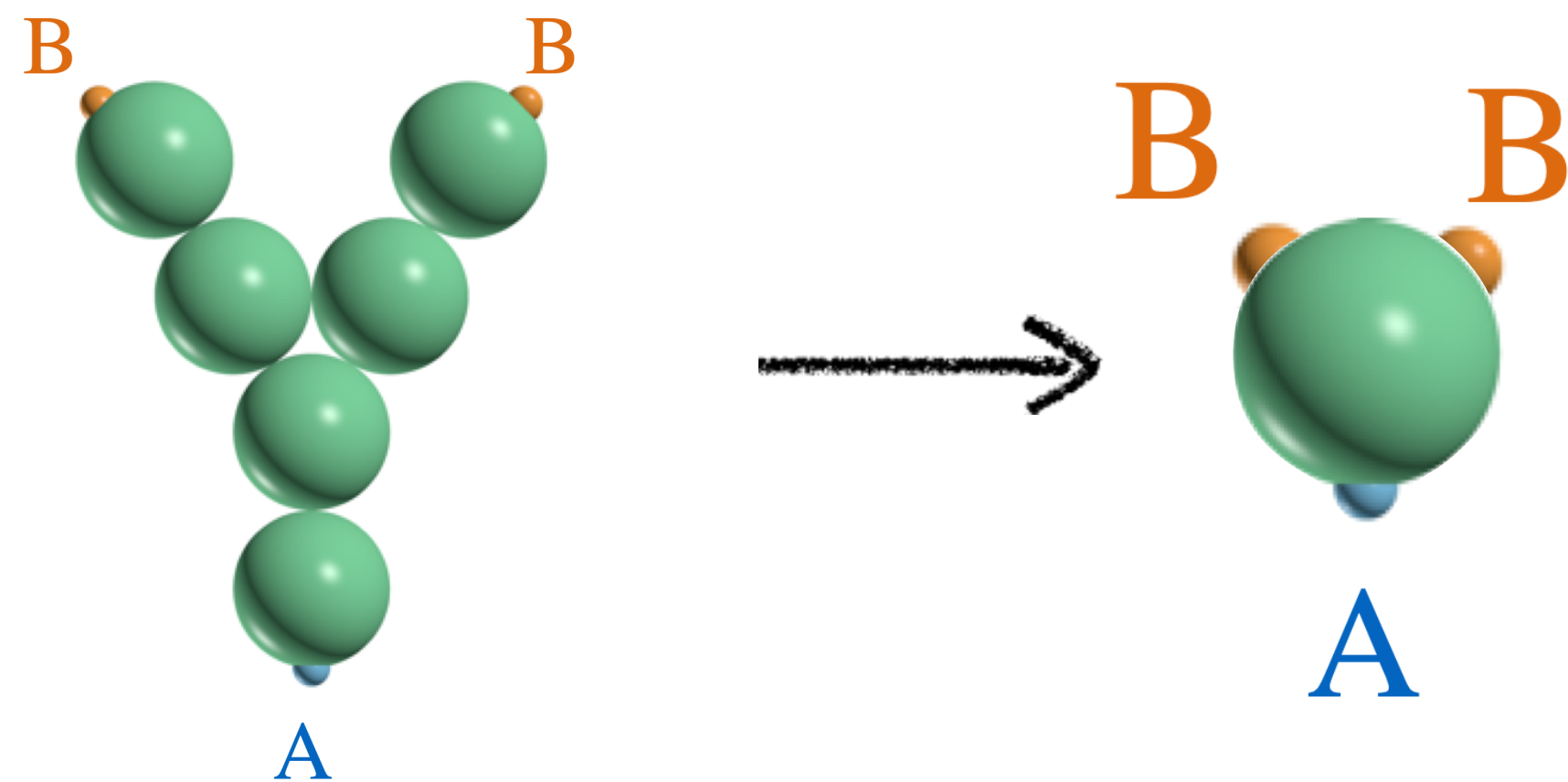
**WE USE  $p \equiv p_B$  FROM WERTHEIM THEORY**

# Comparison with experiments: calculating $S(0)$

We use Wertheim theory to calculate  $S(0) = \left( \frac{d\beta P}{d\rho} \right)^{-1} = \left( \boxed{\frac{d\beta P_{HS}}{d\rho}} + \boxed{\frac{d\beta P_b}{d\rho}} \right)^{-1}$

dependent on temperature and packing fraction of reference hard sphere system  $\Phi_{HS}$

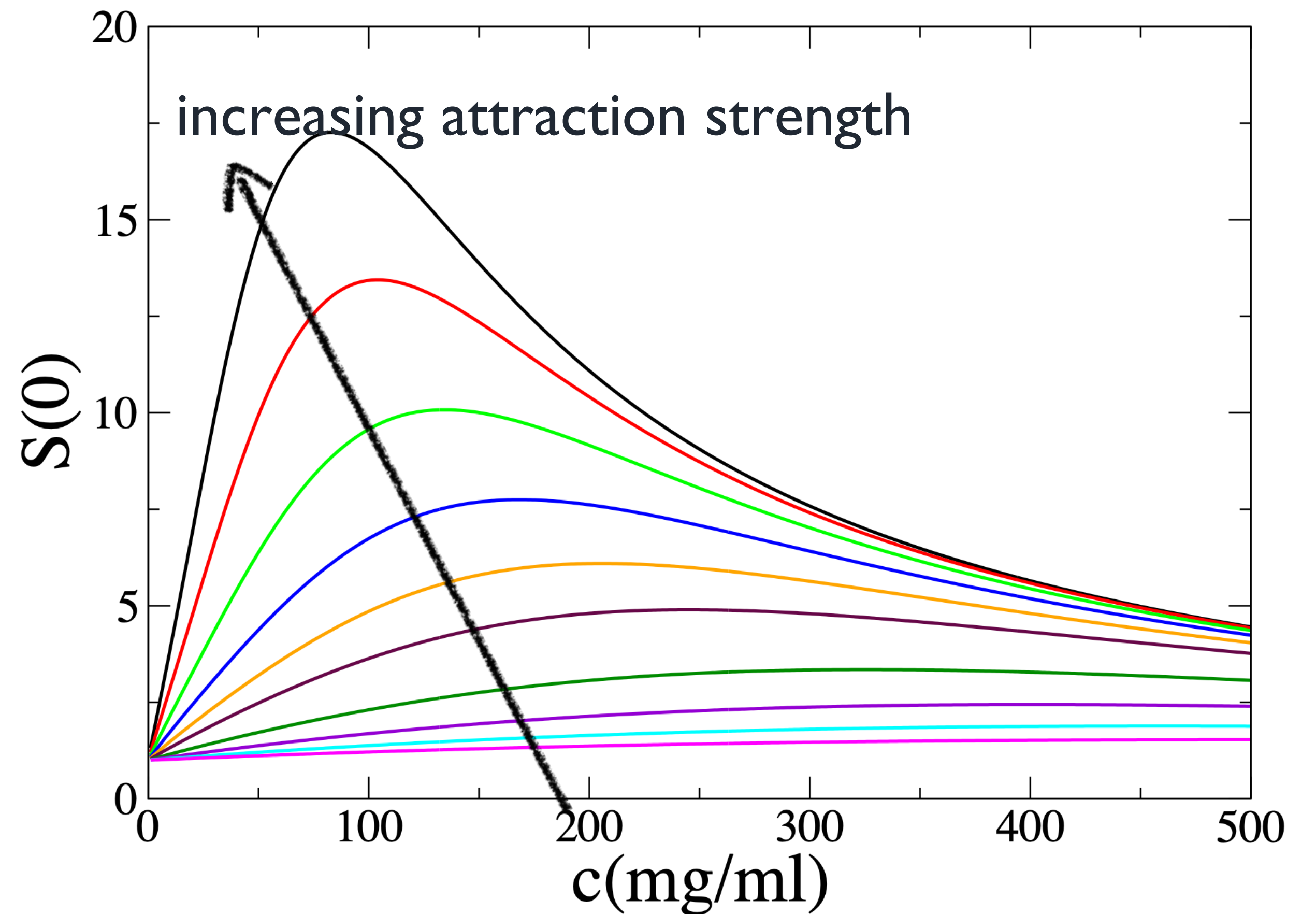
we use effective patchy spheres



which effective diameter?

Carnahan-Starling

bond contribution



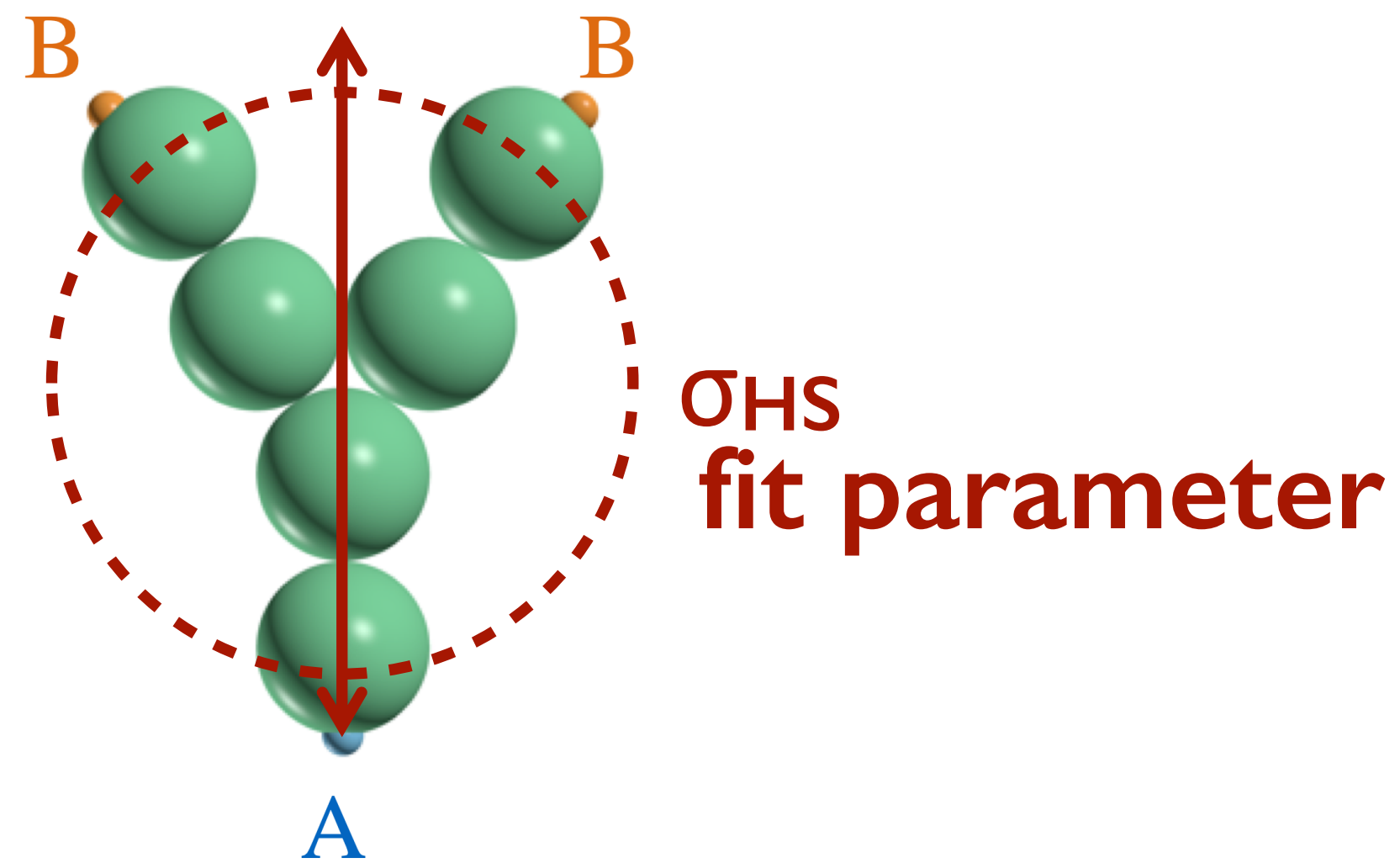
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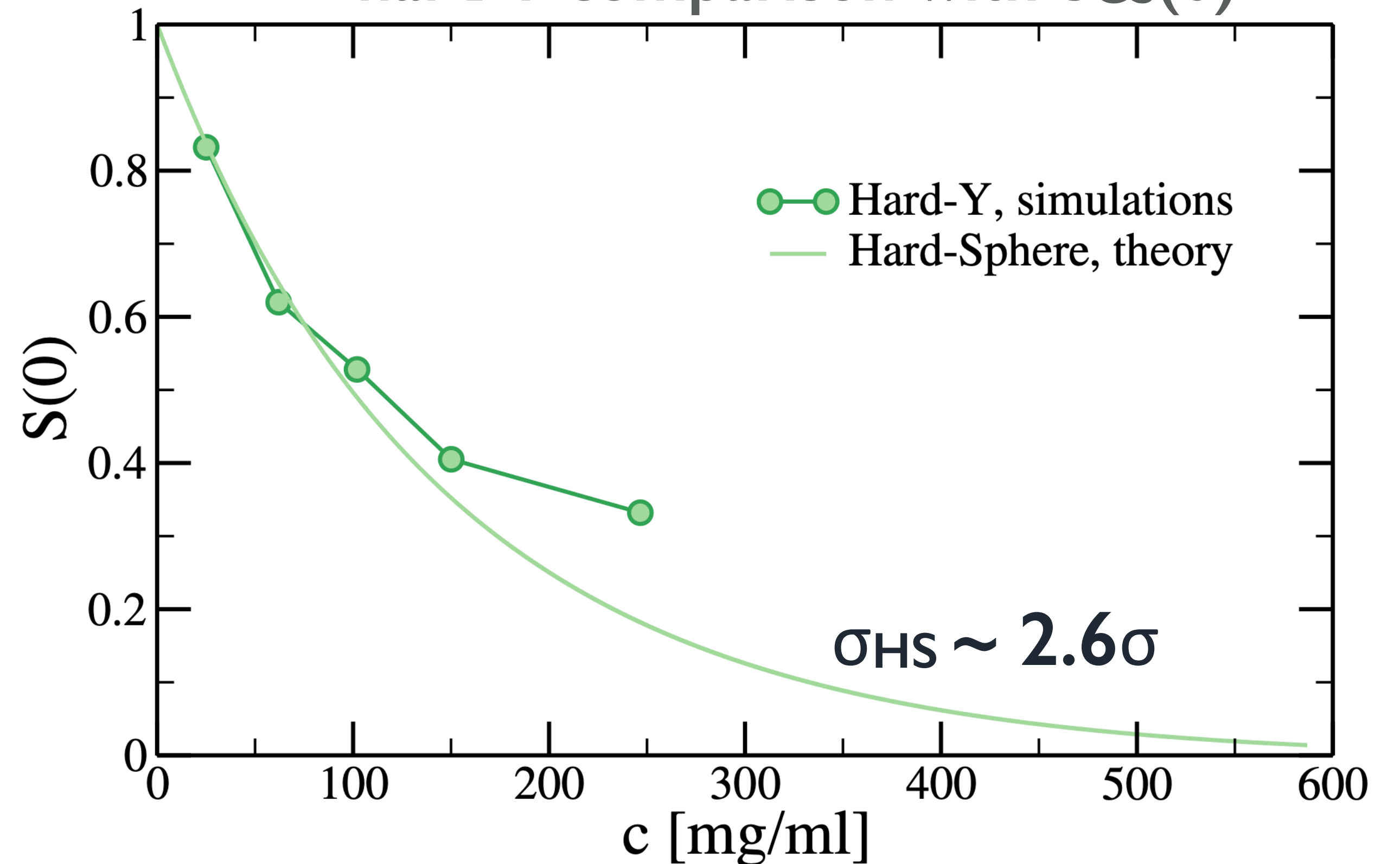
Carnahan-Starling

bond contribution

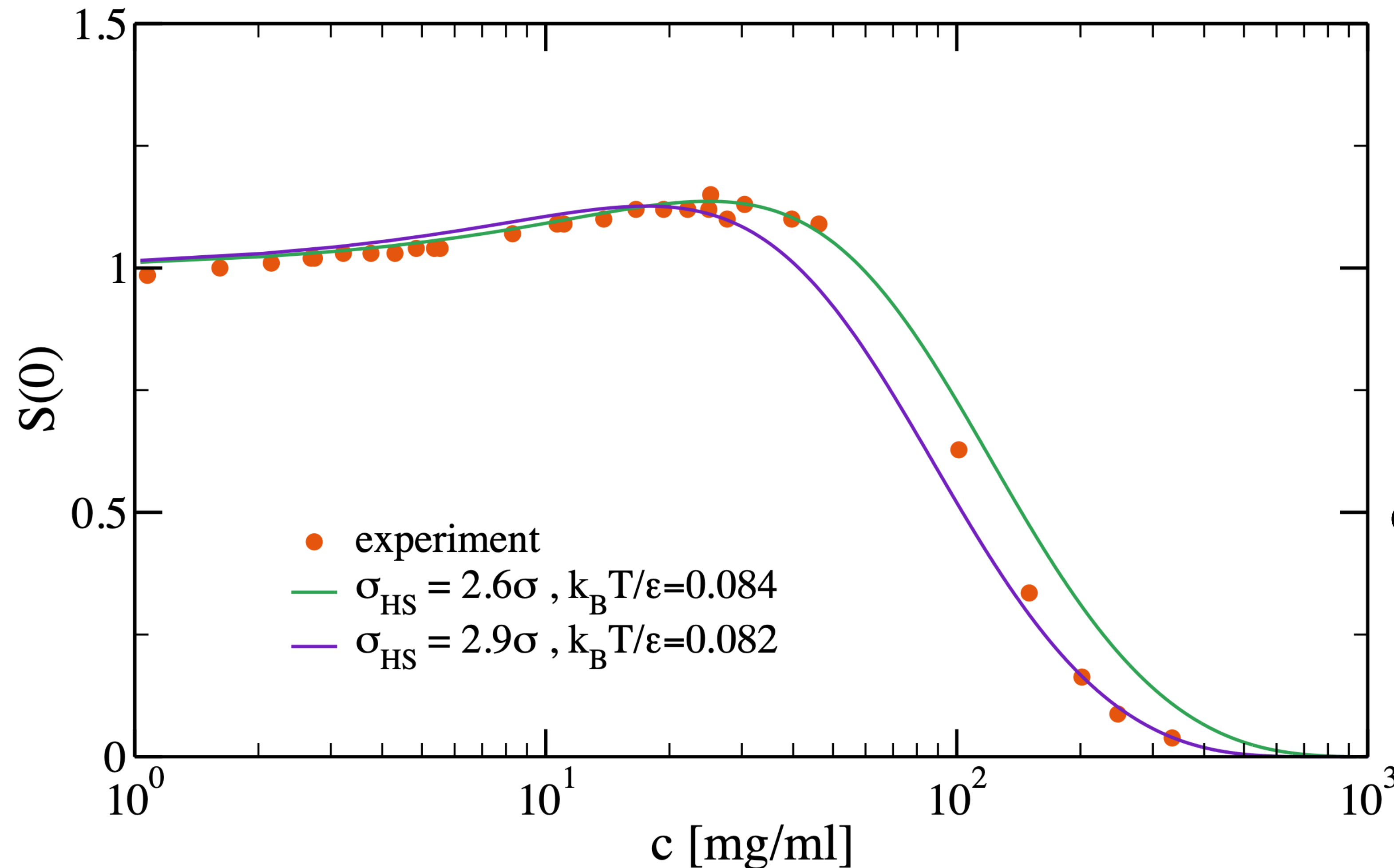


conversion to experimental units  
based on hydrodynamic radius

hard Y comparison with  $S_{CS}(0)$



# Comparison with experiments: calculating $S(0)$



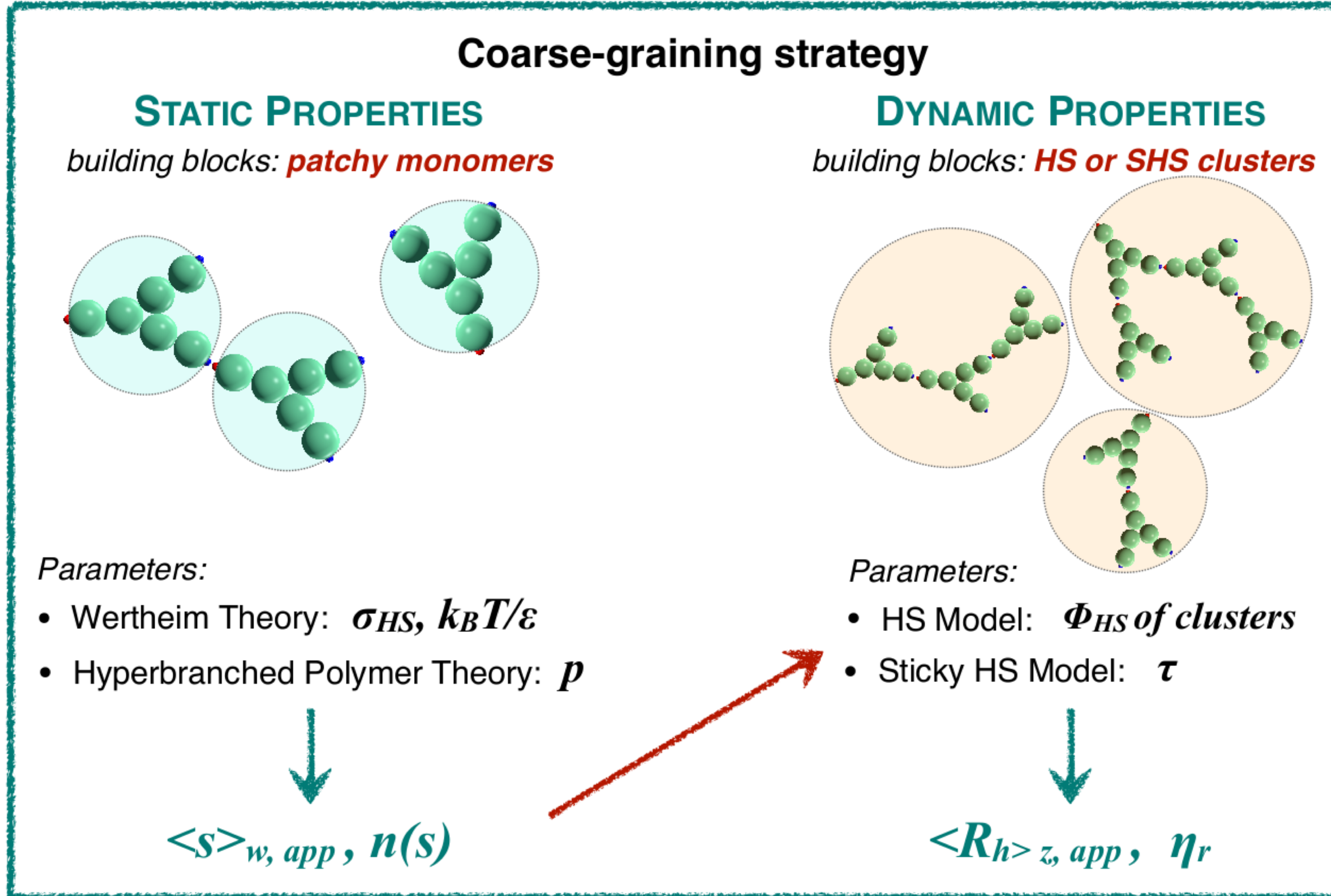
**bond attraction strength  $\sim 12$  kBT**

**effective size much smaller than geometric Y size**

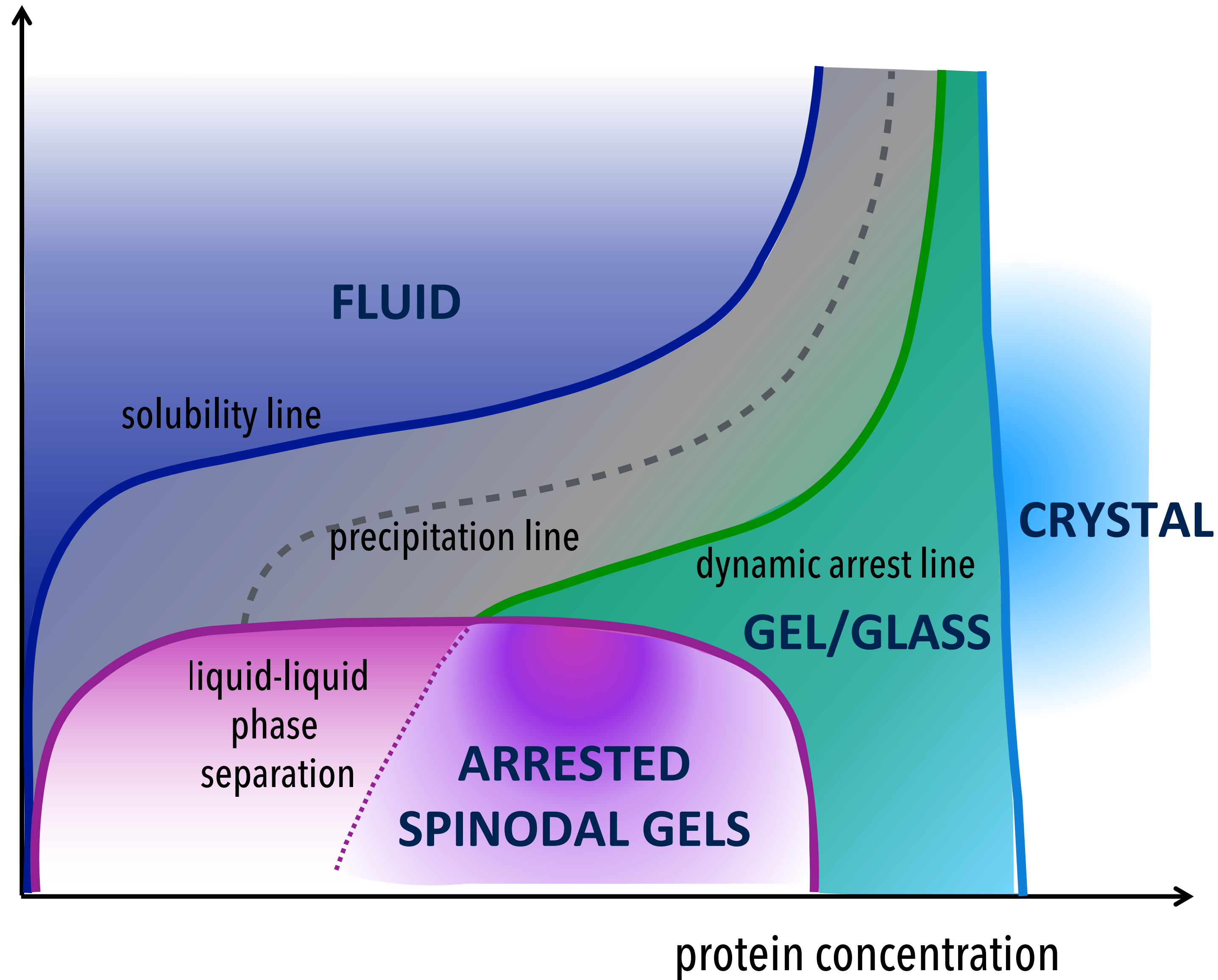
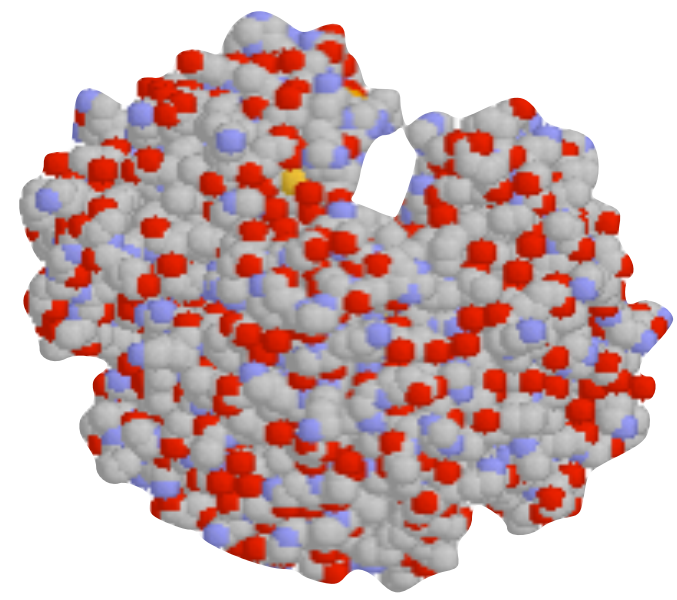
**quite good description of the data with the YAB patchy model**

**patchy interactions are not enough: importance of the shape**

# From static to dynamic properties



# Back to generic phase diagram: interplay with dynamics

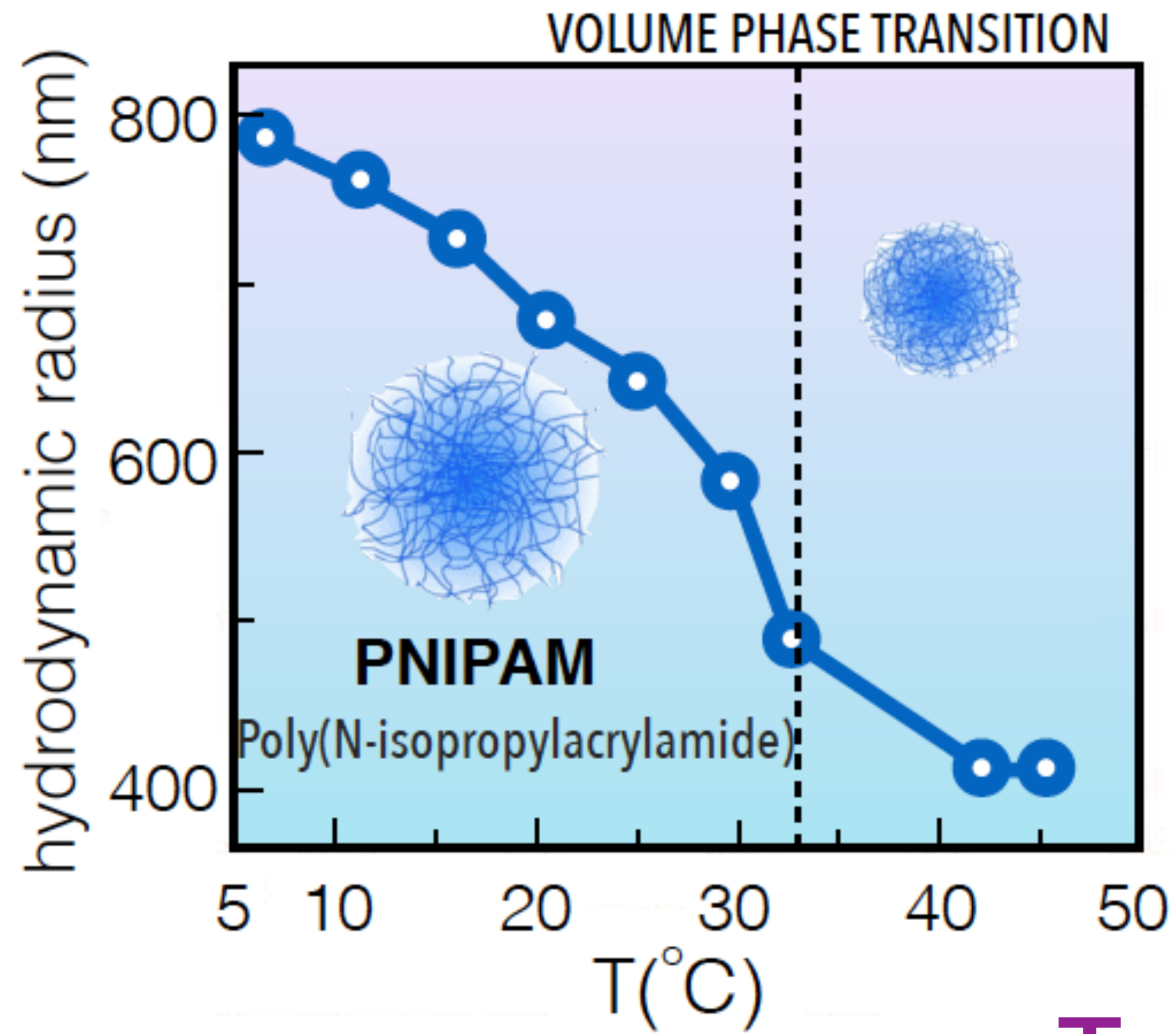




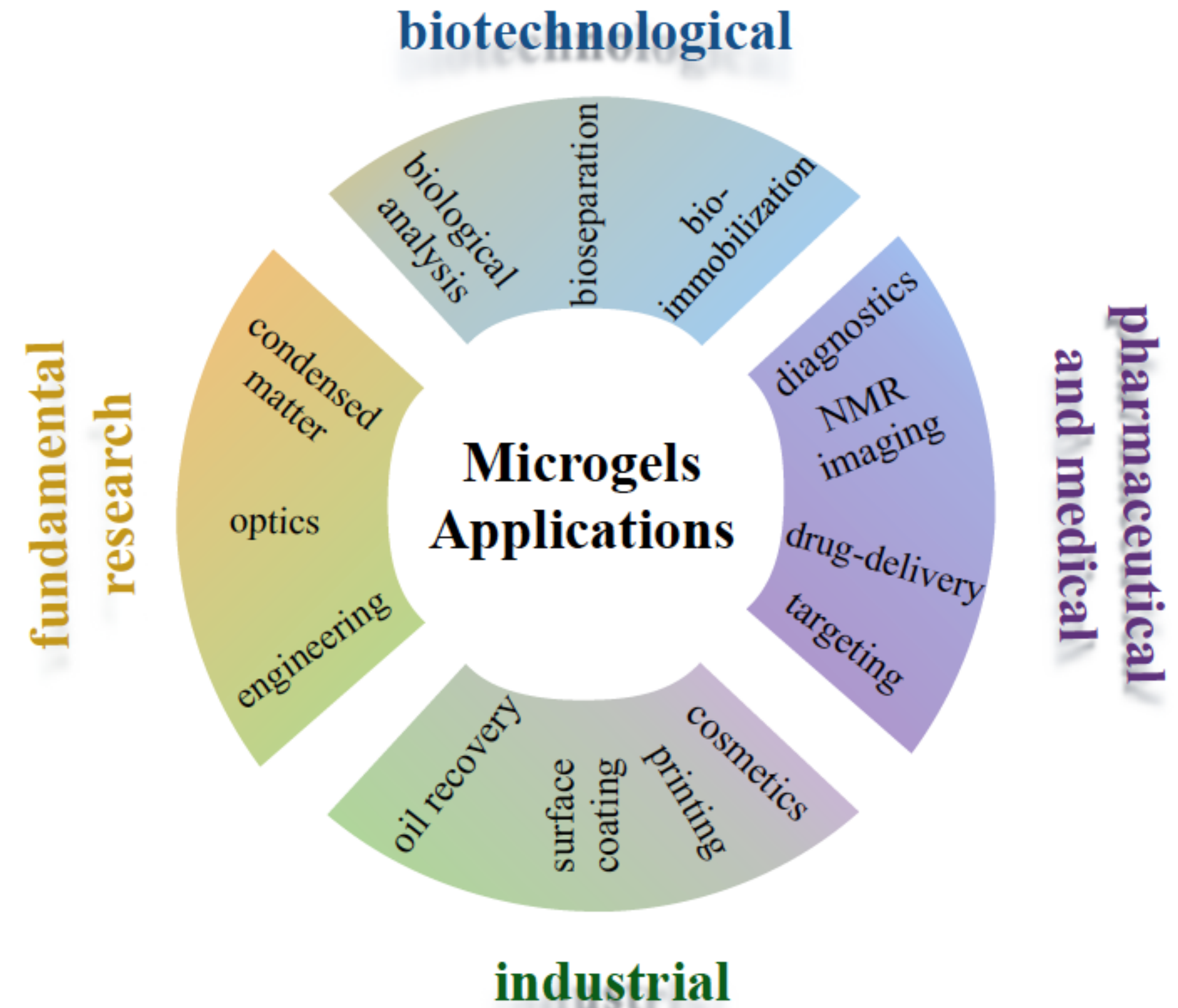
# Why microgels?

## COLLOIDS MADE BY CROSS-LINKED POLYMER NETWORKS

- responsive
- easy to synthesize
- versatile for applications



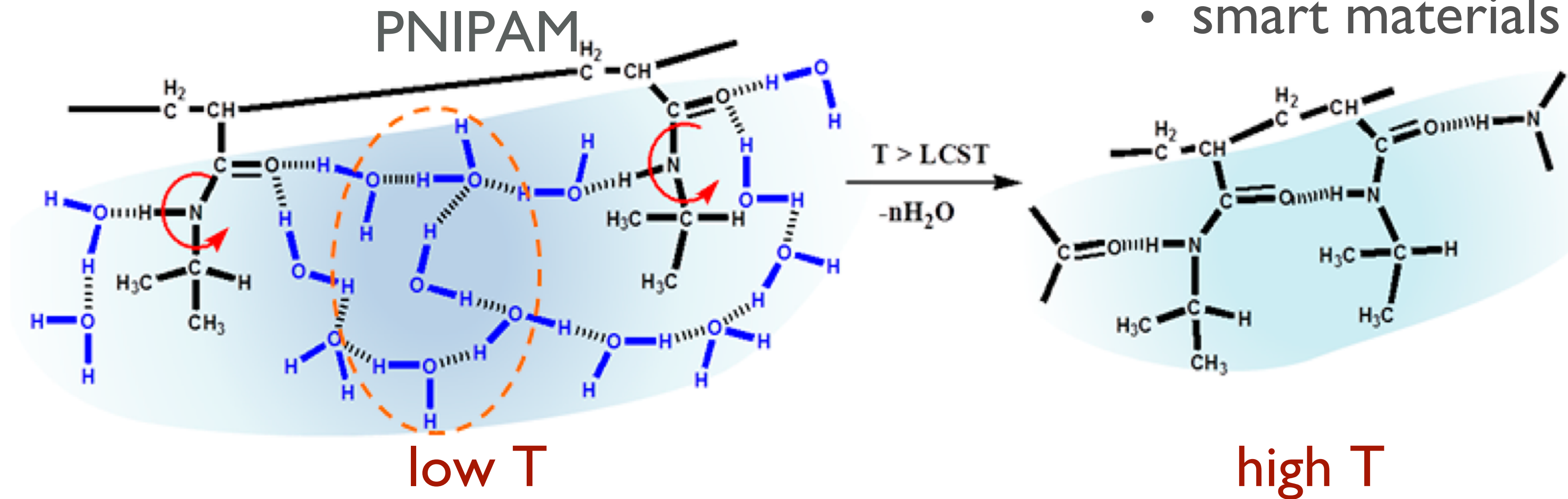
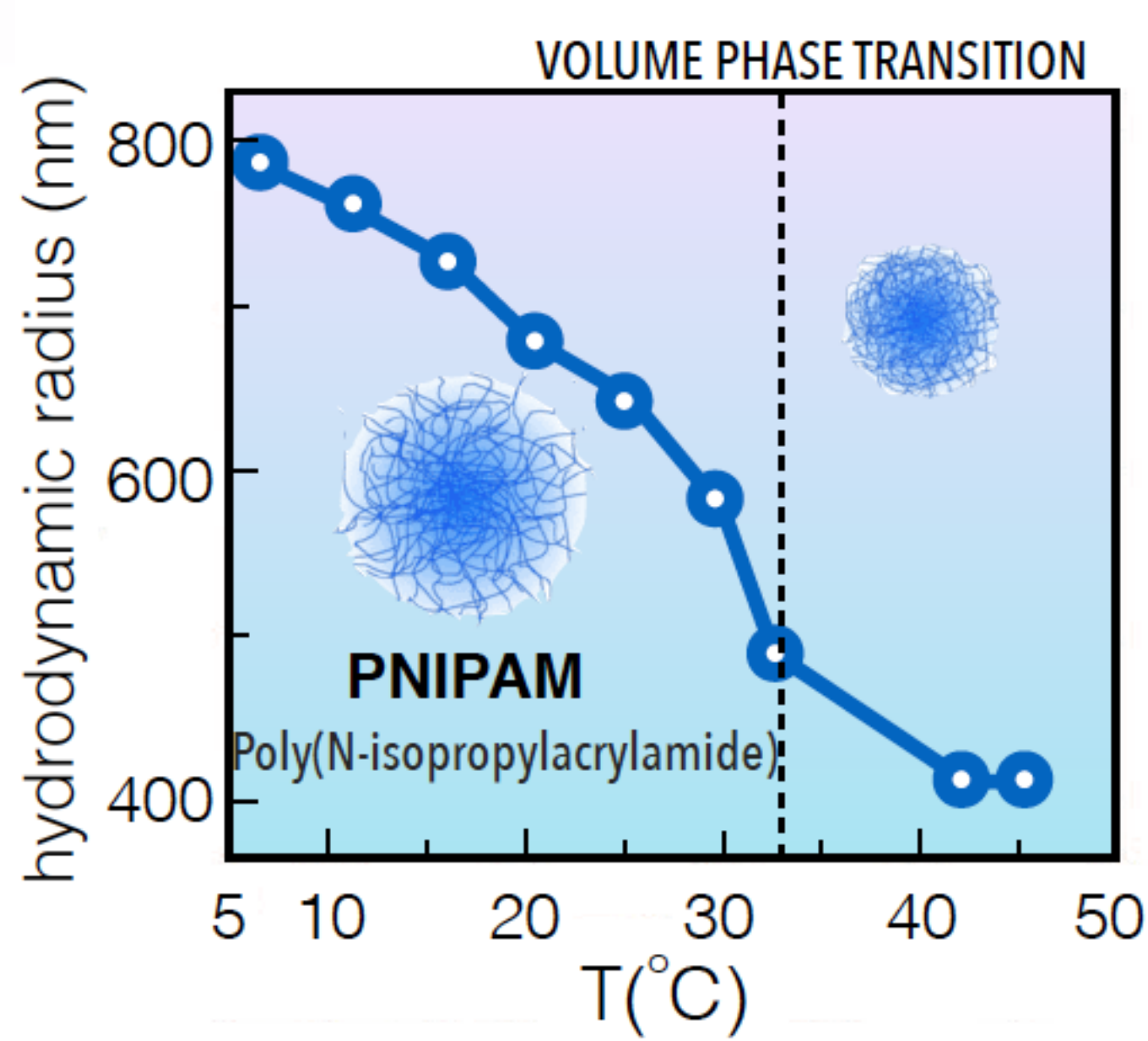
$T_{VPT} \sim 32^{\circ}\text{C}$



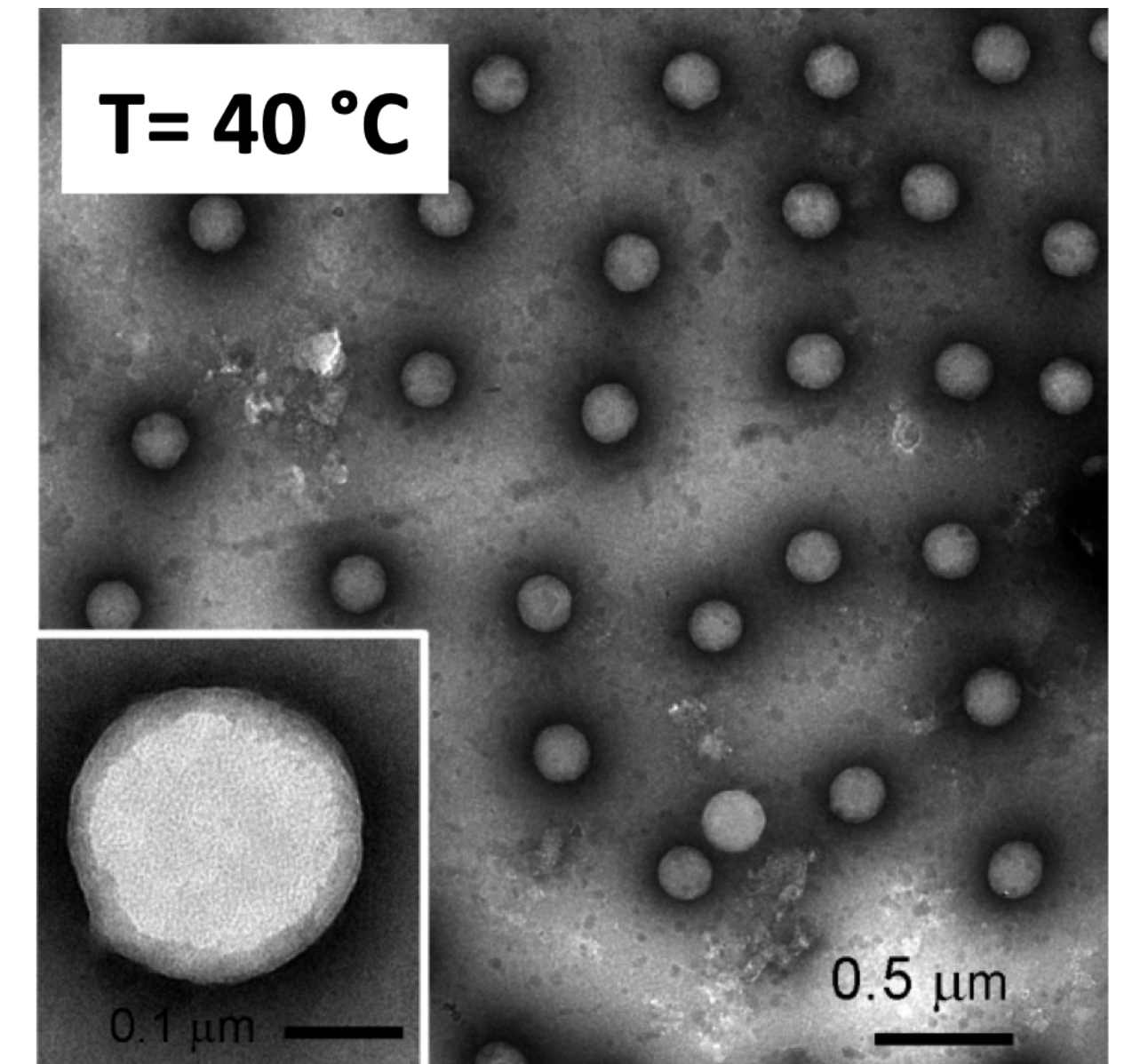
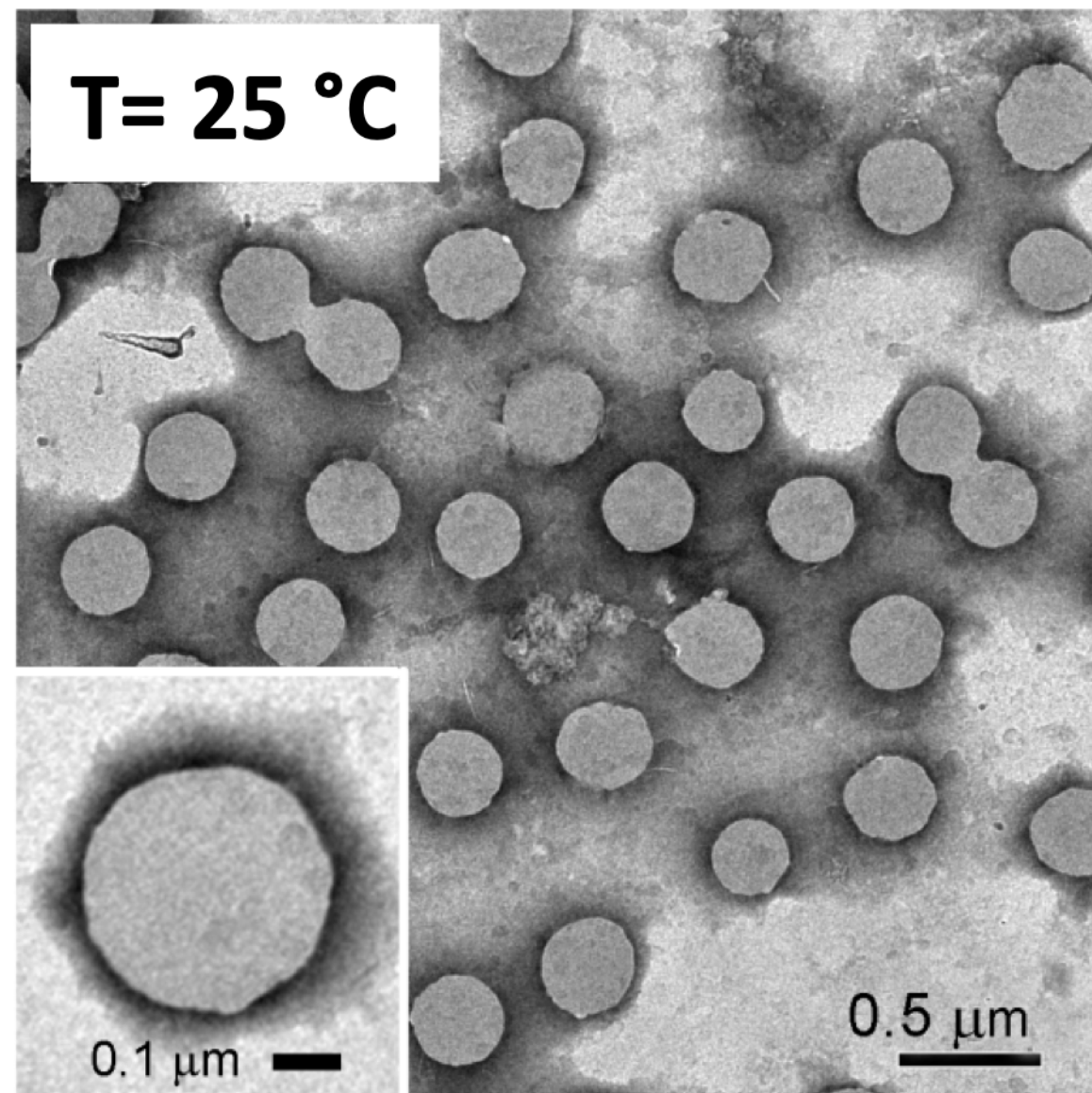
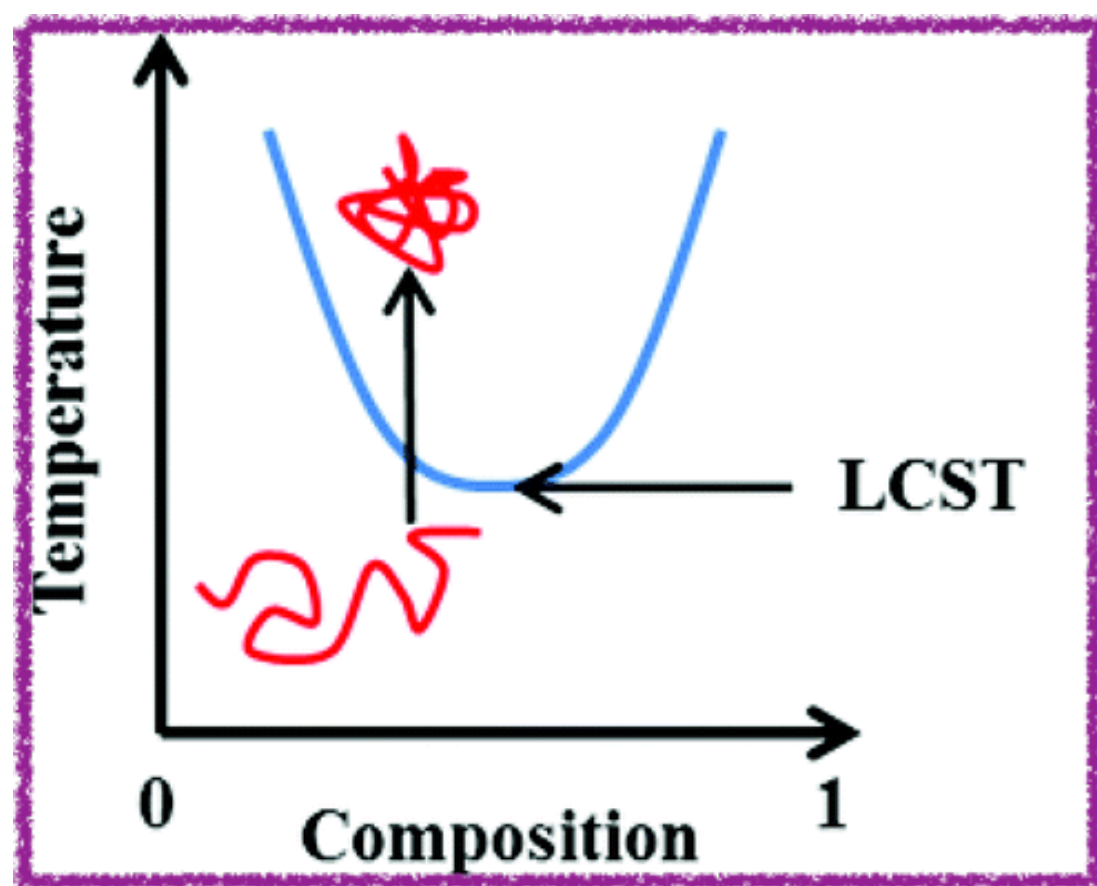


# Volume phase transition

- fully reversible
- smart materials

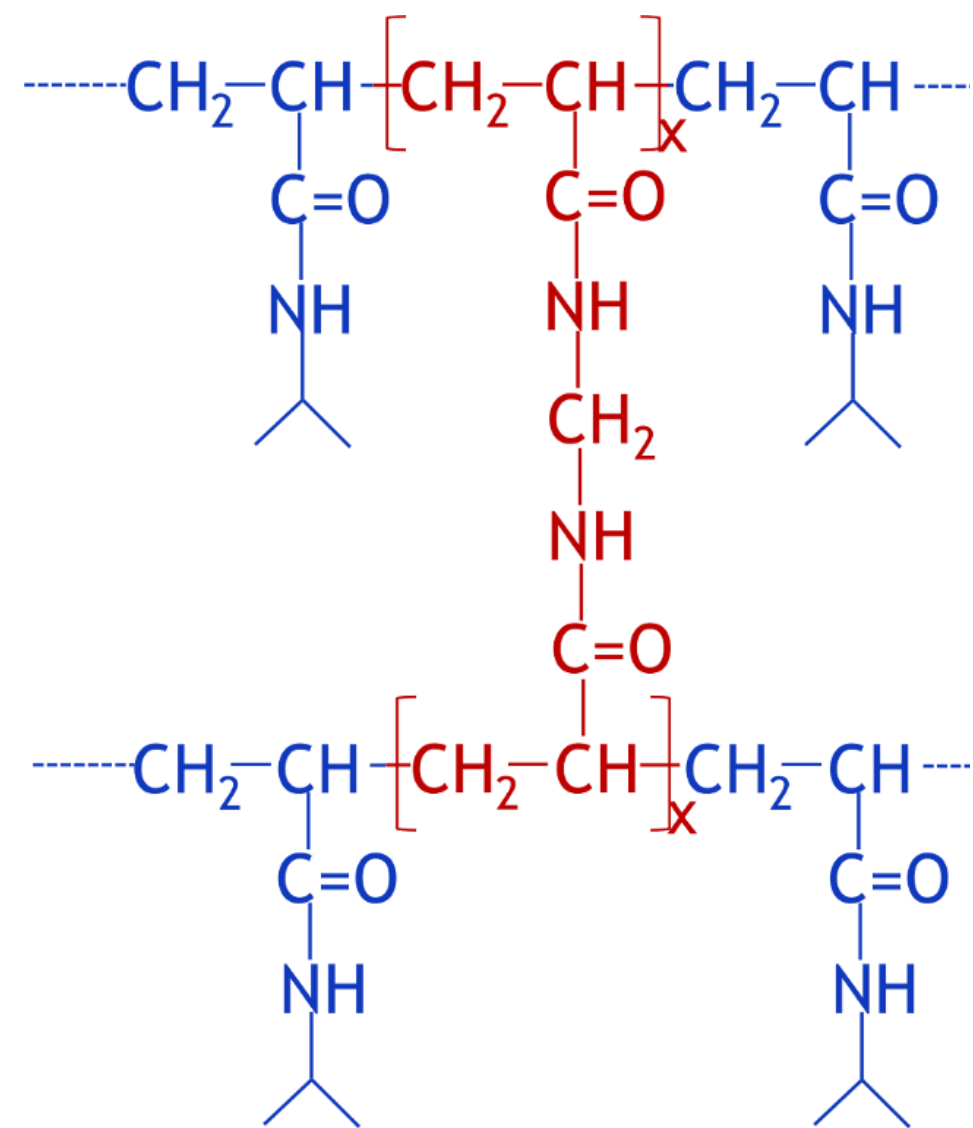


echo of the **COIL-TO-GLOBULE TRANSITION** in PNIPAM chains



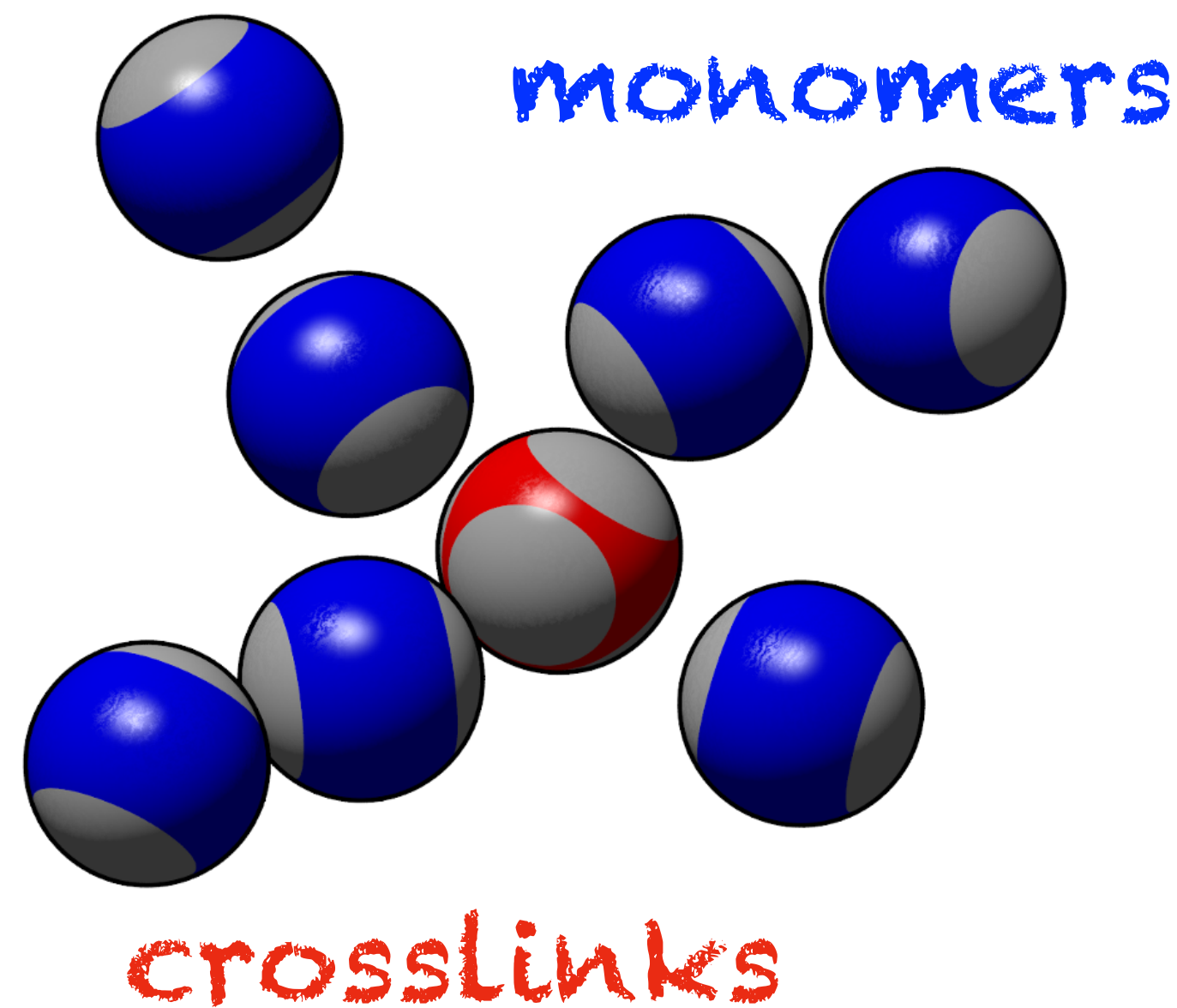
# Microgels assembled from patchy particles

polymerization of  
BIS crosslinkers  
+ NIPAM monomers

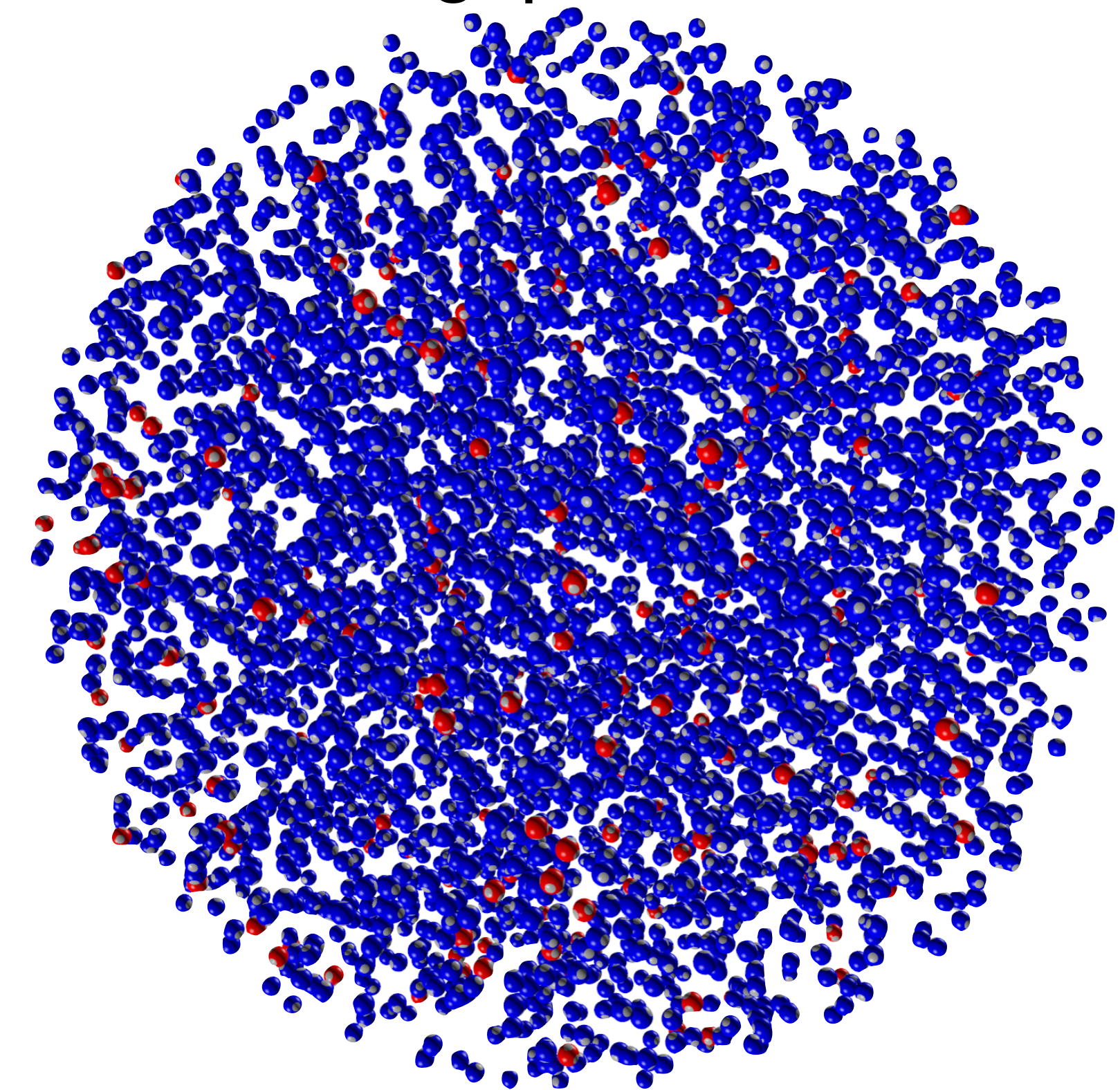


First step  
→

self-assembly of  
a binary mixture of  
4-patch + 2-patch particles

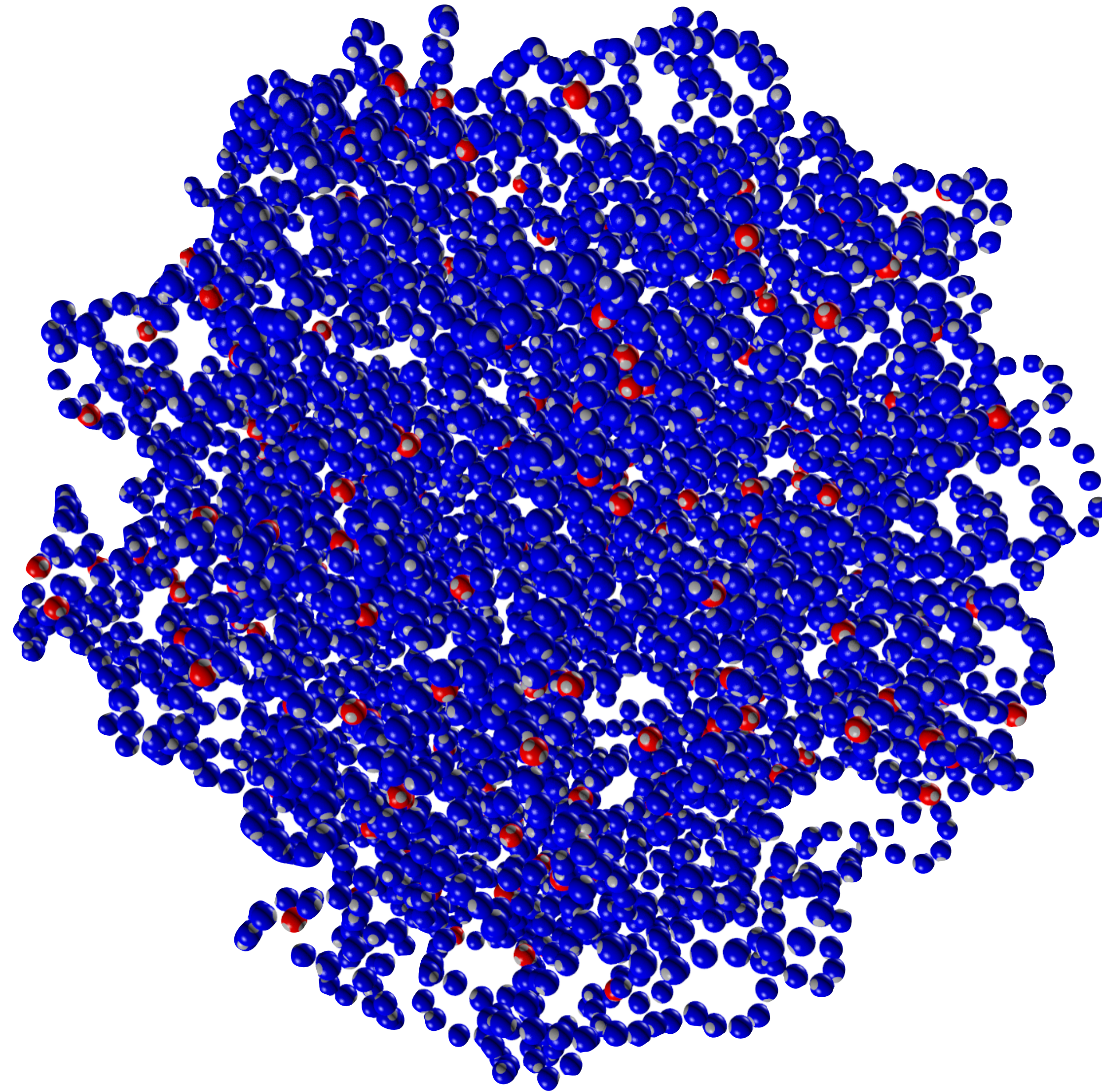


use a confining sphere of radius  $Z$



how to assemble  
fully-bonded networks?

# Fully assembled network

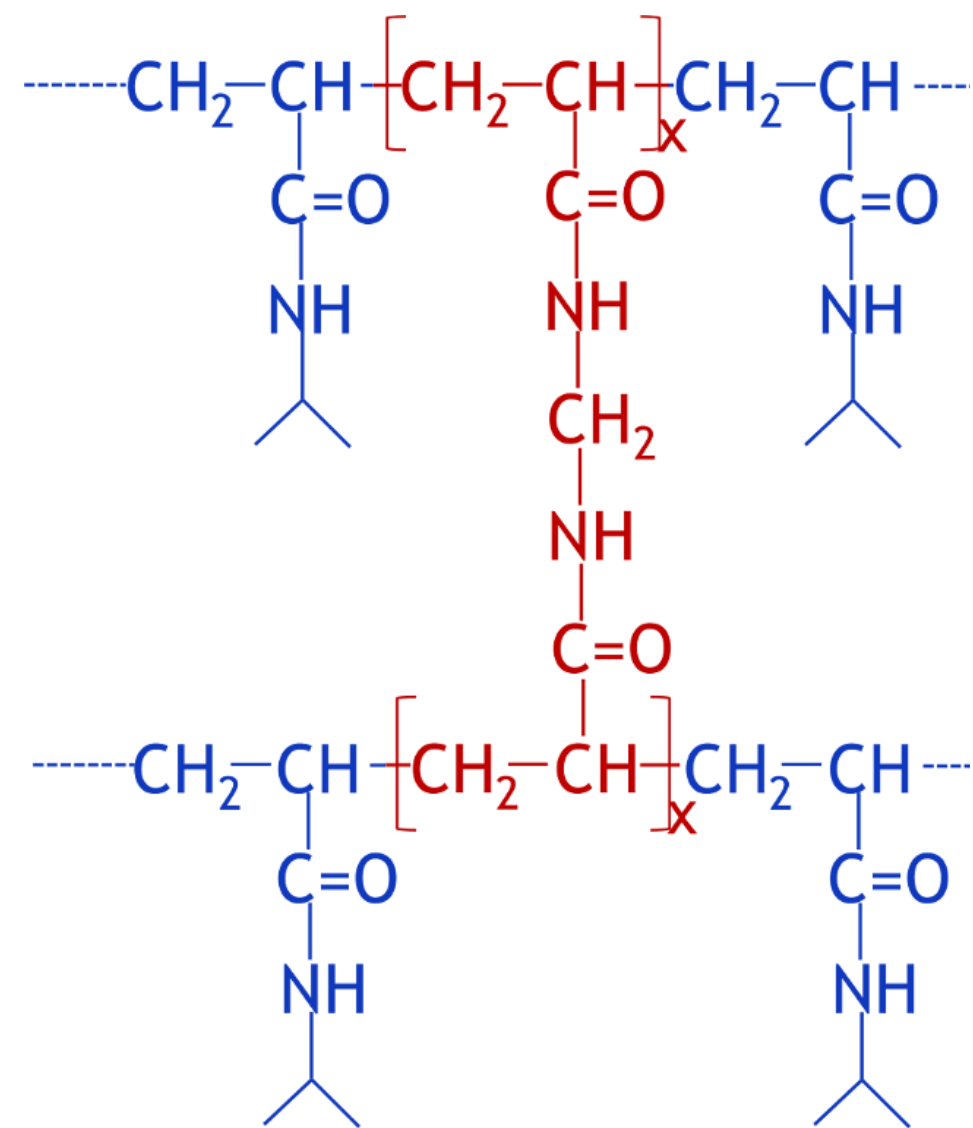


blue = monomers

red = crosslinks

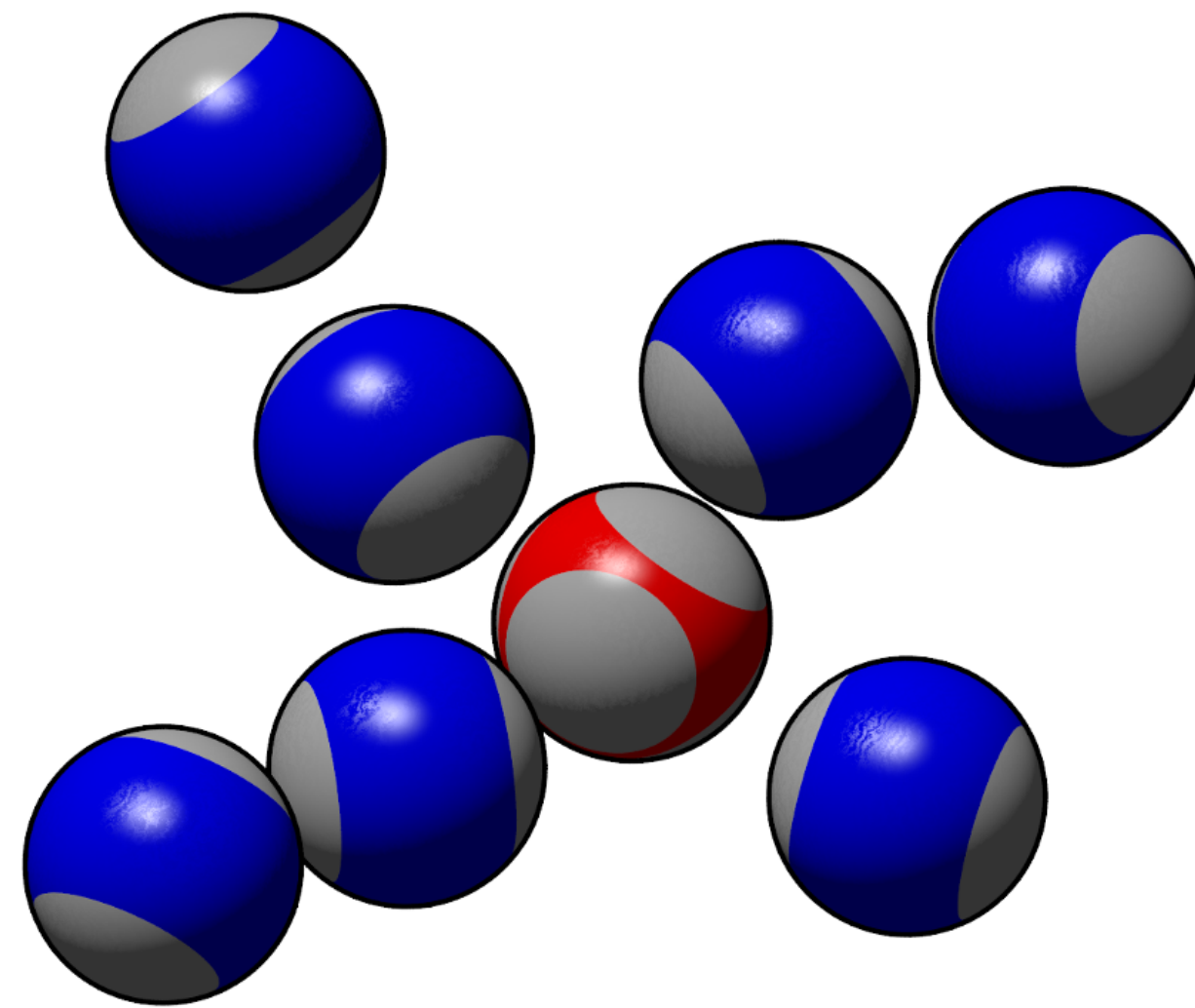
# Microgel assembly protocol

polymerization of  
BIS crosslinkers  
+ NIPAM monomers

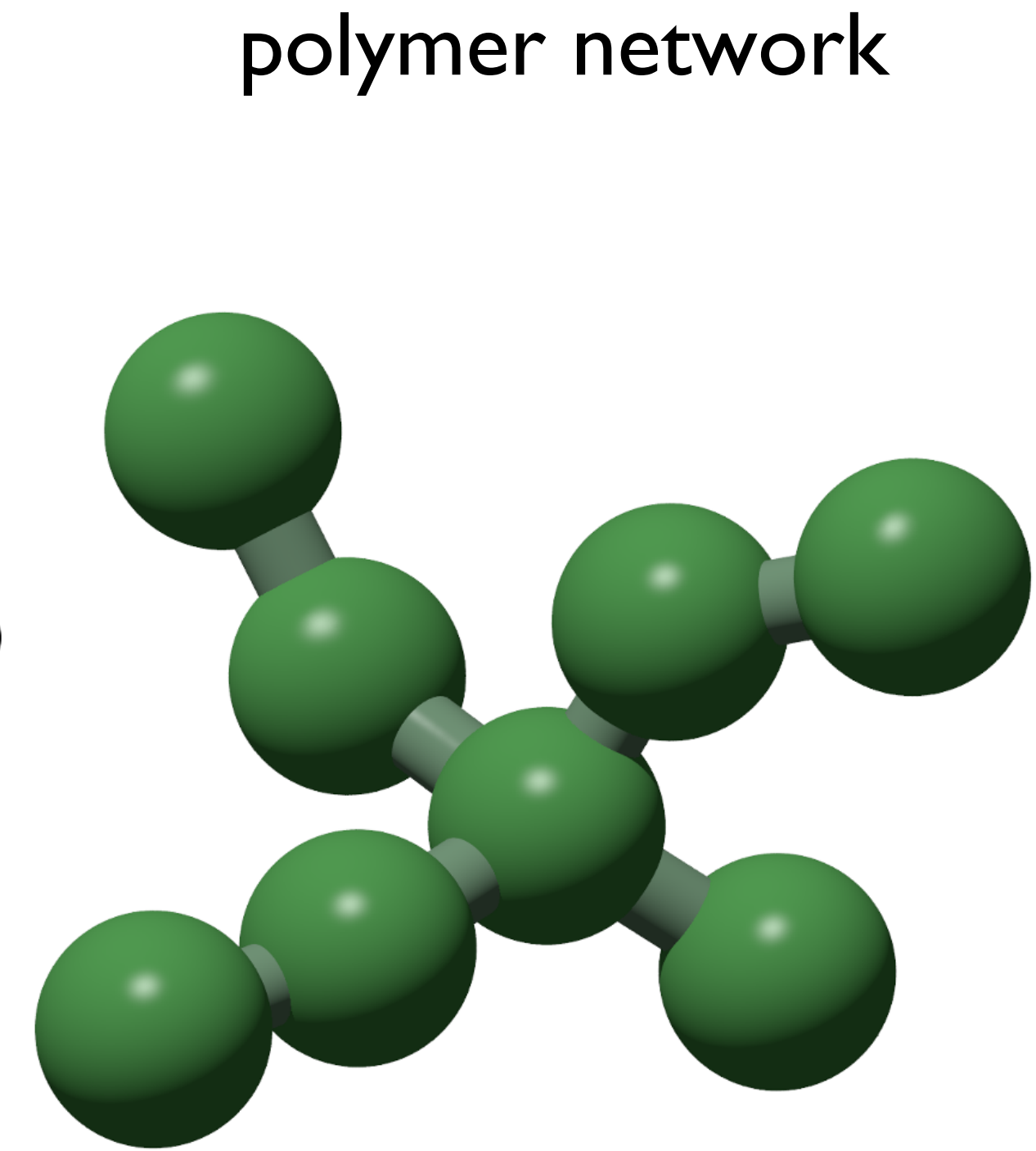


First step  
→

self-assembly of  
a binary mixture of  
4-patch + 2-patch particles



Second step  
→



$$V_{WCA}(r) = \begin{cases} 4\epsilon \left[ \left(\frac{\sigma}{r}\right)^{12} - \left(\frac{\sigma}{r}\right)^6 \right] + \epsilon & \text{if } r \leq 2^{\frac{1}{6}} \sigma \\ 0 & \text{otherwise} \end{cases}$$

$$V_{FENE}(r) = -\epsilon k_F R_0^2 \ln\left(1 - \left(\frac{r}{R_0 \sigma}\right)^2\right) \text{ if } r < R_0 \sigma$$

between all particles

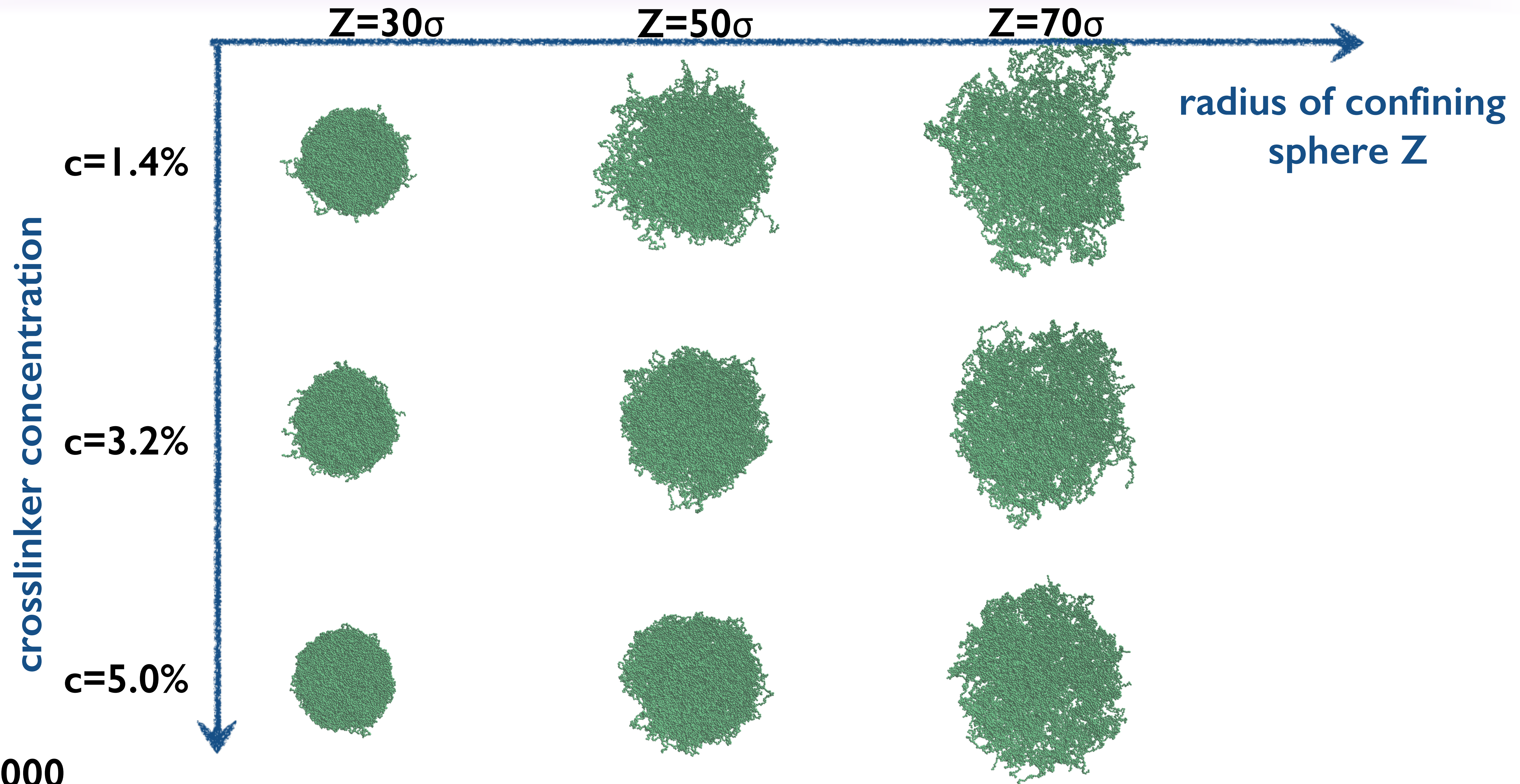
between bonded particles

classical

bead-spring model

Grest & Kremer, Phys. Rev. A  
(1986)

# Results: assembled microgels — swollen regime



**$N=42000$**

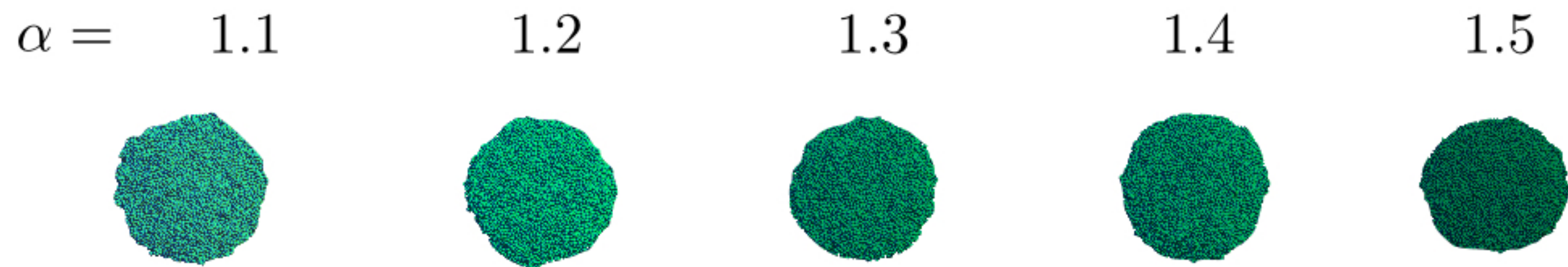
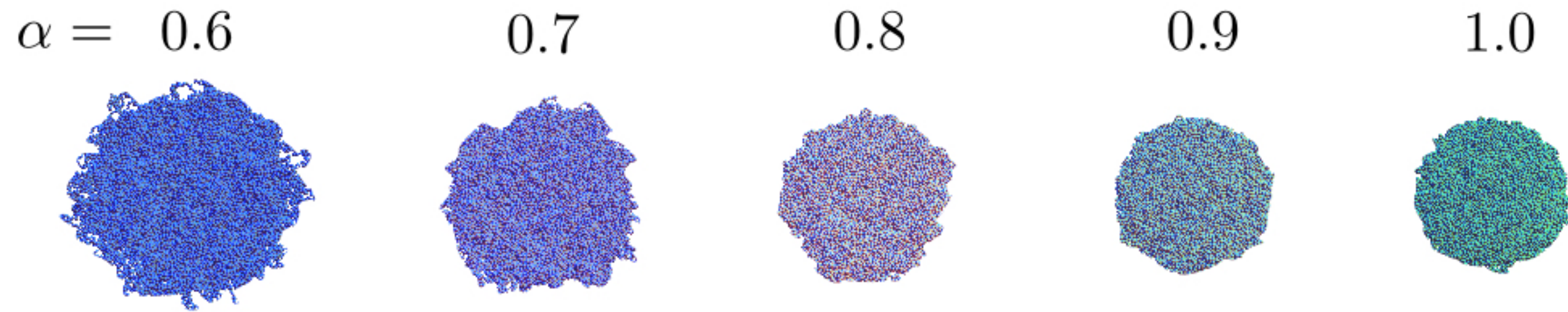
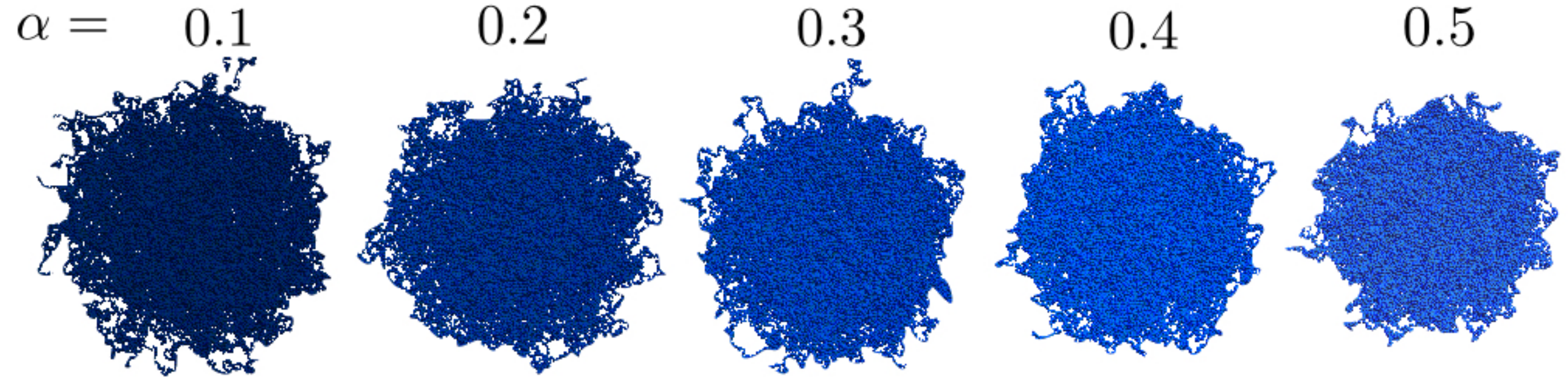
# Swelling behaviour

We complement the  
bead-spring model  
adding a solvophobic term

Soddemann, Dunweg, Kremer Eur. Phys. J. E (2001)

$$V_{\alpha}(r) = \begin{cases} -\varepsilon\alpha & \text{if } r \leq 2^{1/6}\sigma \\ \frac{1}{2}\alpha\varepsilon[\cos(\gamma(r/\sigma)^2 + \beta) - 1] & \text{if } 2^{1/6}\sigma < r \leq R_0\sigma \\ 0 & \text{otherwise} \end{cases}$$

$\gamma = \pi(2.25 - 2^{1/3})^{-1}$   
 $\beta = 2\pi - 2.25\gamma$

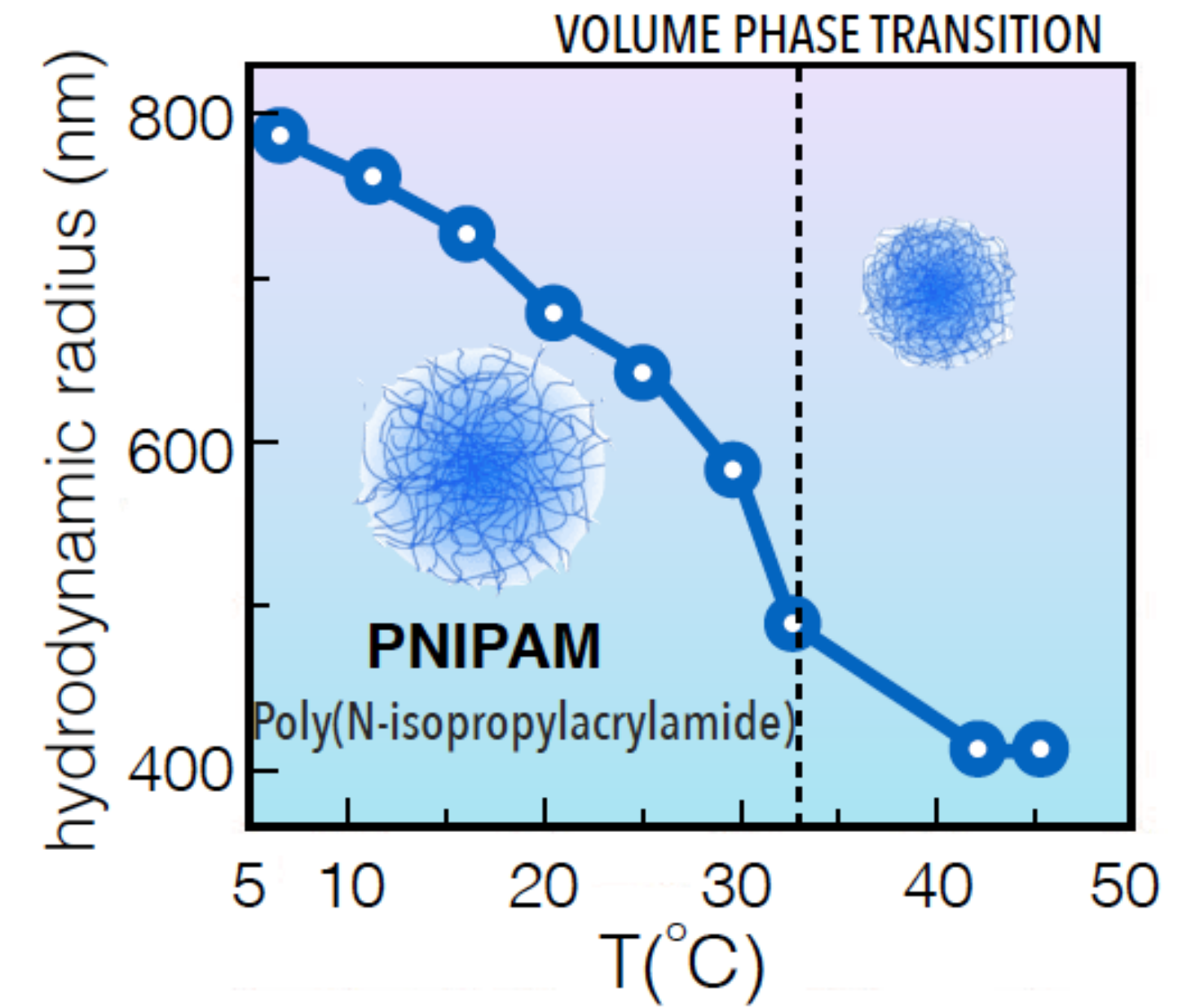
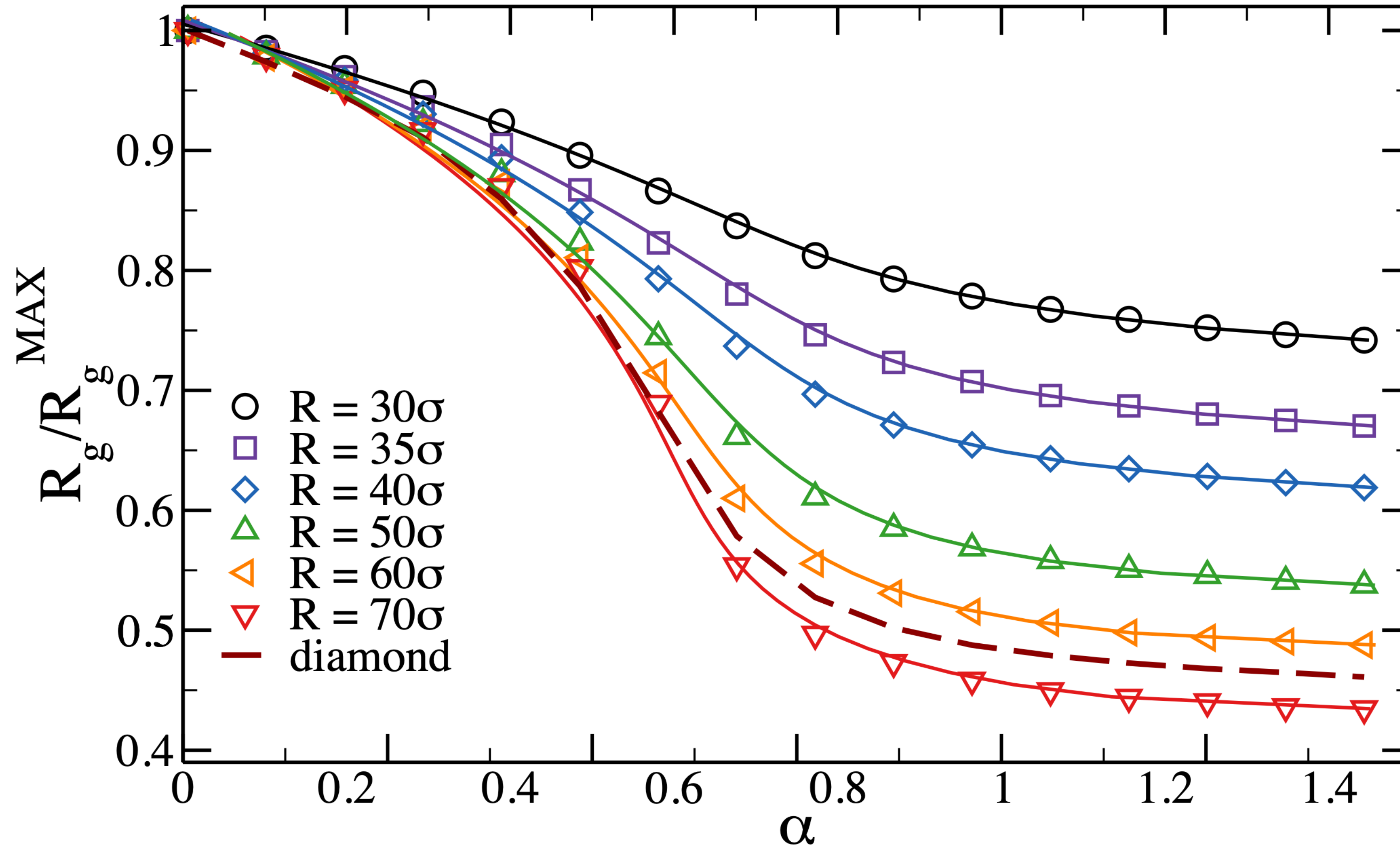


$\alpha$  : effective temperature

Volume Phase  
Transition (VPT)

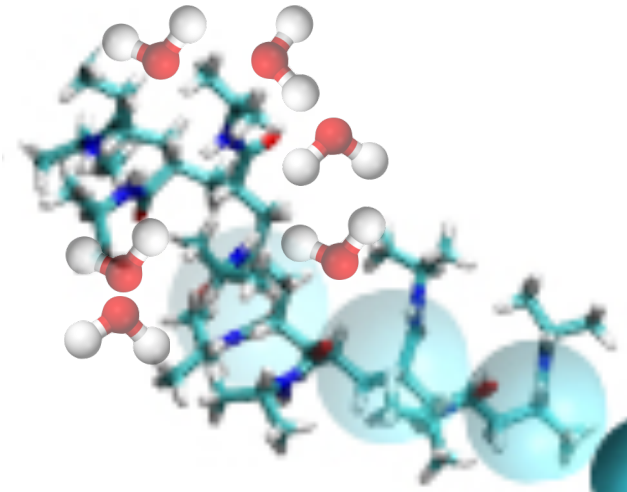
# Swelling behaviour

$c=3.2\%$

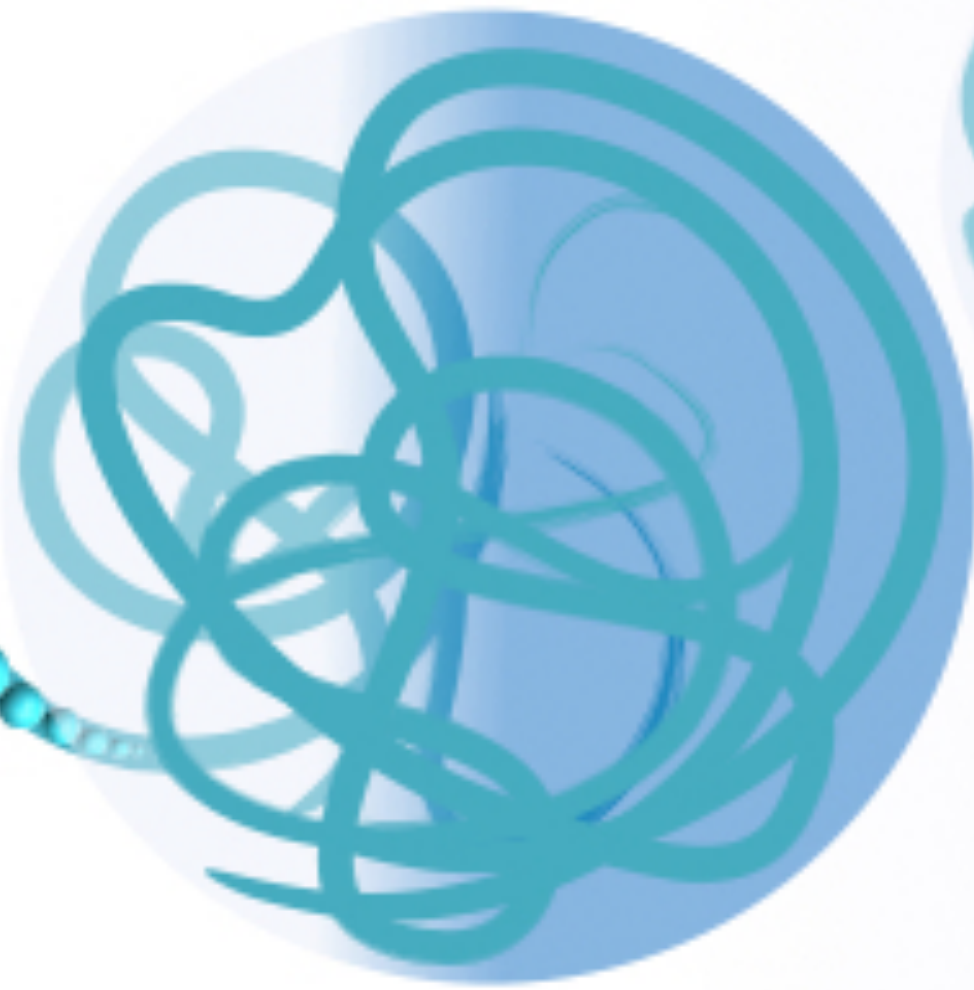


# A MULTISCALE APPROACH TO MICROGELS

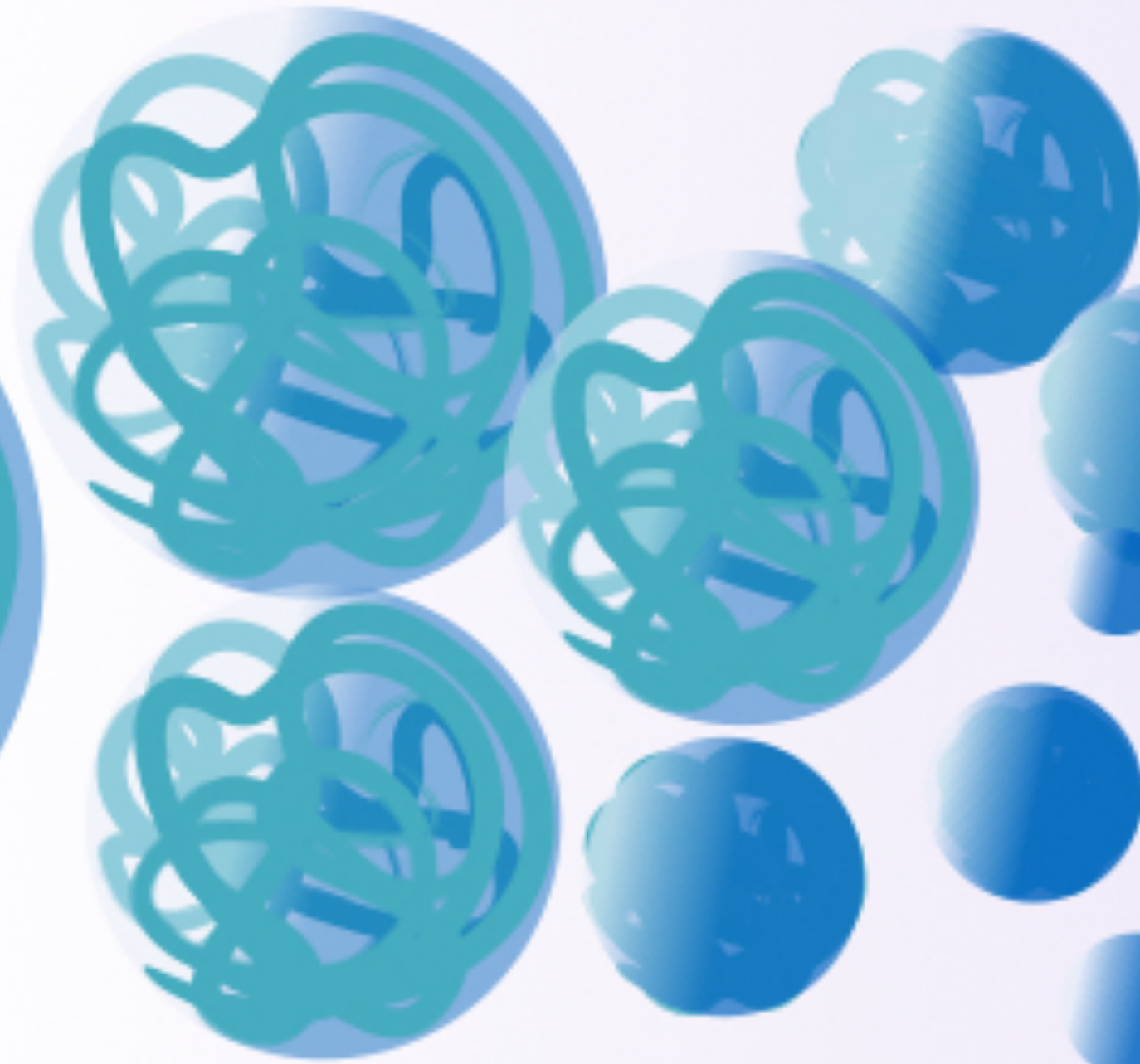
water-chain  
interactions



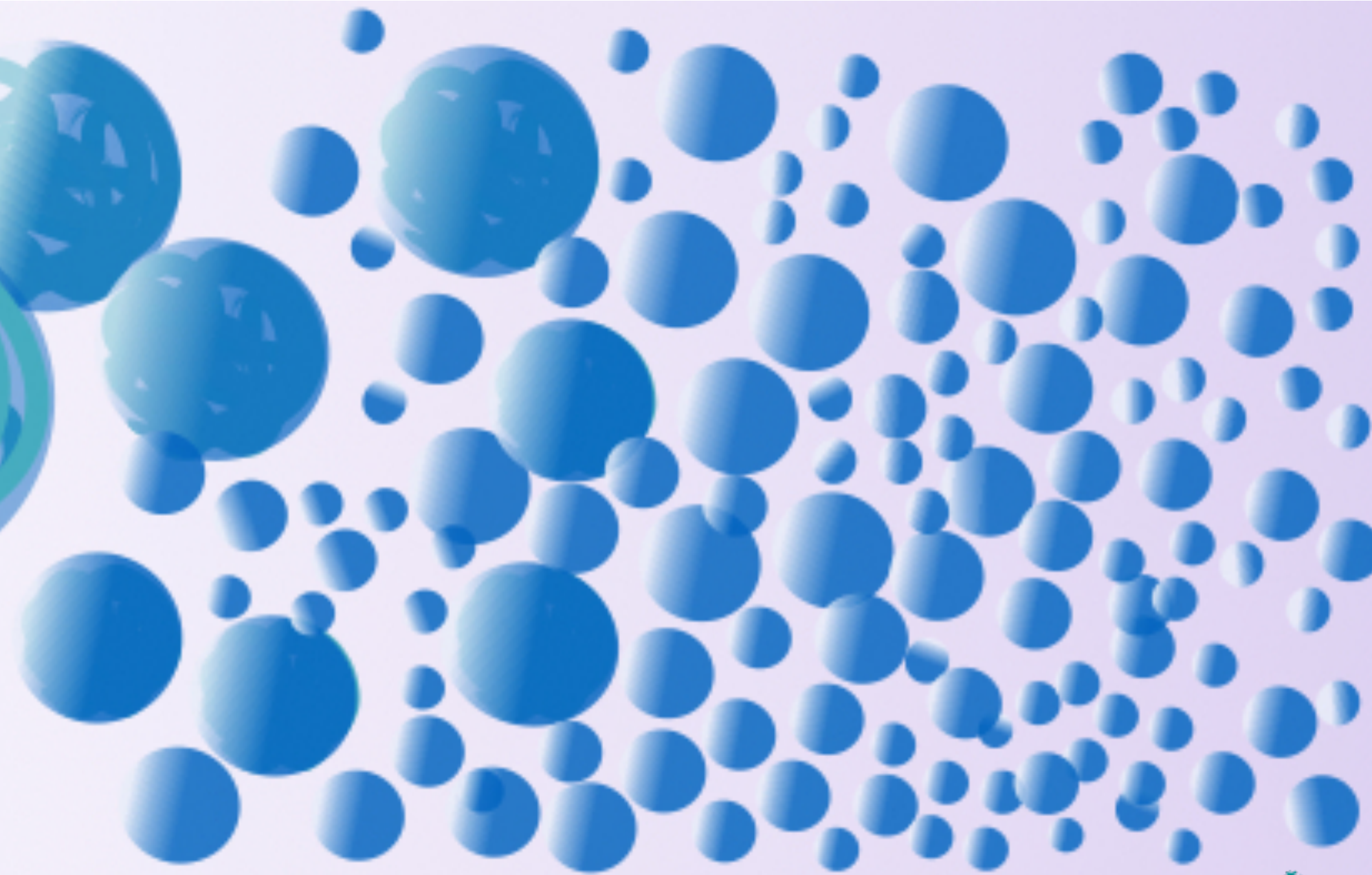
internal  
chain-chain



inter-particle



material



atomic

polymeric

colloidal

macroscopic

